TABLE 4-7. Grid cells assigned to oil fields not in the original MED files.

Oil Field/Facility	Grid Cell	
Identification Code	(I,J)	Comments
Los Angeles County		•
Alondra/100101	43, 13	Read from USGS oil and gas map
Canton Creek/100155	37, 28	Facility ID code assigned in this study; read from USGS oil and gas map
Hyperion/100112	43, 14	Rough guess from KVB UTMs
Las Llajas/100115	38, 24	Read from standard map
Lawndale/100116	43, 14	Read from USGS oil and gas map
Lyon Canyon/100146	40, 24	Read from USGS oil and gas map
Newgate/100123	49, 14	Rough guess from KVB UTMs
Newhall-Potrero/100148	40, 25	Field should have been in original MED file; read from USGS oil and gas map
Saugus/100152	40, 26	Read from USGS oil and gas map
Orange County		
Brea-Olinda/100104 (Los Angeles County ID code)	52, 15	Same grid cell as Brea-Olinda No. 100202
Kraemer/100213	53, 14	Facility ID code assigned in this study; read from USGS oil and gas map

### 5 TASK 3: SPECIES PROFILE DEVELOPMENT

The profiles used to speciate TOG emissions, which identify individual organic compounds and their weight percents, are considered the most uncertain component of the inventory. Accordingly, two methods were used in this study to develop improved organic gas species profiles: (1) sampling and analysis techniques and (2) a review of existing data. This section presents the species profiles developed using these two methods. The development of TOG species profiles is somewhat complicated by the fact that TOG emission estimates and emission factors are not always consistent in terms of the set of compounds that they include. Emission estimates and emission factors may be representative of TOG, total organic compounds (TOC), volatile organic compounds (VOC), HC, or some other set of compounds. In many cases, the kinds of compounds included in an emission estimate may not be clear. During the profile development work performed for this study, an attempt was made to develop profiles that represented all organic gas emissions and that were therefore compatible with TOG emission estimates.

### PROFILES DEVELOPED THROUGH ANALYTICAL WORK

At the beginning of this task, two general types of source testing and analysis were considered—one directed at improving emission estimates and one directed at improving speciation. After studying the options, it was concluded that all testing should be directed toward improving speciation. This conclusion was based on several considerations.

The current set of species profiles used in the emission inventory is uncertain. In fact, speciation is considered to be the most uncertain element in the inventory. Profiles were originally developed on the basis of limited testing and analysis. Because the number of profiles currently used in the inventory is limited, their applicability to many of the source categories in the inventory is also questionable.

The current set of emission factors is, in most cases, based on relatively extensive testing programs. Emission factors that have received little attention usually represent unusual source types that do not involve a large amount of emissions (e.g., a specialized chemical process).

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A number of profiles can be developed through laboratory testing using limited field sampling. The cost effectiveness of this approach in reducing uncertainties is much better than that of emission rate testing, which generally requires extensive field sampling.

# Selection of Source Categories for Speciation Testing

After the decision was made to devote Task 3 efforts to speciation testing and analysis, the source categories to be tested were selected using the following considerations:

The improvement in the certainty of TOG speciation expected to result for a particular source category.

The total amount of ROG emissions that would be affected by the analytical work.

The estimated reactivity of the emissions being reviewed.

The estimated cost of the investigations, tests, and analyses.

On the basis of these considerations, the following source categories and general approaches were selected.

Gasoline--Analyses of both liquid and vapor samples for each product type (e.g., unleaded regular, leaded regular, etc.) and for both winter- and summer-blend gasolines were performed.

Industrial surface coatings—Samples of the most widely used coating types (e.g., lacquer, enamel, etc.) were obtained from both surface coating facilities and coating manufacturers. Composite samples for the different coating types were developed and analyzed.

Asphalts—slow cure and medium cure—Composite samples of two types of asphalt were developed and analyzed. However, after analysis it was found that some of the samples had been purposely altered at a refinery before shipment. As a result, the species profiles developed in this study deserve further attention.

Architectural surface coating--water-based coatings--A composite sample of water-based architectural coatings was developed. The volatile portion of the coating was then distilled off and analyzed.

Architectural surface coating--solvent usage--A sample of solvents used in conjunction with architectural coatings was developed and analyzed.

Internal combustion engines--reciprocating-natural gas--Two different exhaust gas samples from stationary gas-fired IC engines were obtained and analyzed using gas chromatography. One of the samples was also analyzed for aldehydes.

# Sampling Methodology

An integral part of the speciation test program was the determination of the composition of the samples to be analyzed and their acquisition. The following paragraphs discuss the makeup of the composite samples, the process used to obtain each sample, and the development of the final composites.

### Gasoline

The approach used to develop composite samples was identical for the winter-blend and summer-blend gasoline samples. Composite samples were developed for each product type on the basis of total gasoline sales for the five largest refiners in California. The composition of the samples is presented in Table 5-1 by product type. The sales data were obtained from the 1979-1980 annual report of the California State Board of Equalization. The samples were obtained at service stations in the SOCAB. The winter- and summer-blend samples were taken in December 1983 and May 1984, respectively.

One gallon of gasoline was dispensed prior to pouring a sample. The gasoline was then dispensed into a chilled 40 milliliter (ml) glass container with a teflon seal cap. These samples were shipped in a refrigerated ice chest for next-day delivery to Radian's laboratories. Once received, the samples were stored in a walk-in freezer at  $-15\,^{\circ}\text{C}_{\, \cdot}$ composite samples were developed inside the walk-in freezer and the transfers were made using a graduated cylinder. All containers were kept tightly sealed except during the actual transfer process. The composite samples remained in the walk-in freezer until they were analyzed. All composite samples except the heated and agitated vapor samples were developed and analyzed within 14 days of the date they were obtained. Difficulties were encountered during the initial analyses of the heated and agitated vapor samples because of the high concentrations of hydrocarbons. The gas chromatograph (GC) had not been set to analyze such high concentrations. This resulted in the GC being removed from operation until these high concentrations could be purged. Once the GC was back in operation, scheduling constraints resulted in the samples being stored in a freezer for approximately 60 days before the final analyses were performed.

TABLE 5-1. Formulation of gasoline composite samples.

<

	Annual	-	7 7 7 7 1	70 700	papeal
Top Five Companies by Taxable Sales	<pre>1979-1980 Taxable Sales (1000 gallons)*</pre>	Unleaded Regular (% of Sample)	Unleaded Premium (% of Sample)	Regular (% of Sample)	Premium (% of Sample)
Chevron USA	2,052,202	28.2	34.4	34.4	1
Shell Oil Co.	1,566,939	21.5	26.2	26.2	}
ARCO	1,358,944	18.6	22.7	22.7	;
Union Oil Co.	1,313,394	18.0	1	1	100.0
Mobil Oil Corp.	995,451	13.7	16.7	16.7	1
Subtotal	7,286,930				
Total Annual Taxable Sales in California	11,916,829	·			

\* Data obtained from the 1979-1980 annual report of the California State Board of Equalization.

## Industrial Surface Coatings

An important part of determining the makeup of the composite industrial surface coating samples concerned the nomenclature used for the coatings. There are seven SCCs currently used to categorize surface coatings. The coating descriptors for these SCCs are: 1) solvent-based paint, 2) water-based paint, 3) varnish/shellac, 4) lacquer, 5) enamel, 6) primer, and 7) adhesive. All of these SCCs have been used to some extent to categorize emissions from industrial surface coating in the inventory. The difficulty with this categorization system is that the distinction between some of the coatings is not clear. Thus, a surface coating sample could be categorized as one of several different coating types depending on the individual doing the categorization.

To clarify the categorization of industrial coatings, several coating manufacturers were contacted. These conversations resulted in the following general conclusions.

Paint is a generic classification and is used for the majority of coatings.

Water-based coatings are used on a limited basis in industrial coating operations.

Varnishes and shellacs are also used on a limited basis in industrial coating operations.

We decided that speciation testing should concentrate on four coating types: lacquer, enamel, primer, and adhesive. Attempts were made to obtain samples of these coating types from both industrial surface coating facilities and industrial surface coating manufacturers.

To develop the composite samples, the completed survey results were first reviewed. Because surveyed facilities had been randomly selected, we concluded that the coatings they used would be representative of coatings used throughout the SOCAB. Therefore, the surveyed facilities were ranked by emissions from each of the coating types. Facilities with the greatest emissions for each of the coating types were then contacted and samples of these coatings were requested.

This process resulted in the acquisition of three to five samples each of enamel, primer, and lacquer. The volatile portions of these samples were distilled off using a method described at the end of this section. These distillates were then combined in proportion to the emissions of the coating type and facility from which they were obtained. For example, if facility X had 200 tons/year of emissions from enamel usage and facility Y

had 100 tons/year of emissions from enamel usage, then the distillates from the enamel samples would be combined in a two to one ratio. Table 5-2 presents the makeup of the three composite samples that were developed. Also presented in this table is the volume that was distilled from 5.0 grams of the sample coating.

These composite samples were developed from coating types for cases in which three or more samples had been obtained from industrial surface coating facilities. To obtain additional samples of each coating, surface coating manufacturers in the 1981 ARB industrial surface coating survey were also contacted. Samples from nine coating manufacturers were requested for the following coating types:

varnish/shellac lacquer enamel primer adhesive

Only six of the nine manufacturers actually supplied coating samples and in most cases, the manufacturer produced only one or two of the coating types. Most manufacturers were concerned that the samples they provided might not be representative of the general coating type. Representative industrial surface coating samples were difficult to obtain because each coating is formulated for a specific application.

Two additional composite samples—an adhesive and an enamel sample—were also developed. Two samples of adhesive were obtained from industrial coating facilities and one sample was obtained from a coating manufacturer. Because no other approach to developing a composite sample appeared justified, the distillates from these three samples were combined in equal amounts.

The second additional composite sample was an enamel sample. During the acquisition of samples from the coating manufacturers, enamel samples were those most frequently provided. For this reason, and because of the large amount of emissions categorized as enamels in the revised inventory, a composite sample of enamels from coating manufacturers was also developed. The formulation of the adhesive and enamel samples is presented in Table 5-3. Also presented in this table is the volume of distillate trapped from each coating sample.

The same general distillation process was used for all coating samples and was taken from the ASTM Method D 2369--Standard Test Method for Volatile Contents of Surface Coatings. A small quantity of the coating was weighed

TABLE 5-2. Formulation of composite coating samples obtained from industrial surface coating facilities.

Coating Type and Facility	Percent of Composite Sample	Volume of Distillate Trapped* (ml)
Enamel		
General Motors	46.0	3.3
Steelcase West	41.3	0.3
Utility Trailer	6.7	0.5
Questor Furniture	4.0	1.2
Western Metal	2.0	0.8
Primer		
General Motors	50.9	3.3
Questor Furniture	34.1	0.7
Douglas Aircraft	15.0	1.8
Lacquer		
Questor Furniture <sup>†</sup>	32.2	2.8
Questor Furniture§	31.9	3.0
H. W. Hull	21.2	2.0
General Motors	14.8	2.0

<sup>\*</sup> This column represents the amount of distillate that was produced from 5.0 grams of coating.

<sup>&</sup>lt;sup>†</sup> This sample was categorized as paint in the original inventory, but was found to be lacquer.

<sup>§</sup> This sample was not the same as that designated t; it was correctly categorized as lacquer.

TABLE 5-3. Formulation of composite coating samples obtained from industrial surface coating facilities and manufacturers.

Coating Type and Facility or Manufacturer	Percent of Composite Sample	Volume of Distillate Trapped* (ml)
Ename 1	<i>*</i>	
DuPont	25.0	0.6
PPG	25.0	0.1
Fuller-O'Brien	25.0	0.4
Mobil Chemical	25.0	0.1
Adnesive		
DuPont	33.3	1.6
Steelcase West	33.3	3.4
Advanced Structures	33.3	4.0

<sup>\*</sup> This column represents the amount of distillate that was produced from 6.0 grams of coating for the enamel samples and from 4.0 grams of coating for the adhesive samples.

in a flask, which was then heated to  $110^{\circ}\text{C}$  for 60 minutes. During this time, the contents of the flask were purged with a continuous stream of organic free nitrogen. The resulting distillate was captured in a cryogenic trap.

The advantage of using the same general approach (110°C for 60 minutes) as the standard method is that the standard method is used for the development of emission factors for coatings. Therefore, the organics that are analyzed for emission factor development and are assumed to evaporate as the coating is applied are the same organics that were analyzed for speciation purposes.

### Asphalts--Slow Cure and Medium Cure

An extensive investigation of the composition of different types of asphalt was conducted prior to making the recommendation for asphalt speciation testing. The result of this investigation was the recommendation to analyze one sample of slow-cure and one sample of medium-cure asphalt. It was also determined that only two refineries in the SOCAB currently produce cutback asphalts, and that the most widely used slow-cure asphalt is SC-800 and the most widely used medium-cure asphalt is MC-250. The actual production rates of these grades of asphalt, or of cutback asphalts in general, were not available for these refineries. As a result, composite samples of SC-800 and MC-250 were developed by combining equal amounts of these asphalts from the two refineries.

A distillation process similar to the one described for industrial surface coatings was then used to generate organic samples representative of the emissions during application. However, the temperatures at which these samples were heated were the average application temperatures of the asphalts obtained from CalTrans "Standard Specifications" published in January 1981. The temperatures that were used were approximately 100°C for SC-800 and 85°C for MC-250. The distillate was also cryogenically trapped for analysis.

# Architectural Surface Coating--Water-Based Coatings

The first step in developing composite samples of water-based architectural surface coatings was to obtain the 1980 ARB statewide architectural surface coating survey results, as presented in Table 5-4. Using the results for water-based coatings, four product types were identified as comprising the majority of the total emissions: interior flats, exterior flats, interior nonflats, and exterior nonflats. These four product types

ARB 1980 architectural coating survey results. TABLE 5-4.

PRODUCT TYPE	PRODUCT CODE	SOLVENT SALES 1000 GAL	WATER SALES 1000 GAL	SOLVENT EMISSIONS T/D	WATER EMISSIONS T/D	THIN CLEAN (1 PT/GAL) T/D	T0TAL EMISSIONS T/0
1 2 4 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	9	,	( ) ( )		•	!	
	001.	44.51	8445.53	12.	4.43	.05	4.65
Exterior Flats	110	78.45	7689.16	.38	4.79	60.	(0.5) (0.5)
Interior Nonflats	120	1353.19	3596,69	5.56	2.78	1.48	9,83
Exterior Nonflats	130	1291.43	3567,60	5.21	2.59	25.1	( E)
Clears	140	2315,37	56.49	15.11	.07	2.54	17.72
Primers-General	200	670.78	358,75	2.97	.23	74	3, 93
Sealers	210	383.07	193.50	2.50	.07		2.53
Waterproofers	215	486.23	31.42	3.89	.01	67	4.44
Undercoaters	220	440.84	374.58	2.11	· 60	. 48	2.73
whood Preservatives	300	617.61		4.68	00.		200
⇔Opaque Stains	310	536.80	1427.99	2.39	84	65.	3.53
Semi-Transparent Stains	320	1232.94	174.60	8.40	.16	1.32	9.89
Industrial Metal Primers	400	609.27	20.45	3,29	.01	79.	1000
Maintenance Finish-Lt. Duty	410	1062.06	60.28	5.57	.05	1.16	6.78
Maintenance Finish-Heavy Duty	420	498.15	8.55	2,38	.01	្រុះ	2,93
Mastic-Texture	500	4.98	115.60	.02	0.	[0.	. 050.
Mill White	520	15.44	.40	.07	00.	.02	60.
Roof Coating	530	2734.58	148.83	8.78	.02	00.	8.80
Swimming Pools	540	59.70	00.	.34	00.	.07	04.
Tilostic railits.	000	60	Ċ	Ċ		•	
Cich (Combie Auto) Cotting	300	60.77	00.	2;	, uŭ	50.	.12
Sign (Graphic Arts) coatings	5/0	34.83	01.	-	00.	•04	<u>.</u>
Multicolor	980	.42	40.00	00.	.15	00.	51.
Metallic	590	59.46	00.	.32	00.	.07	35
Fire Retardant	009	3.66	3.80	00.	00.	.00	.00
Aerosol	800	573.60	3,29	3.59	00.	.63	4.22
Other	006	.15	00.	00.	00.	00.	[J.
TOTAL		15141.40	26471.07	78.03	16.50	13.56	108.69
* Product not surveyed in 1980.	·						

represented 14.69 tons/day of the total 16.50 tons/day of TOG emissions from water-based architectural coatings in the state in 1980.

To develop a composite sample from the various manufacturers of these architectural coating types, the detailed confidential results from the 1980 ARB survey, which present sales and emission data by product type and company, were used. The individual coatings for the four product types were ranked by emissions, and the seven coatings with the largest amount of emissions were selected for the composite sample. The seven samples represented 5.68 tons/day, or 39 percent of the 14.69 tons/day of emissions from the four product types.

The seven individual coatings were purchased at retail outlets in Sacramento, California and were combined in proportion to their 1980 statewide sales to develop 210 ml composite samples; these composites were considered representative of the mix of water-based architectural coatings. The 210 ml composite samples were then spiked with two surrogate standards, acetonitrile for water soluble compounds and deuterated bromobenzene for non-water-soluble compounds, to determine the efficiency of the distillation process. The amount of each standard added was 0.25 grams, resulting in a concentration of approximately 0.1 percent for each standard. The sample was then distilled at 110°C for 60 minutes, which was also selected to be consistent with ASTM Method D 2369.

Architectural Surface Coating--Solvent Usage

The intention was to develop a composite sample for this source category that was representative of the solvents used in conjunction with architectural surface coatings. Based on discussions with ARB staff, it was decided that solvents should be purchased from the primary retail paint outlets in Sacramento used by painting contractors because contractors represent a large percentage of the total solvent used. The retail outlets that were recommended were Dunn-Edwards, Fuller-O'Brien, and Standard Brands. The managers at each of these outlets were requested to identify their largest selling solvents and the approximate percent of total solvent sales that each solvent represented. The solvents were then made into a composite on the basis of sales within each retail outlet and then combined in equal proportions for the three outlets. The formulation of the composite solvent sample is presented in Table 5-5.

Internal Combustion Engines--Reciprocating-Natural Gas

Exhaust gas samples from two different natural-gas-fired IC engines were obtained and analyzed. The first sample was taken from a White Superior

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TABLE 5-5. Formulation of the composite sample of solvents used with architectural coatings.

Retail Outlet and Solvent	Percent of Retail Outlet Sales	Percent of Sample Volume
Dunn-Edwards		•
Paint thinner	50	16.7
Lacquer thinner	35	' 11 <b>.</b> 7
Benzine	15	5.0
Total	100	33.4
Fuller-O'Brien		
Paint thinner	80	26.7
Lacquer thinner	10	3.3
Brush cleaner	10	3.3
Total	100	33.3
Standard Brands		
Paint thinner	75	25.0
Lacquer thinner	20	6.7
Terpentine	5	1.7
Total	100	33.4

8G-825 500 horsepower (hp) engine and collected in an evacuated metal cylinder for analysis by a flame ionization detector/photo ionization detector gas chromatograph (FID/PID GC). The sample was taken during a SCAQMD compliance test on 13 September 1983 while the engine was operating at approximately 70 percent load. After reviewing the data collected during the source test, the engine was considered to be operating under normal conditions.

The second sample was taken from a Waukesha L7042 600 hp engine. In this case, one sample was collected in an evacuated cylinder for GC analysis, and another sample was collected using midget impingers containing 0.009N  $\rm H_2SO_4$  for analysis of total aldehydes. The exhaust gas was passed through the impingers at a rate of approximately 1.0 liter per minute for one hour. This engine was also undergoing a compliance test at the time the samples were obtained. On the basis of data collected during this test, the engine was considered to be operating under normal conditions.

# Analytical Methodology

Many of the analytical procedures used during the speciation test program were developed in consultation with ARB chemists.

### Gasoline

Eight analyses were performed on each of the winter- and summer-blend gasolines for liquid and static vapor samples from each of the four product types. In addition, a heated and an agitated vapor sample of summer-blend gasoline were analyzed for a total of 18 gasoline analyses. To analyze the liquid gasoline samples, two microliters of each of the four composite gasoline samples were injected into four 2.8 liter stainless steel evacuated canisters. A heated injection port and a flow of zero grade nitrogen was used to insure complete volatilization and transfer into the canisters, which were then pressurized to four atmospheres. Five ml of this dilute gas phase mixture was then analyzed. The five-ml sample was cyrogenically concentrated with liquid oxygen and then thermally desorbed onto a 60-m fused silica capillary column at -50°C. This initial temperature was held for two minutes and then programmed up to 100°C at 6° per minute.

The analysis was performed on a dual detector FID/PID GC. The instrument was a Varian 3700 GC equipped with a FID and a 10.2 eV PID. The column effluent was split into the FID and PID at a ratio of four to one, respectively. The detector responses were plotted and integrated on a dual channel Varian 401 chromatographic data system. A second data system, using an Apple II minicomputer, identified the peaks by comparing the

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retention times and normalized response ratios with previously established standard values. The FID response was used to quantify each component. Response factors were determined using propane and hexane standards.

The static vapor samples were developed by allowing 30 ml of liquid and 10 ml of vapor to reach equilibrium at room temperature (23°C or 73°F). The heated vapor sample was developed by heating a static vapor sample to 35°C. The agitated vapor sample was produced by shaking a sample (also 30 ml liquid and 10 ml vapor) using a mechanical shaker for one minute and immediately analyzing the vapor. In all cases, a five microliter vapor sample was taken from the 40 ml container using a gas-tight syringe. This sample was then cryogenically trapped and analyzed using an FID/PID GC as described previously.

In addition to these tests, analyses were performed to evaluate the concentrations of ethylene dibromide (EDB) and other halogenated species in gasoline. Due to the relatively low concentrations of halogenated species relative to other organic species in gasoline, the FID/PIC GC analysis was unable to detect the halogenated species. Therefore, a detector that could selectively identify halogenated species was used to analyze two gasoline samples: the liquid summer-blend leaded regular and leaded premium composites. Only two samples were analyzed because this work was not included in the scope of the original analytical approach.

The analyses were performed using a Tracor 560/700A hall electroconductivity detector (HECD). A Ni-reactor tube was used at 900°C with a hydrogen reactant gas. A methanol electrolyte solution was also used at a flow rate of approximately 5 ml per minute. A certified gas-phase standard was employed to quantify all halogenated species except EDB since a gas-phase EDB standard was not available. Therefore, the response factor of 1,2-dichloroethane was used to quantify EDB. The EDB analyses showed concentrations of 0.0063 milligrams of carbon per ml for the leaded regular sample and 0.014 milligrams of carbon per ml for the leaded premium sample. In both cases, these EDB concentrations represented less than 0.01 percent by weight. Concentrations of all identified halogenated species are shown in Appendix D for the appropriate sample.

### Industrial Surface Coatings

The industrial surface coating samples were analyzed using both a GC and a gas chromatograph/mass spectrometer (GC/MS). Initially, the composite distillate from the various surface coating samples was injected directly into the GC. If the concentrations of compounds under these conditions were too high, the sample was diluted in a noninterfering solvent and injected again.

The GC/MS results were used to identify compounds and the GC was used with an FID for quantification. A 60-m fused silica capillary column was used on both instruments. The temperature program started at  $35^{\circ}$ C and was increased to  $310^{\circ}$ C at  $8^{\circ}$ C per minute.

Asphalts--Slow Cure and Medium Cure

The asphalt samples were analyzed using the same procedures and instruments as were used for industrial surface coatings.

Architectural Surface Coating--Water-Based Coatings

A 210 ml composite sample of water-based architectural coatings was distilled as described previously, resulting in approximately 112.2 ml of total distillate. Acetonitrile and bromobenzene-d $_5$  were used as recovery surrogates to estimate the percent of recovery during distillation. Total recovery was measured at 105 percent for acetonitrile and 130 percent for bromobenzene-d $_5$ . The distillate consisted of three fractions: an organic layer (2.0 ml) with a density less than water, an aqueous layer (110 ml), and an organic layer (approximately 0.2 ml) with a density greater than water. Organic speciation of this sample was developed through an analysis of the first two fractions. The 0.2 ml fraction was not chromatographable and probably consisted of some sort of polymeric substance. The analyses were performed by GC/MS using a 60-m capillary column. The temperature program had an initial temperature of 35°C for one minute and then was increased to 320°C at 10° per minute. Toluene-d $_8$  and phenol-d $_6$  were used as internal standards.

Architectural Surface Coating--Solvent Usage

The solvent sample was also analyzed using the same procedures and instruments as were used for industrial surface coatings.

Internal Combustion Engines -- Reciprocating - Natural Gas

The first of the two IC engine exhaust gas samples was analyzed using a FID/PID GC; the second sample was analyzed using both a FID/PID GC and the MBTH method. In the first method, the exhaust gas sample was injected directly onto a 60-m fused silica capillary column at  $-50^{\circ}$ C. This initial temperature was held for two minutes and then programmed up to  $100^{\circ}$ C at  $6^{\circ}$ C per minute.

The MBTH method (see Appendix C) was used to determine total aldehydes in the exhaust gas. In general, the method consists of trapping the aldehydes in a dilute sulfuric acid solution. The trapped aldehydes are then reacted in the laboratory with MBTH to form an azine. Oxidation of the excess MBTH forms a reactive cation, which adds to the azine to form a blue dye. The intensity of this blue dye is directly proportional to the concentration of the aldehydes. This procedure allows the aldehydes to be measured by a colorimetric method.

# Species Profile Development from Analytical Results

The unreduced results of the analyses previously described are presented in Appendix D. These results were used to develop species profiles for each of the source categories under consideration. Table 5-6 lists the 68 profiles developed from both test results and existing information.

Gasoline.

Species profiles for gasoline were developed directly from the analytical results. The units of the analytical results for liquid samples were milligrams of carbon per milliliter (mg-C/ml). The units of the analytical results for vapor samples were parts of carbon per thousand by volume (ppthv-C).

For vapor samples, to convert a concentration expressed as carbon to a concentration expressed as a specific compound, the following calculation was made:

$$C_{carb}(16/MW_{carb}) = C_{comp}$$

where

 $C_{carb}$  = Concentration expressed as carbon

 $C_{comp}$  = Concentration expressed as the compound

 $MW_{carb} = Molecular$  weight of the carbon atoms in the compound

For liquid samples, to convert a concentration expressed as carbon to a concentration expressed as a specific compound, the following calculation was made:

$$C_{carb} (MW_{comp}/MW_{carb}) = C_{comp}$$

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TABLE 5-6. New profiles developed in this study.

Profile _Code	Profile Description
701	LIQUID GASOLINE - UNLEADED REGULAR - SUMMER BLEND
702	LIQUID GASCLINE - LEADED REGULAR - SUMMER BLEND
703	LIQUID GASCLINE - UNLEADED PREMIUM - SUMMER BLEND
70 <b>4</b>	LIQUID GASOLINE - LEADED PREMIUM - SUMMER BLEND
705	GASOLINE VAPORS - UNLEADED REGULAR - SUMMER BLEND
706	GASOLINE VAPORS - LEADED REGULAR - SUMMER BLEND
707	GASOLINE VAPORS - UNLEADED PREMIUM - SUMMER BLEND
708	GASOLINE VAPORS - LEADED PREMIUM - SUMMER BLEND
709	LIQUID GASOLINE - COMPOSITE OF PRODUCT - SUMMER BLEND
710	GASOLINE VAPORS - COMPOSITE OF PRODUCT - SUMMER BLEND
711	INDUSTRIAL SURFACE COATING - COMPOSITE LACQUER
712	INDUSTRIAL SURFACE COATING - COMPOSITE ENAMEL
713	INDUSTRIAL SURFACE COATING - COMPOSITE PRIMER
714	INDUSTRIAL SURFACE COATING - COMPOSITE ADHESIVE
715	SLOW CURE ASPHALT
716	MEDIUM CURE ASPHALT
717	ARCHITECTURAL SURFACE COATING - WATER BASED PAINT
718	ARCHITECTURAL SURFACE COATING - COMPOSITE SOLVENT
719	INTERNAL COMBUSTION ENGINE - RECIPROCATING-NATURAL GAS FIRED
721	LIQUID GASOLINE - UNLEADED REGULAR - WINTER BLEND
722	LIQUID GASOLINE - LEADED REGULAR - WINTER BLEND
723	LIQUID GASOLINE - UNLEADED PREMIUM - WINTER BLEND
724	LIQUID GASOLINE - LEADED PREMIUM - WINTER BLEND
725	GASOLINE VAPORS - UNLEADED REGULAR - WINTER BLEND
726	GASOLINE VAPORS - LEADED REGULAR - WINTER BLEND
727	GASOLINE VAPORS - UNLEADED FREMIUM - WINTER BLEND
728	GASOLINE VAPORS - LEADED PREMIUM - WINTER BLEND
729	LIQUID GASOLINE - COMPOSITE OF PRODUCT - WINTER BLEND
730	GASOLINE VAPORS - COMPOSITE OF PRODUCT - WINTER BLEND
731	HEATED GASOLINE VAPORS - UNLEADED REGULAR - SUMMER BLEND
732	AGITATED GASOLINE VAPORS - UNLEADED REGULAR - SUMMER BLEND
751	ACRYLONITRILE-BUTADIENE-STYRENE (ABS) RESIN MFG.
752	POLYSTYRENE RESIN MFG.
753	STYRENE
754	CHLOROSOLVE
755 756	TRICHLOROTRIFLOUROETHANE
756	OIL AND GAS PRODUCTION FUGITIVES-LIQUID SERVICE
757 750	OIL AND GAS PRODUCTION FUGITIVES-GAS SERVICE
758 750	OIL AND GAS PRODUCTION FUGITIVES-VALVES-UNSPECIFIED SERVICE
759 760	OIL AND GAS PRODUCTION FUGITIVES-FITTINGS-UNSPECIFIED SERVICE
760 761	EVAPORATIVE EMISSIONS-DISTILLATE FUEL
761 762	EVAPORATIVE EMISSIONS-NAPHTHA
762 763	BTX (BENZENE/TOLUENE/XYLENE)
763	PHTHALIC ANHYDRIDE MFG - XYLENE OXIDATION
764 765	FLUOROCARBON - 12/11 MANUFACTURING
765	FLUOROCARBON - 23/22 MANUFACTURING
766	FLUOROCARBON - 113/114 MANUFACTURING
767	FLUOROCARBON - 11

TABLE 5-6 (concluded)

Profile Code	Profile Description
768 769 770 771 772 773 774 775 776 777 778 779 780 781 782 783 784 785 786 787	FLUCROCARBON - 113 FLUCROCARBON - 114 CHICROFLUCROCARBONS CARBON TETRACHLORIDE ORTHO-XYLENE FLUCROCARBON MFG - VALVES, PUMPS, ETC ISOBUTYL ACETATE ISOBUTYL ALCOHOL ISOBUTYL ISOBUTYRATE METHYL AMYL KETONE METHYL ISOBUTYL KETONE N-BUTYL ACETATE N-PROPYL ACETATE N-PROPYL ACETATE N-PROPYL ACETATE N-PROPYL ALCOHOL HEXYLENE GLYCOL INDUSTRIAL SURFACE COATING - SOLVENT BASED PAINT SYNTHETIC RUBBER MFG - STYRENE-BUTADIENE RUBBER ETHYLENE OXIDE METHYL ALCOHOL CARBON BLACK MANUFACTURING

where

 $MW_{comp}$  = Molecular weight of the compound

Values for  $\mathrm{MW}_{\mathrm{carb}}$  and  $\mathrm{MW}_{\mathrm{comp}}$  are presented next for some example compounds.

Compound	<sup>MW</sup> carb	MW <sub>c omp</sub>
Ethane	24	30
Butane	48	58
Butene	48	56

To convert liquid sample concentrations expressed as carbon to percent by weight, it was assumed that the difference between a weight percent expressed as carbon and a weight percent expressed as the specific compound was insignificant. This assumption is justified for two reasons:

The ratio of the molecular weight of a compound to the molecular weight of carbon in the compound  $({\rm MW_{comp}/MW_{carb}})$  does not vary much for the organics found in gasoline.

The imprecision introduced by this assumption is small relative to the overall imprecision of the analytical results.

Once this assumption was made, the calculation of weight and volume percents was straightforward. For liquid samples, the concentration of a compound in mg-C/ml was divided by the total nonmethane hydrocarbon concentration of the sample and multiplied by 100. This resulted in the weight percent of that compound. For vapor samples, this same calculation as applied to a concentration in ppthy resulted in the volume percent of the compound.

A total of eighteen composite gasoline samples were analyzed. Sixteen of these samples consisted of liquid and vapor samples for each of the four product types for both winter- and summer-blend gasolines. The additional two samples were a heated and an agitated vapor sample for summer-blend unleaded regular. The eighteen species profiles that were developed from these analytical results are presented in Appendix E-1.

Gasoline evaporative emissions in the current 1979 SOCAB inventory are not differentiated by product type. Therefore, it was necessary to develop composite summer- and winter-blend gasoline profiles representative of the 1979 product mix. Using 1979 data from the California Energy Commission production and consumption report (CEC, 1981) as presented in Table 5-7,

TABLE 5-7. Sales of gasoline products in California in 1979.

Product Type	1979 California Sales (1000 barrels)	Percent of Total Sales
Unleaded regular	100,178	39.8
Leaded regular	95,206	37.8
Unleaded premium	0	0.0
Leaded premium	56,587	22.4
Total	251,971	100.0

four liquid and vapor composite profiles were developed for the winterand summer-blend gasolines. These profiles are also presented in Appendix E-1.

Several comparisons were made to evaluate the new gasoline species profiles. The new profiles were first compared to the profile in the VOC species data manual (EPA, 1980) for evaporative gasoline emissions. This profile, No. 98, is quite similar to the new liquid gasoline profiles we developed. Assuming that profile No. 98 was based on an analysis of liquid gasoline, the profile often used for gasoline evaporative emissions to date is inappropriate for emissions of gasoline vapors. Furthermore, gasoline vapors, in contrast to the complete evaporation of liquid gasoline, make up the great majority of gasoline evaporative emissions.

To further evaluate the new gasoline profiles, a comparison was made of the weight percents of the primary compound groups in each sample (Table 5-8). This comparison shows that there is a good agreement in the trends among all samples analyzed except the heated vapor sample. The low paraffin content in the heated vapor sample may be the result of the loss of some of the lowest molecular weight compounds during heating or storage. Trends for the liquid and static vapor samples are as follows:

The paraffin content in static vapor samples is much higher than in liquid samples. This trend occurs because the lower molecular weight and more volatile paraffins tend to move into the vapor phase more readily than do the aromatics.

The unleaded liquid samples tend to have a higher aromatic content than do the leaded liquid samples.

The comprehensive analyses performed for gasoline samples represent more extensive results than originally called for by the scope of the project. Profiles of liquid gasoline are now available for evaporative losses resulting from such occurrences as gasoline spillage. Similarly, related gasoline profiles for static vapor emissions are available for such conditions as storage tank losses.

Industrial Surface Coatings

Five composite industrial surface coating samples of the following coating types were analyzed:

Lacquer Primer

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TABLE 5-8. Comparison of gasoline species profiles developed in this study.

		Percent by	Weight
Sample	Paraffins	Olefins	Aromatics
Summer Blend			
Liquid		-	
Unleaded regular	51.2	8.0	39.7
Leaded regular	62.4	8.5	28.9
Unleaded premium	48.5	7.6	43.9
Leaded premium	56.3	3.5	40.2
Vapor			
Unleaded regular	85.2	13.4	1.3
Leaded regular	84.2	14.9	0.9
Unleaded premium	83.4	13.3	3.2
Leaded premium	92.9	5.2	1.9
Heated Vapor			
Unleaded regular	37.9	10.0	52.0
Agitated Vapor			
Unleaded regular	54.8	13.0	32.2
Winter Blend			
Liquid			
Unleaded regular	51.0	7.2	41.8
Leaded regular	57.6	10.0	32.4
Unleaded premium	48.3	7.3	44.3
Leaded premium	61.8	4.2	34.0
Vapor			
Unleaded regular	85.4	10.7	3.9
Leaded regular	80.7	15.2	4.2
Unleaded premium	83.2	12.6	4.3
Leaded premium	92.8	4.7	2.6

Adhesive

Enamel obtained from surface coating facilities

Enamel obtained from surface coating manufacturers

Species profiles were developed for each of these samples. A sixth industrial surface coating profile, representative of general enamel samples, was also developed by taking an arithmetic average of the two enamel profiles. The new industrial surface coating profiles that were assigned profile numbers are presented in Appendix E-2.

The following conclusions can be drawn from a comparison of the new species profiles with the existing profiles for these coating types.

The profiles of the two enamel samples analyzed in this study contain significant differences. This demonstrates that composite samples of industrial coatings composed of three to five individual coating samples are probably not representative of general coating types. A greater number of individual coating samples appears necessary to achieve a representative coating type.

The new industrial coating profiles have a high aromatic content, especially of toluene and xylene. This is in contrast to the currently used profiles, which consist primarily of "mineral spirits." Mineral spirits refer to a mixture of primarily  $\mathbf{C}_6$  to  $\mathbf{C}_{10}$  straightchain paraffins.

Asphalts--Slow Cure and Medium Cure

Species profiles were developed for composite slow cure and medium cure cutback asphalt samples. However, after the samples were analyzed and profiles were developed, it was discovered that one of the individual samples used in each composite sample had been purposely altered by the supplier. Unknown to us before our analyses, these samples had been distilled and then washed with caustic in an attempt to remove the lighter, more volatile compounds. Because the profile currently used for cutback asphalts is the result of a grab sample of emissions taken from a hot-mix asphalt pile, we feel the new profiles, though still deserving future investigation, better represent emissions from cutback asphalts. The new profiles are shown in Appendix E-3.

Architectural Surface Coating--Water-Based Coatings

The new species profile for water-based architectural surface coating is presented in Appendix E-4. These results indicate that three compounds make up over 65 percent of the organics in the sample. These three compounds are listed next along with their approximate weight percents in the sample.

2-Methylpropanoic acid, 2,2-dimethyl-1-(2-hydroxy-1-methylethyl) propyl ester: 23.2 percent by weight

2-Methylpropanoic acid, 3-hydroxy-2,4,4-trimethylpentyl ester: 23.8 percent by weight

1-Butanol: 19.5 percent by weight

It is important to note that the identification and quantification of these compounds were not confirmed by the use of standards. Therefore, particularly for the more complex species, the results should be considered an indication of the type of species present in the sample, but not a positive identification of the particular species.

Architectural Surface Coating--Solvent Usage

Using the analytical results of the composite solvent sample, the species profile presented in Appendix E-5 was developed. This profile consists of a wide variety of different compounds, with few halogenated hydrocarbons.

Internal Combustion Engines--Reciprocating-Natural Gas

Two gas-fired IC engine exhaust samples were analyzed using gas chromatography, and one of these samples was also analyzed for aldehyde content. The measured aldehyde content was assumed to be applicable to both exhaust samples, and was used to develop two profiles. These two profiles were then arithmetically averaged to develop the composite species profile shown in Appendix E-6. As can be seen from the composite profile, the primary constituent of the organic gases is methane, but other low molecular weight paraffins and olefins are present in measurable quantities.

PROFILES DEVELOPED FROM LITERATURE, SURVEY RESULTS, AND OTHER EXISTING DATA

In addition to developing species profiles through analytical work, efforts were made to develop profiles from existing data. Moreover, the speciation of TOG emissions in the inventory was improved significantly

through changes to SCCs used to identify emission sources and to assign speciation profiles. An example of this improvement is a source that originally was identified with SCC 3-01-999-99, chemical manufacturing-unspecified. After reviewing the survey results and emission fee data for the appropriate facility, this SCC was changed to a more specific SCC such as 3-01-018-01, chemical manufacturing-plastics production-polyvinyl chloride. Use of this revised SCC resulted in the assignment of a more specific and reliable speciation profile which, in turn, led to an improved inventory.

To develop new organic gas species profiles from existing data, several steps were taken.

Computerized literature search

Manual search of Radian's library

Contacts with individuals at Radian and other companies and agencies

Review of data obtained in the Task 2 survey

# Computerized Literature Search

Several computerized data bases were searched using combinations of various key words:

DOE Energy
National Technical Information Service
Chemical Abstracts Search
Compendex
Pollution Abstracts

Various combinations of keywords were used to emphasize organic gas speciation. However, no documents were identified that specifically addressed the development of a species profile. Those documents that did treat organic gas speciation did so in a limited manner (e.g., documents might have emphasized the speciation of polynuclear aromatic compounds or the speciation of  $C_2$  through  $C_6$  hydrocarbons, but not full organic gas speciation). Therefore, no profiles were added to the inventory through this effort.

# Other Existing Data

During additional data collection efforts, a number of documents were identified that contained speciation information. However, in most cases

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these data were too limited to allow the development of species profiles. The documents we reviewed are briefly described:

Control Technique Guideline Documents. These documents, in particular those related to surface coating (e.g., EPA, 1977), normally identify the specific organic compounds emitted from the source category of interest. However, no quantification of the relative amounts of these compounds is provided.

Vapor Control Methods for Gasoline Marketing. This report (Radian, 1975) contains useful information on gasoline vapor speciation as well as on the hydrocarbon concentration gradient that exists between the liquid surface and the outlet in underground gasoline storage tanks.

Analysis of Polynuclear Aromatic Hydrocarbon (PNA) in Liquid Fuels. This document (Radian, 1979) contains a detailed analysis of PNAs in jet fuels and diesel fuels. However, full speciation of these fuels was not obtained in this study.

Assessment of Emissions from Petroleum Refining. In this study (Radian, 1980), more than 20 analyses of refinery products and intermediates were made. Because the emphasis was on quantifying aromatics, full speciation was not available.

1974 Composition of Los Angeles Gasolines. In this analysis (ARB, 1975), full hydrocarbon speciation was developed for several gasoline samples.

Vehicular Evaporative Emissions. Hydrocarbon speciation was obtained in this study (Mayrsohn and Crabtree, 1975) for numerous gasoline samples.

Source Reconciliation of Hydrocarbons in the SOCAB. This document (ARB, 1976) presents limited speciation data for a variety of fuels and solvents. Compounds greater than  ${\rm C}_6$  are not differentiated beyond the number of carbon atoms present.

In addition to these documents, two ongoing studies were identified that may later provide valuable speciation data. Mr. John Wood at the South Coast Air Quality Management District is investigating the speciation of emissions from surface coating activities before and after incineration devices. In addition, Dr. Charles Spicer at Battelle Laboratories in Columbus, Ohio is performing a major study of the composition of jet engine exhaust.

# Species Profile Development

The documents just described provided a useful perspective, but were not directly used to develop profiles. Nevertheless, a total of 37 new species profiles labeled as profile codes Nos. 751 through 787 were developed from existing data. More than half of these profiles consisted of one specific compound, and thus no documentation of their development is necessary. A brief documentation of the development of the remaining profiles is provided next. Each of these profiles is presented in Appendix E-7.

Profile 751--Acrylonitrile-Butadiene-Styrene (ABS) Resin Manufacturing. On the basis of a conversation with one of the environmental coordinators at an ABS resin plant, emissions from ABS manufacturing were assumed to be evaporative emissions from a liquid mixture at standard conditions. This liquid mixture was further assumed to be 20 percent acrylonitrile and 80 percent styrene. Profile 751 was developed on the basis of the vapor pressures of these materials at standard conditions.

<u>Profile 752--Polystyrene Resin Manufacturing</u>. From the survey responses and literature review, we found that (1) styrene and ethylbenzene are present in emissions from polystyrene manufacturing, and (2) styrene is the major constituent of these emissions. Profile 752 was developed on the basis of this information.

Profiles 756 through 759--0il and Gas Production Fugitives. The first step in developing these profiles was to determine the ratio of liquid service and gas service components for valves and fittings in the SOCAB. This was done using information from two sources. The number of wells in the SOCAB in each of four gas-to-oil ratio (GOR) categories was determined using the "67th Report of the State Oil and Gas Supervisor, 1981." Data for 1981 were felt to be appropriate because the profiles that were developed would be used for both 1979 and future inventories. These well counts were then used in conjunction with information on the number of liquid and gas service components for each GOR category. This information was developed by KVB and was contained in a ARB technical note entitled, "Estimation of Fugitive PROC Emissions from Oil and Gas Operations and Gas Processing Plants." The calculation of the ratio of liquid and gas service valves and fittings in the SOCAB is shown next.

# Well Populations by GOR in the SOCAB

Gas-to-Oil Ratio <100 100-400 400-900 >900 Number of Wells 2,203 3,271 3,054 1,204

These data were then used in conjunction with information on the number of liquid and gas service components for each GOR category, as shown next.

# Valve and Fitting Populations per Well by GOR

Gas-to-Oil Ratio	<100	100-400	400-900	>900
Liquid Service Valves/Well	31.2	23.1	24.7	10.9
Gas Service Valves/Well	0.7	16.4	18.1	10.5
Liquid Service Fittings/Well	24.2	197.4	145.8	63.4
Gas Service Fittings/Well	5.2	98.4	108.0	72.8

Using these data, component populations for gas and liquid service valves and fittings in the SOCAB were developed:

# Valve and Fitting Populations in the SOCAB

Gas-to-Oil Ratio	<100	100-400	400-900	>900
Liquid Service Valves	68,734	75,560	75,434	13,124
Gas Service Valves	1,542	53,644	55,277	12,642
Liquid Service Fittings	53,313	645,695	445,273	76,334
Gas Service Fittings	11,456	321,866	329,832	87,651

Therefore, the total ratio of liquid service to gas service valves is 232,852 to 123,105 or 0.654 to 0.346. The total ratio of liquid service to gas service fittings is 1,220,615 to 750,805 or 0.619 to 0.381.

The second step in developing profiles for oil and gas production was to develop composite profiles based on the liquid service and gas service species data developed for API by Rockwell. These species data, from page E-4 of the API/Rockwell report, are presented next.

Weight Fraction of Species

	<sup>C</sup> 1	$c_2$	c <sub>3</sub>	c <sub>4</sub>	c <sub>5</sub>	c <sub>6</sub> +
Liquid Service Gas Service	0.375 0.614	0.064 0.079				

Using these species data and the ratio of gas and liquid service components, the following species data were developed:

Weight Fraction of Species

	c <sub>1</sub>	$c_2$	c <sub>3</sub>	C <sub>4</sub>	С <sub>5</sub>	C <sub>6</sub> +
Valves	0.458	0.069	0.090	0.067	0.044	0.272
Fittings	0.466	0.070	0.089	0.065	0.043	0.267

To obtain detailed profiles, this information was used in conjunction with Table 4-03-001C from the VOC species data manual (EPA, 1980). We assume that the resulting profiles contain a fair amount of uncertainty. The following steps were taken in developing these profiles:

- (1) The  $C_4$  component was divided into n-butane and isobutane in proportion to the weight percentages of these compounds in Table 4-03-001C of the VOC species data manual.
- (2) The C<sub>6</sub><sup>+</sup> component was divided into isomers of hexane, heptane, and octane; C-7 and C-8 cycloparaffins; and benzene in proportion to the weight percentages of these compounds in Table 4-03-001C of the VOC species data manual.

Profiles 760 and 761--Evaporative Emissions-Distillate Fuel and Naphtha. These profiles were obtained from the species data manual. Profile 760 consists of Table 3-06-008F and profile 761 consists of Table 3-06-008G. The series of profiles 3-06-008A through 3-06-008G were developed for fugitive leaks from valves and flanges carrying various petroleum products and intermediates. Because data for emissions from storage tanks containing a variety of petroleum products do not exist, these data were felt to be the best available for storage tank evaporative emissions.

The current profile for distillate fuel is based on a test of weed oil and consists of most naphthalenes. Profile 760, which consists primarily of paraffins with no aromatics, is considered to be more representative of

distillate fuel because aromatics are normally not added to distillate fuel in order to lower the self-ignition point of the fuel.

Profile 761, which contains both paraffins and aromatics, is believed to be representative of evaporative emissions from middle distillate petroleum intermediates such as naphtha.

Profile 762-BTX (Benzene/Toluene/Xylene). BTX is a common term for a mixture of aromatic compounds including benzene, toluene, and xylene. Although it is known that the composition of BTX is variable, no data were available on the average composition of BTX. As a result, it was assumed for this profile that BTX emissions contain the three main constituents in equal proportions.

Profile 763--Phthalic Anhydride Manufacturing-Xylene Oxidation. The composition of emissions from phthalic anhydride manufacturing, as well as the primary and secondary chemical reactions involved, is discussed in several documents. This information was reviewed and a rough estimate of the composition of these emissions was made. The compounds that are emitted are similar in chemical structure. Therefore, inaccuracies in the weight percent of each compound are less significant than they would otherwise be. The documents reviewed include

"Report on the Emissions from a Gas-Fired Afterburner Serving a Phthalic Anhydride Plant," SCAOMD, 1976.

EPA Publication No. AP-42 (including supplements 1 through 14), EPA, 1983.

Industrial Chemistry, 7th Edition, Kent, 1974.

Industrial Chemicals, Lowenheim and Moran, 1975.

"Potential Pollutants from Petrochemical Processes," Monsanto Research Corp., 1973.

Profiles 764 through 766--Fluorocarbon Manufacturing. These profiles were developed using two sources of information: "Organic Chemical Manufacturing" prepared by IT-Enviroscience for EPA and the 1979 SCAQMD emission fee data for a fluorocarbon manufacturing facility. These two information sources presented different estimates for the composition of emissions from each of the types of fluorocarbon manufacturing. The individual estimates were therefore averaged to develop profiles 764 through 766. The derivation of these profiles is shown in Table 5-9.

TABLE 5-9. Derivation of Profiles 764 through 766.

	Percent	by Weight	
	Organic Chemical	SC AQMD	
Product Constituents	Manufacturing Estimate	Estimate	Average
\			
Fluorocarbon 12/11			
Fluorocarbon 11		5.8	2.9
Fluorocarbon 12	99.0	47.2	73.1
Fluoroca <u>rbon 13</u>	1.0	47.1	24.0
Fluorocarbon 23/22			
Fluorocarbon 22	33.3	1.6	17.4
Fluorocarbon 23	66.7	98.4	82.6
Fluorocarbon 113/114/			
Fluorocarbon 113	39.6	NA*	39 • 6
Fluorocarbon 114	58.0	NA	58.0
Fluorocarbon 115	2.4	NA	2.4

<sup>\*</sup> NA indicates that data were not available.

Profile 773--Fluorocarbon Manufacturing-Miscellaneous Emissions. This profile was developed to represent a composite of all TOG emissions related to fluorocarbon manufacturing. To develop this profile, a facility primarily engaged in fluorocarbon manufacturing that had been surveyed in Task 2 was identified (No. 435.19). The inventory for this facility after inclusion of the recommended changes was reviewed, and all devices and SCCs that specifically related to fluorocarbon manufacturing were identified. A composite profile was then developed by weighting the profiles for each of these SCCs in proportion to their rate of emissions at the facility. The devices, SCCs, emissions, and percent that each profile was weighted are as follows:

<u>Device</u>	SCC	TOG Emissions (tons/year)	Percent of Total
101 -	3-01-127-02	1.8	6.6
102	3-01-127-03	12.5	46.1
103	3-01-127-12	2.0	7.4
104	3-01-127-13	7.3	26.9
105	3-01-127-14	0.3	1.1
106	3-01-127-11	3.2	11.8
Total		27.1	99.9

Profile 783-Industrial Surface Coating-Solvent Based Paint. From conversations with industrial surface coating manufacturers, it was concluded that the term "paint" is a general word used to categorize surface coatings that would more appropriately be classified as enamels, primers, and lacquers. Therefore, a composite profile was developed for paint from the profiles for enamels, primers, and lacquers. To estimate the relative amounts of enamels, primers, and lacquers that were used in the SOCAB and might be classified as paints, the surveyed facilities from the EIS file and their emission fee data were reviewed. These data provided a detailed breakdown of emissions by coating type. Using the survey results and emission fee data, total emissions for the surveyed facilities were calculated by coating type. These emission totals were: enamel--1003.6 tons/year; primer--356.8 tons/year; lacquer--207.0 tons/year.

The species profile developed for paint was a composite of the profiles for these coatings in proportion to their individual emission totals. The profiles that were used for the composite and their weighting are as follows.

Profile Number and Name	Percent of Composite Profile
711Composite Lacquer	13.2
712Composite Primer	22.8
713Composite Enamel	64.0

Profile 784--Synthetic Rubber Manufacturing-Styrene-Butadiene Rubber.

AP-42 (EPA, 1983) contains a discussion of emissions from synthetic rubber manufacturing which states that styrene-butadiene rubber contains less than 45 percent styrene. This report also states that VOC emissions from the various processes are primarily styrene and butadiene. From this information, a rough estimate of the composition of organic gas emissions of synthetic rubber manufacturing was made.

Profile 787--Carbon Black Manufacturing. AP-42 (EPA, 1983) presents a species profile for the main process vent from carbon black manufacturing by the oil furnace process. The oil furnace process accounts for over 90 percent of carbon black manufacturing in the United States, and the majority of emissions from this process come from the main process vent. Therefore, this species profile was assumed to be applicable to general carbon black manufacturing.

## 6 TASK 4: INVENTORY IMPROVEMENT

PROCEDURES USED TO DEVELOP RECOMMENDATIONS FOR CHANGES TO THE 1979 INVENTORY

This section describes the procedures used to develop recommendations for changes to the 1979 inventory on the basis of new information described in Sections 4 and 5. The first step in developing these recommendations was to review existing information in the Emission Data System (EDS), including turnaround documents (TADs) and summary inventory printouts for the facilities under consideration. The TADs contain a complete set of the information in EDS, whereas the summary inventory printouts are organized by SCC for each facility and contain the following information:

Device identification number Permit identification number Process rate Process rate units  $\mathrm{NO}_{\mathrm{X}}$  emission factor  $\mathrm{NO}_{\mathrm{X}}$  emissions TOG emissions TOG emissions

During the initial review of the TADs and summary inventory printouts, data for devices possibly needing revision, such as entries for unspecific SCCs, were identified. In addition, if a SCC for a facility did not appear reasonable considering the SIC code, the appropriate device was identified.

The next step was to compare the existing inventory with other available emission information—primarily that contained in the Task 2 survey responses and the 1979 emission fee data. On the basis of this comparison, changes to the inventory were developed for specific point sources, which in many cases required the development of new SCCs.

<sup>\*</sup> Some of this information (e.g., permit number and process rate) is not available for devices at all facilities.

# Description of Recommended Changes

Study recommendations for changes to the inventory were generated for each facility in a document containing four major sections. The first section of each document contained identifying facility information and the name of a facility contact. This introductory information included

Facility identification number

Company and facility name

Facility location

Name, title, and telephone number of an appropriate individual

The second section, labeled recommended changes, contained instructions for adding, deleting, or modifying data for devices at the facility. The third section, documentation, described the supporting activity data, emission factors, and general approach that were used to calculate the revised emissions. The documentation section was correlated with the recommended changes section through the use of consistent entry numbers. The final section, uncertainties after recommended changes, presented uncertainty codes and associated comments for devices and is discussed in detail later in this section. The facility-specific recommended changes were developed in this manner so that ARB staff would be able to use this information to make corresponding changes to the EDS data base at a later date.

An example of the format and content of the document developed for each facility receiving inventory changes is presented in Exhibit 6-1 for a surface coating facility with relatively low emissions. This facility is only one of the many types of facilities that were surveyed and revised in the inventory. The specific changes recommended for a more representative set of six facilities, together with the summary inventory printouts for these facilities, are presented in this same format in Appendix F.

## Development of New SCCs

During the development of inventory changes to individual facilities, we generated and entered SCCs that had not been used before. These new SCCs permitted a more specific identification of certain emission sources at the surveyed facilities, which improved the overall inventory. Better identification of the operations at these facilities through the use of new SCCs resulted in the assignment of more specific speciation profiles, which, in turn, reduced inventory uncertainty. Thus, for each new SCC, either an existing or a new species profile developed for organic gases in this study was assigned. The 148 new SCCs and their assigned profile codes are presented in Table 6-1.

### EXHIBIT 6-1.

An Example of the Changes Recommended for a Specific Facility.

Facility ID: "Number"
"Facility Name"

Facility Location: "Address"

Contact Name:
"Name"
"Title"

"Phone No."

### A. Recommended Changes

- (1) Delete Device 1 with SCC 4-01-002-99 Organic Solvent-Vapor Degreasing - Open Top - Miscellaneous. TOG Emissions - 0.0 ton/yr.
- (2) Delete Device 2 with SCC 4-02-999-99 Organic Solvent -Surface Coating - Miscellaneous. TOG Emissions - 1.7 ton/yr.
- (3) Add Device 101 with SCC 4-01-002-03 Organic Solvent -Vapor Degreasing - Open Top - Perchloroethylene. TOG Emissions - 6.1 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (4) Add Device 102 with SCC 4-02-001-01 Organic Solvent -Surface Coating - Paint General - Solvent Base Ctng. TOG Emissions - 0.1 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (5) Add Device 102 with SCC 4-02-004-01 Organic Solvent -Surface Coating - Lacquer - General. TOG Emissions - 0.3 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.

(continued)

## EXHIBIT 6-1 (continued)

- (6) Add Device 102 with SCC 4-02-005-01 Organic Solvent -Surface Coating - Enamel - General. TOG Emissions - 4.6 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (7) Add Device 102 with SCC 4-02-006-01 Organic Solvent -Surface Coating - Primer - General. TOG Emissions - 1.3 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (8) Add Device 103 with SCC 4-02-009-01 Organic Solvent -Surface Coating - Thinning Solvent - Miscellaneous. TOG Emissions - 5.5 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (9) Add Device 103 with SCC 4-02-009-06 Organic Solvent -Surface Coating - Thinning Solvent - Cellosolve. TOG Emissions - 1.4 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.
- (10) Add Device 103 with SCC 4-02-009-22 Organic Solvent -Surface Coating - Thinning Solvent - Toluene. TOG Emissions - 1.8 ton/yr. Assumed operating hours - 8 hr/day, 5 day/wk, 52 wk/yr.

### B. Documentation

- (1) This device was deleted to allow for more specific identification of each solvent that was used in vapor degreasing in 1979.
- (2) This device was deleted to allow for more specific identification of each solvent that was used in surface coating in 1979.
- (3) 1979 usage of perchlorethylene comes from questionnaire responses.
  An emission factor of 13.5 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table.
  (898 gal/yr) (13.5 lb/gal) (1 ton/2000 lb) = 6.1 ton/yr.
- (4) 1979 usage of epoxy enamel comes from questionnaire responses. An emission factor of 5.5 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (33 gal/yr) (5.5 lb/gal) (1 ton/2000 lb) = 0.1 ton/yr.

(continued)

### EXHIBIT 6-1 (concluded)

- (5) 1979 usage of polyurethane comes from questionnaire responses. An emission factor of 5.2 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (129 gal/yr) (5.2 lb/gal) (1 ton/2000 lb) = 0.3 ton/yr.
- (6) 1979 usage of enamel comes from questionnaire responses. An emission factor of 4.5 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (2029 gal/yr) (4.5 lb/gal) (1 ton/2000 lb) = 4.6 ton/yr.
- (7) 1979 usage of primer comes from questionnaire responses. An emission factor of 5.0 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (519 gal/yr) (5.0 lb/gal) (1 ton/2000 lb) = 1.3 ton/yr.
- (9) 1979 usage of butyl cellosolve comes from questionnaire responses. An emission factor of 7.5 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (365 gal/yr) (7.5 lb/gal) (1 ton/2000 lb) = 1.4 ton/yr.
- (10) 1979 usage of toluene comes from questionnaire responses. An emission factor of 7.2 lb/gal was taken from SCAQMD, Fee Form B-3, Emission Factor Table. (495 gal/yr) (7.2 lb/gal) (1 ton/2000 lb) = 1.8 ton/yr.
- C. Uncertainties After Recommended Changes

Device(s)	Uncertainty <u>Code(s)</u>	Comments
101-108	42	
101-108	1	Usage has been extrapolated from 1982 data
102	34	Made up of epoxy enamel
103	34	Consists of polyurethane
107	34	Consists of butyl cellosolve

TABLE 6-1. New source classification codes developed in this study.

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Description of SCC

Profile Code

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200405       200405       2006044       3006044       1009220       1009221       1009221       1009221       1009221       1001404       101404

Code

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GAS PRODN.-NATURAL GAS PRODN-PUMP SEALS
GAS PRODN.-NATURAL GAS PRODN-COMPRESSOR SEALS
GAS PRODN.-NATURAL GAS PRODN-COMPRESSOR SEALS
GAS PRODN.-NATURAL GAS PRODN-WELL HEADS
GAS PRODN.-NATURAL GAS PRODN-WELL HEADS
SOLVENT-VAPOR DEGREASING-OPEN TOP-METHYLETHTYLKETONE (MEK)
SOLVENT-VAPOR DEGREASING-OPEN TOP-CHLOROSOLVE
SOLVENT-VAPOR DEGREASING-OPEN TOP-CHLOROSOLVE
SOLVENT-COLD CLEANING-STODDARD SOLVENT
                                                                                                                                                 ORGANIC SOLVENT-SURFACE COATING-PAINT-POLYVINYL CHLORIDE
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-HEXANE
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-METHYLENE CHLORIDE
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-METHYLENE CHLORIDE
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-LACQUER THINNER
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-HEXYLENE GLYCOL
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-HEYLENE GLYCOL
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-HEPTANE
ORGANIC SOLVENT-SURFACE COATING-THINNING SOLVENT-ETHYLENE OXIDE
ORGANIC SOLVENT-SURFACE COATING-RESINS-FIBERGLASS RESIN
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               GAS PRODN-FITTINGS-LIQUID SERVICE
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TANKS-BREATHING LOSS-GAS FUEL
TANKS-BREATHING LOSS-ETHYLENE DICHLORIDE
  PRODN-FITTINGS-GAS SERVICE
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TANKS-BREATHING LOSS- SEC-BUTYL ALCOHOL
TANKS-BREATHING LOSS- METHANOL
                                                                                                                                                                                                                                                                                                                                                            TANKS-BREATHING LOSS-RECOVERED OIL
TANKS-BREATHING LOSS-FUEL OIL
TANKS-BREATHING LOSS-LIGHT NAPHTHA
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LOSS-ASPHALT
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Profile Code

SCC

Description of SCC

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STORAGE-FIXED ROOF TANKS-WORKING LOSS-ETHYLENE DICHLORIDE STORAGE-FIXED ROOF TANKS-WORKING LOSS-BTX STORAGE-FIXED ROOF TANKS-WORKING LOSS-CARBON TETRACHLORIDE STORAGE-FIXED ROOF TANKS-WORKING LOSS-ORTHO-XYLENE STORAGE-FIXED ROOF TANKS-WORKING LOSS-ASPHALT STORAGE-FIXED ROOF TANKS-WORKING LOSS-NAPHTHA STORAGE-FIXED ROOF TANKS-WORKING LOSS-NAPHTHA STORAGE-FIXED ROOF TANKS-WORKING LOSS-NAPHTHA STORAGE-FIXED ROOF TANKS-WORKING LOSS-CALCHLOROFTHYLENE	NG ROOF TANKS-WORKING LOSS-METHANDL NG ROOF TANKS-STANDING LOSS-RECOVER NG ROOF TANKS-STANDING LOSS-FUEL OI NG ROOF TANKS-STANDING LOSS-LIGHT N NG ROOF TANKS-STANDING LOSS-LIGHT N NG ROOF TANKS-STANDING LOSS-AVGAS. NG ROOF TANKS-WORKING LOSS-JET FUEL NG ROOF TANKS-WORKING LOSS-DISTILLA NG ROOF TANKS-WORKING LOSS-BISTILLA NG ROOF TANKS-WORKING LOSS-FUEL OIL	AGE-FLOATING ROOF TANKS-WORKING LOSS-TUL AGE-FLOATING ROOF TANKS-WORKING LOSS-GA: AGE-FLOATING ROOF TANKS-WORKING LOSS-GA: AGE-FLOATING ROOF TANKS-WORKING LOSS-AV AGE-TOATING ROOF TANKS-WORKING LOSS-NC AGE-TANK CLEANNOT CLASSIFIED TERMINALS-STORAGE-FIXED ROOF-ETHYL ACE: TERMINALS-STORAGE-FIXED ROOF-ISOBUTYL / TERMINALS-STORAGE-FIXED ROOF-ISOBUTYL / TERMINALS-STORAGE-FIXED ROOF-ISOBUTYL /	BULK TERMINALS-STORAGE-FIXED ROOF-ISOPROPYL ACETATE BULK TERMINALS-STORAGE-FIXED ROOF-METHYL AMYL KETONE BULK TERMINALS-STORAGE-FIXED ROOF-METHYL ISOBUTYL KETONE BULK TERMINALS-STORAGE-FIXED ROOF-N-BUTYL ACETATE BULK TERMINALS-STORAGE-FIXED ROOF-N-BUTYL ACETATE BULK TERMINALS-STORAGE-FIXED ROOF-N-BUTYL ACETATE BULK TERMINALS-STORAGE-FIXED ROOF-N-PROPYL ACCHOL BULK TERMINALS-STORAGE TANK-FIXED ROOF-LUBE OILS BULK TERMINAL-STORAGE TANK-FIXED ROOF-LUBE OILS BULK TERMINALS-MISCELLANEOUS-SUMPS AND PITS
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Profile Code

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Description of SCC

BULK PLANTS-STORAGE-FIXED ROOF-KEROSENE BULK PLANTS-STORAGE-FIXED ROOF-GASOLINE	BULK PLANTS-STORAGE-FIXED ROOF-LUBE OILS RIIK PLANTS-MISCFILANFOLIS-SIMPS AND PITS	BULK PLANTS-MISCELLANEOUS-MECHANICAL OIL/WATER SEPARATORS	BULK PLANIS-STORAGE-FLOATING ROOF- GASOLINE OTE AND GAS PRODUCTION-CRUDE OTE STORAGE-UNSPECIFIED	PETROLEUM MARKETING-TANK CARS/TRUCKS-ETHYLENE DICHLORIDE-SUBMERGED LOADING	PETROLEUM MARKETING-MARINE VESSELS-FUEL OIL-LOADING	PETROLEUM MARKETING-MARINE VESSELS-BTX-LOADING	CHEMICAL MARKETING-MISCELLANEOUS-VALVES, FLANGES, PUMPS	CHEMICAL MARKETING-TANK CARS/TRUCKS-SUBMERGED LOADING-PERCHLOROETHYLENE	CHEMICAL MARKETING-TANK CARS/TRUCKS-SUBMERGED LOADING-1,1,1-TRICHLOROETHANE
761 710	768	33.	71B	78	768	762	600	85	87
48488234 48488235	40400236 40400261	48488262	4 <i>B</i> 4 <i>BB</i> 285 4 <i>B</i> 4 <i>BB</i> 399	40600111	40600206	40600207	40700151	40700201	40700202

## Assignment of New Profiles to Existing SCCs

This study resulted in many new TOG speciation profiles. To fully benefit from this improved data set, the original assignment of species profiles to existing SCCs was reviewed to determine if more appropriate profiles were available. As a result, the existing SCCs and CESs shown in Table 6-2 were assigned new species profiles.

## Revision of the MED Inventory

Several procedures were followed in making the actual revisions to the MED files. An important consideration was the conversion to kilograms per hour (kg/hr) of the emission rates presented in tons per year (tons/yr) in the documents developed for facilities receiving inventory changes. The equation we used followed the ARB procedure for converting point source EDS data (tons/yr) to the MED format (kg/hr) for a summer weekday (Johns, 1983).

$$kg/hr = tons/yr \frac{(fraction for July) (fraction for Thursday) (907.1848 kg/ton)}{(hours/day)}$$

where

Fraction for July = fraction of emissions occurring in July;

Fraction for Thursday = 1/number of days per week of operation, in most cases;

Hours per day = end hour - begin hour + 1, for source operations;

Weeks per month = 4.345

Information on seasonal operations for individual facilities was limited. Therefore, the fraction for July equaled 1/12 during the conversion process. The fraction for Thursday used the ARB convention as follows:

TABLE 6-2. Existing SCCs and CESs assigned new species profiles.

New Profile

01d Profile

SCC

Description of SCC

TABLE 6-2 (continued)

New Profile

01d Profile

SCC Code

Description of SCC

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NT- COATING- COMPOSITE  NT- COATING- PRIMER- GENERAL  NT- COATING- PRIMER- GENERAL  NT- COATING- COMPOSITE  NT- COATING- ADHESIVE- GENERAL  NT- COATING- ADHESIVE- GENERAL  NT- COATING- OVEN- UNSPECIFIED  NT- COATING- OVEN- UNSPECIFIED  NT- COATING- OVEN- VITS F- UNSPECIFIED  NT- COATING- OVEN- UNSPECIFIED  NT- SRFC COATING- TANK- BREAT  GASOLINE- FIXED ROOF TANK- WORKING- OF TANKS-GAS-RVP 13-550K BF- TIXED ROOF TNKS-GAS-RVP 13-550K BF- TIXED ROOF TNKS-CAS-RVP 13-550K BF- TIXED ROOF TNKS-CAS-RVP 13-550K BF- TIXED ROOF TNKS-CAS-	ETRO STORAGE - DIST #Z- FIXED KOOF- ETRO STORAGE - GASOLINE STORAGE ETRO STORAGE - GASOLINE STORAGE ETRO STORAGE - FLOAT ROOF TNKS-GAS-RV ETRO STORAGE - FLOAT ROOF TNKS-GAS-RV ETRO STORAGE - GASOLINE RVPIØ- FLOAT ETRO STORAGE - FLOAT ROOF TNKS-GAS-RV ETRO STORAGE - FLOAT ROOF TNKS-GAS-RV ETRO STORAGE - DIST #Z- FLOATING - 67 ETRO STORAGE - DIST #Z- FLOATING - 67 ETRO STORAGE - DIST #Z- FLOATING - 25 ETRO STORAGE - FLOAT ROOF TNKS-DIST.O
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030114	57.4	76.8	ETRO STORAGE-FLOAT ROOF TNKS-EXTNL-SCDRY SEAL-DI
Ø3Ø11E	U1	718	ETRO STORAGE-FLOAT ROOF TNKS-INTERN ROOF-GAS
838115	514	760	ETRO STORAGE-FLOAT ROOF TNKS-INTERN ROOF-DIST. OIL
Ø3Ø12&	98	71.0	O STORAGE - GASOLINE RVP13 - VAR VAPOR SPACE - FILLIN
Ø3Ø12£	98	718	O STORAGE-VAR.VAPOR SP.TNKS-GAS-RVP 109-FILLING LOS
Ø3Ø12k	U	71.0	O STORAGE-VAR.VAPOR S
Ø3Ø12&	514	76.8	D STORAGE-VAR.VAPOR SP.TNKS-DIST FUEL #4-FILLING LOSS
848818	86	71.8	TERMINALS-FIXED ROOF TNKS-GAS-RVP 13-67K BBL-BREAT
848818	86	718	TERMINALS- GASOLINE RVP1Ø- FIXED ROOF- 67K BBL BR
848018	98	718	TERMINALS-FIXED ROOF TNKS-GAS-RVP 7-67K BBL-BREAT
040012	86	718	TERMINALS-FIXED ROOF
040010	98	718	TERMINALS-FIXED ROOF TNKS-GAS-RVP 10-250K BBL-BREATHN
040016	98	710	TERMINALS-FIXED ROOF TNKS-GAS-RVP 7-250K BBL-BREATHN
648012	86	718	TERMINALS-FIXED ROOF TNKS-GAS-RVP 13-WORKING LOSS
848818	98	710	TERMINALS- GASOLINE RVP10- FIXED ROOF- WORK
040010	86	71.0	TERMINALS-FIXED ROOF
040011	86	718	TERMINALS-FLOAT ROOF TNKS-GAS-RVP 13-67K BBI
040011	86	71.0	TERMINALS - GASOLINE RVP10 - FLOATING - 67K BB! STANDING TO
040011	86	718	TERMINALS GASOLINE RVP7 - FLOATING A7K RRI STANDING LOSS
040011	96	710	TERMINALS-FLOAT ROOF TNKS-CAS-RVP 13-258K RRL-STAND IS
G A G G T 1	α	7 - 2	TERMINALS CASOLINE DVD18 - CLOATING STANDS
949911	α	718	TERMINAL CHELOAT BOOK TAKE CASTANG COME DEL STANDING COS
040011	o 0	2	TEDMINAL CACOLINE DVINA 403 NVI - 1.600N DIE - 3.7NV I
040011	o 0	718	DOLY TERMINALS 4550CHINE NOTICE FLOATING 550C DOL WITHDRAWAL
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860013	<b>←</b>	760	MARKETING- DIST OIL - TANK TRICKS- SHRWERGED
50013	D,	710	MARKETING- GASOLINE- TANK TRUCKS- SPLASH LOAF
80014	514	768	MARKETING-TNK CARS/TRKS-DIST 011-SPI ASH-1
0014	g	71.0	MARKETING-TNK CARS/TRKS-GAS-SBMRG-10AD-RAIANCF SVC
00014	86	71.0	MARKETING-TNK CARS/TRKS-GAS-SPIASH-I DAD-BAI ANCE
3014	86	710	RO MARKETING-TNK CARS/TRKS-GASOLINE-SBMRG-LOAD-CLEAN
0015	86	71.0	ETRO MARKETING- GASOLINE- TANK TRICKS- LINI DADING
80015	-	768	ETRO MARKETING-TNK CARS/TRKS-DIST DII - HNI DAD
00016	51.4	76.0	FIRD MARKETING-THK CARA/TRKS-DIST OIL-SBM6G-LOAD-CLEA
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	New Profile
(continued)	01d Prof11e
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Description of SCC

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TABLE 6-2 (concluded)

01d Profile

SCC

Description of SCC

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	COATING—GLASS PRUDUCIS COATING—TRANSPORTATION EQUIPMENT #2 COATING—TRANSPORTATION EQUIPMENT #3 COATING—LUMBER & WOOD PRODUCTS COATING—MACHINERY CONSTRUCTION COATING—FABRICATED STRUCTURAL METAL COATING—PAINT MANUFACTURING COATING—PAINT MANUFACTURING COATING—NON FERROUS METALS COATING—TRANSPORTATION EQUIPMENT—MISSLES COATING—CHEMICAL & ALLIED COATING—RUBBER & PLASTICS FABRICATION COATING—MACHINERY—ELECTRICAL COMPONENTS	EVAPORATION- TANK EVAPORATION- TANK EVAPORATION- TANK EVAPORATION AT SER EVAPORATION AT SER EVAPORATION AT SER EVAPORATION- TANK EVAPORATION- TANKE	ORATION- MANUFACTURING ORATION- WATER BASED ORATION- SOLVENT CK ASPHALT EVAPORATION OIL EVAPORATION NATURAL GAS NERATION) APPLICATION LOSS - CHE EVAPORATION LOSS - CHE
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1931 2010 2010 2010 2010 2010 2010 2010 20	AAA25916 AAA27854 AAA27928 AAA21995 AAA31583 AAA31963 AAA32342 AAA35212 AAA35212 AAA37861 AAA39824	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	4674 4674 4677 4687 7468 7468 9888 9888

Days per Week	Fraction for Thursday
1	0.20000
2	0.00000*
3	0.20000
4	0.20000
5	0.20000
6	0.16667
7	0.14286

To determine the number of hours per day of point source operations, we used the beginning and ending hours of equipment activity; we determined these beginning and ending hours from the daily hours of equipment operation, again using the ARB procedure.

Daily Hours	Begin	End
of Operation	Hour	<u>Hour</u>
1	8	15
2	8	15
3	8	15
4	8	15
5	8	15
6	9	14
7	9	15
8	8	15
9	8	16
10	7	16
11	7	17
12	6	17
13	8	20
14	8	21
15	8	22
16	8	23
17	7	23
18	6	23
19	3	21
20	3	22

<sup>\*</sup> This value is assigned by ARB by convention; the 2 days per week are assumed to be a weekend.

<sup>84052(</sup>a)r 10

21	2	22
22	1	22
23	1	23
24	0	23

To make the conversion from tons/yr to kg/hr, therefore, it was necessary to estimate the number of days/week and hours/day of operation for each source to be added to the inventory. Because we had obtained summaries of the TADs for many surveyed facilities, we used the days/week and hours/day information given in the TADs for devices at a facility. In cases for which we did not have facility TADs, operating schedules were estimated from a knowledge of typical source operations and from survey responses.

Several other procedures were followed in revising the inventory:

- (1) If emissions other than TOG or  $NO_X$  were listed in the inventory for a facility device that was to be deleted from the file, all emissions for the device were deleted.
- (2) If some but not all devices with the same SCC at a facility were to be deleted, then the ratio of the individual emissions was used to determine the amount of emissions to retain.
- (3) When emissions from multiple devices with the same SCC were added for a facility, the emissions were summed before being entered into the file.
- (4) If an SCC was added and the same SCC already existed and was retained for a facility in the file, then the new emission rate for the SCC was entered separately from the emission rate for the existing SCC.
- (5) New emission rates were rounded to the nearest tenth of a kg/hr unless the value was less than 0.05 kg/hr, in which case we rounded the value to one significant digit and entered the new emission rate.

One specific change to emissions in the area source file was made in addition to the reassignment of speciation profiles. As discussed in Section 4 and Appendix B, revisions were made to the inventory for emissions from internal combustion engines that affected both the point and area source MED files. To change the area source file, the ratio of the total area source emissions for CES 66787 (internal combustion engines, natural gas) in the original file to the revised emissions by pollutant was used to derive multipliers for revising the inventory. The following factors were used for every grid cell with CES 66787 emissions in the area source file:

TOG	NOX	CO	$\frac{S0_{X}}{}$
0.4633	0.3582	0.3903	1.0

Before entering the revisions to the inventory, we first produced a new set of the original point, area, and speciation files on the computing system. Changes were then made to these MED files, and the new files were transmitted to the ARB. No changes were made to the stack data file.

### RESULTS OBTAINED FROM INVENTORY REVISIONS

Results obtained using the "old inventory category" (OIC) system (among others) for grouping sources are presented in this section. It was necessary to generate new OIC codes for combinations of SIC/SCC pairs because (1) new SCCs and new SIC/SCC pairs were developed in this study, and (2) many existing SIC/SCC pairs in the original inventory were not assigned OIC codes in our November 1982 version of the file "category." The OIC codes we assigned to SIC/SCC pairs for reporting purposes are listed in Appendix G.

In addition, two unusual SCCs in the original point source file appeared to be in error. The SCCs occurred at two facilities, Nos. 6550 and 7182, in Los Angeles County. We revised these two SCCs to what appeared to be their correct codes. There were also four SCCs that were not assigned speciation profiles in the original MED files: 4-02-002-07, 4-02-004-11, 4-02-005-11, and 4-09-009-22. A miscellaneous profile was used to speciate emissions for these SCCs.

Appendix H contains a listing of all facilities in the point source file that underwent revisions during the study. The revised emission rates for devices at these facilities were developed from several sources of information including survey responses and SCAOMD emission fee data. emission changes affected facilities in all four counties of the SOCAB and included additions, deletions, and modifications, primarily for TOG and  $NO_x$ ; revisions to  $SO_x$ , CO, and PM were also made for combustion equipment at some facilities. All changes were initially generated on a tons/yr basis and then converted to kg/hr for the MED inventory as described. In addition, minor revisions to the MED files for surveyed facilties were also occasionally incorporated (e.g., the revision of an SIC code). All changes to the MED inventory were documented in a set of handwritten pages for each facility (e.g., SCC entries, tons/yr and kg/hr emission rates, beginning and ending hours, justification for the revisions) for subsequent use by the ARB and SCAQMD in updating their data bases. This documentation has been transmitted to the ARB.

A substantial body of data is contained in the documentation of study recommendations for changes to the inventory. Changes in emission rates were made to 241 facilities. We made an average of about seven changes per facility, so about 1500 MED entries (or computer records) were revised during the project. In addition to these changes, modifications to the MED inventory were incorporated for speciation profiles. Most of the emission changes to the MED point source file were on the order of 0.2 to 5.0 kg/hr, but changes as large as 150 kg/hr were also made. Eight cases of double counting of facilities were also uncovered. Double-counted facilities were deleted from the MED file, as appropriate. This topic is discussed later in this section.

## Emission Data and Classes of Organic Gases

The overall effects of these revisions on inventory totals are demonstrated through a comparison of tables containing categories of emissions for the original and revised inventories. Tables 6-3 and 6-4 present emission rates using the OIC system of categorizing sources for the original and revised inventories, respectively. Emissions from the point and area source files are summarized in these tables, but the totals do not represent the complete SOCAB inventory because on-road motor vehicle emissions are excluded. In all of these tables, SOCAB emissions are given in tons/day for a 1979 summer weekday. Emissions of TOG and ROG are expressed as methane and NO $_{\rm X}$  emissions are expressed as nitrogen dioxide.

Although changes to the spatial resolution (e.g., from oil field revisions) and temporal resolution (e.g., from the modified diurnal profile for power plants) of the inventory cannot be ascertained from such tabulations, Tables 6-3 and 6-4 do show that overall TOG and ROG emissions have increased by 6.2 and 7.5 tons/day (0.2 and 1.0 percent), but that NO $_{\rm X}$  emissions have decreased by 15.6 tons/day (3.0 percent). Total CO and SO $_{\rm X}$  emissions have decreased a fraction. We note that some particulate matter emissions not shown in the tables were added to the inventory for petroleum refining activities.

A comparison of source categories presented in these tables reveals some interesting trends. Important changes occur within the petroleum industry for organic gases. Emissions of TOG from petroleum production (OIC 120 plus 410) increase, whereas ROG emissions decrease somewhat. Both petroleum refining (OIC 130 plus 420) TOG and ROG emissions decrease, as do petroleum marketing (OIC 430) TOG and ROG emissions. However, the reactivity of petroleum production emissions has been reduced significantly as measured by the ROG/TOG ratio for OIC 410. The same ratio for petroleum refining (OIC 420) has hardly changed.

TABLE 6-3. Original emission data by OIC code. (Tons per dav)

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	XOS	2	B.OU	Ø.18	Ψ.	16.84	<b>.</b> , .	1.64	0.00 0.00	0 0 0 0 0 0 0 0 0	2 S S S S S S S S S S S S S S S S S S S	8.00 00.00	B. BB	3.39	Ø.00	Ø.80	ନ୍ଦ ଜନ୍ମ ଜନ୍ମ	9 . S. S	0 . X	20.00	B. BB	B. WU	Ø. ØØ	Ø9.00	Ø. BB	22 r	56.87	80. 80. 80.	88.8 88.8	2.16	ø. Ø2	3.18	1.18.928 22.22	3 . S	8 . 8 . 8 . 8	%3.°% Ø.∪ú	B. 153	Ø. 640	ப்ப். வ	<b>66.</b> €	18. SU	Ø. 110	25	ছ :	20 50 20 50	10 10 10 10 10 10 10 10 10 10 10 10 10 1	3/4.5
	NON	И. П.Й	Ø.03		59.35	9					8 6 6 8	8.68 8.68	W. 00.0	0.42	B. B.J	83.88 83.88	0.00.00 0.1.00	۵.1.3 م جرو	88.8 88.8	8.88	Ø.13	80.03	8.82	Ø . Ø Ø	g. gg	. 9.11	13.30	8 . 8 . 8 . 8 .	8.80 8.80	0.50	Ø.35	4.27	5.54	, Q. Q.	8.5.6	99.8	Ω. W.	18 18 18 18	B. 1815	Ø , Ø 6	18.7.	ا ، از ، کار ا	6.67			18.18.0 18.18.0	
	00	B. B.	B. B.B	1.87	ç,	/1.13	4.	70.7	10.73	7 2	37.33	8.80 8.80	B.BB	Ø.43	0.34	88.88	000.00 000.00	8 . 80 8 . 80 8 . 80	88.38	95.00 00.00	0.01	B. B3	Ø.Øğ	Ø.00	Ø.00	88.88 80.18	13.82	8 . 8 8 8 8 8	3.35 0.88	2.57	Ø.25	ۍ ا	104.61 6 86	6. M1	. 8 8 8	00.00	BO.B		Ø	Ø. Ø	499.94	00.00 €	ان د ان	हात. इ.स. इ	00.00 00.00 00.00	ત કુલ. છે	
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(Tons per day)	T0G	\$ . 9	B. BB	3.79	11.05	7.35	17.83	0 / · c	15.78	2 (	± 50 € 50 € 50 € 50 € 50 € 50 € 50 € 50	ž	Ø.00	Ø.16	8.80 20.50	<b>z</b> -	٠,	40.00 10.88	,	N	Ň	ē	ຕ໌,		<b>S</b> .	س د	, 0 0	7 .	: :D	13.68	ب	1.1.	₹ . C	0 + - ∀ + ∀	1. U.B.	ъ.	•	•	¥.94	æ	00000	: c	: (بد	•		13. 15. 15. 15. 15. 15. 15. 15. 15. 15. 15	
	DE SOURCE NAME	FUE	Ø AGRICULTURAL		PEIROLEUM KEFINING Other Manifacturiation	90740-04 CONT. 232-0		RESIDENTIAL		WAS		RANGE MANAGEMENT			A THE CO IN THE COURSE OF THE									DETROLEIM DROCESS STOBASE * TBANSET	OIL AND CAR EVIDACTION				INDUS		M A A A	METAL			MISC		FARMING OPERATIONS	CONSTRUCTION AND DEMOLITIO	ENTRAINED ROAD DUST - PAVED	- IMPLANKED KOAD DOSI - UNFA			ON NO	LIGHT	LIGHT	HEAVY DUTY GAS TRUCKS	
	CODE	100	118	777	1 4 8	110	168	178	199	288	218	220	230	240	200	318	320	33.0	340	350	36 <i>g</i>	36.36	388	2 C	418	420	430	499	5 Ø Ø	51 <i>B</i>	10 C B	578 578	580	599	6.00 W	610	620	0 2 0 0 4 0	5 4 7 5 7 7 5	200	000 000 000	669	788	718	720	730	

TABLE 6-3 (concluded)

CODE	SOURCE NAME	7.0G	ROG	00	XON	XOS
7448 858 818 818 838 858 878 888	HEAVY DUTY DIESEL TRUCKS MOTORCYCLES OTHER MOBILE OFF ROAD VEHICLES TRAINS SHIPS AIRCRAFT - GOVERNMENT AIRCRAFT - OTHER MOBILE EQUIPMENT UNSPECIFIED SOURCES	M. BB B. BB B. BB B. BB 31. B4 6. B3 1. 78 B. 17 19. 56 24. B7 12. 42 B. DB	8.98 8.98 8.98 26.87 5.98 1.63 8.17 18.94 19.56 19.56	8.88 8.88 8.88 183.78 183.74 3.51 1.53 83.79 168.87 118.97	8.58 8.58 8.52 8.52 21.52 12.78 8.49 11.99 73.27 73.27	8.88 8.82 8.82 8.87 2.14 17.37 1.12 1.12 8.85 8.98
TOTAL		2962.42	771.98	1278.27	528.9 <i>g</i>	217.61

emission data by OIC code.	s per day)
Revised	(Tons
TABLE 6-4.	

TOG ROG CO NOX	B   198   B	27
	10 N N C C C C C C C C C C C C C C C C C	INTRAINED ROAD DUST - PAYED ENTRAINED ROAD DUST - UNPAVED UNPLAHNED FIRES SOLID WASTE LANDFILL OTHER ROAD VEHICLES

# TABLE 6-4 (concluded)

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3000	SOURCE NAME	106	806	60	NON	X03
74.9	HEAVY DUTY DIESEL TRUCKS	18. Z	18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	9.93	10 · · · · · · · · · · · · · · · · · · ·	: :
387	HUTORCYCLES	88° 0	13.67 16	15.98		
3 <i>0</i> 22	OTHER MOBILE	#3.3	19. W. 19.	86.88	F	73
31.9	OFF ROAD VEHICLES	31.34	25.87	1.00.7%	: :00	(B)
329	TRAINS	5,53	86.8	181.00	21.3	
33.0	SHIPS	1.73	1.63	3.5)	12.5	17.37
350	AIRCRAFT - GOVERNMENT	11.17	71.8	1.53	· · · · · · · · · · · · · · · · · · ·	1.4.16
` ₿98	AIRCRAFT - OTHER	19.58	18.94	23.79	0.11	5 t
37.3	MOBILE EQUIPMENT	24.37	03.50	1.68.83	73.57	1 40 27 2 47 2 47
38.0	UTILITY EQSIPMENT	12.42	18.39	16.11.	्र •	34
3,00.5	UNSPECIFIED SOURCES	B.B.B.	, ME. 6	11. UZ	14	W
OTAL		2968.62	779.45	1275.28	513,25	215,8%

Furthermore, some surface coating (OIC 340) TOG and ROG emissions have increased, as have industrial solvent use (OIC 380) TOG and ROG emissions. The ROG/TOG ratio has hardly changed for miscellaneous fuel combustion (OIC 199); both TOG and ROG emissions have decreased for this category. Moreover, the asphalt paving (OIC 350) ROG/TOG ratio has increased substantially due to revised species profiles. We conclude from this comparison that important changes to organic gas reactivity in the inventory have occurred as a result of this study.

In the case of  ${\rm NO_X}$  emissions, the major changes occur for the petroleum refining (OIC 130) and miscellaneous (OIC 199) fuel combustion groups— ${\rm NO_X}$  emissions for refining have increased and those for unspecified sources have decreased. These  ${\rm NO_X}$  changes are the result of specific identification of internal combustion engines as part of the petroleum refining sector.

Another means of reviewing the results in the original and revised inventories is provided in Tables 6-5 and 6-6, which show emissions for each speciation profile. All profiles numbered in the 700 series were developed in this study, so Table 6-6 is longer than Table 6-5. The TOG and ROG emissions for profile 600, a category used for those sources without an appropriate speciation profile, have been reduced in the revised inventory as a result of the reassignment of new profiles appropriate to several of those source types. Other profile-to-profile comparisons are less meaningful. For example, TOG and ROG emissions for profile 96 are signficantly reduced, but a large percentage of the surface coating solvent emissions in the revised inventory have been assigned new species profiles in the 700 series. In addition, overall changes in NO $_{\rm X}$  emissions from internal combustion engines are demonstrated in these tables by comparing the original NO $_{\rm X}$  emissions for profiles 7 and 10 in Table 6-5 with the revised NO $_{\rm X}$  emissions for profile 719 in Table 6-6.

Tables 6-7 and 6-8 provide information on the reactivity of organic gas emissions for source categories in the original and revised files. Individual TOG species were placed into six carbon-bond classes to generate these two tables:

Olefins (OLE)
Paraffins (PAR)
Aromatics (ARO)
Carbonyls (CARB)
Ethylene (ETH)
Unreactive

TABLE 6-5. Uriginal emission data by profile code.

E

TOTAL PROFILE NAME	XOX XON	3.77 3.19 3.24 3.24 3.25 3.25 3.25 3.25 3.25 3.25 3.25 3.25
EXTERNAL COMBUSTION BOILER- NATURAL GAS  BOILERS- CREE OVER GAS  BOILERS- CREE	00	11.3 12.3.7.7.2 12.3.7.7.2 13.3.7.7.2 13.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.
EXTERNAL COMBUSTION BOILER - NATURAL GAS BOILERS - REFINENY GAS BOILERS - COKE OWEN GAS INDUSTRIAL ICE - DISTILLIATE OIL INDUSTRIAL ICE - DISTILLIATE OIL INDUSTRIAL ICE - MAT. GAS - RECIPROCATING GAS PHALT CONCRETE - ROT. DRYERS - NATURAL GAS ASPHALT CONCRETE - ROT. DRYERS - NATURAL GAS ASPHALT CONCRETE - ROT. DRYERS - NATURAL GAS ASPHALT CONCRETE - ROT. DRYERS - REFINE REFINENY COOLING TOWERS FUGITIVE EMISSIO REFINENY COOLING TOWERS FUGITIVE EMISSIO REFINENY COOLING TOWERS FUGITIVE EMISSIO REFINENY COOLING TOWERS FUGITIVE GAS ASPHALT CONCRETE - ROT. DRYERS - LPG REFINENY COOLING TOWERS FUGITIVE GAS ASPHALT CONCRETE - ROT. DRYERS - LPG REFINENY COOLING TOWERS FUGITIVE GAS ASPHALT CONCRETE - ROT. DRYERS - LPG REFINENY COOLING TOWERS FUGITIVE GAS REFINENY COOLING TOWERS FUGITIVE GAS REFINENY COOLING TOWENS FUGITIVE GAS REFINENY COOLING TOWEN GAS REFINENY COOLING TOWEN GAS REFINENT IN COORNOST E COMPOSITE CONTING SOLVENT BETWEEN COATING EVAP - GENERAL PRIMER SURFACE COATING EVAP - GENERAL COMPOSITE SURFACE COATING EVAP - GENERAL PRIMER SURFACE COATING EVAP - CHERRAL	day)	$oxed{ egin{array}{cccccccccccccccccccccccccccccccccccc$
EXTERNAL COMBUSTION BOILER- NATURAL GAS BOILERS- COKE OVEN GAS INTERNAL COMB. ENG TURBINE- NATURAL INDUSTRIAL ICE- DISTILLATE OIL SASPHALT ROOFING- DIPPING ASPHALT ROOFING- DIPPING ASPHALT CONCRETE- ROT. DRYER- NAT. GAS ASPHALT CONCRETE- ROT. DRYER- NAT. GAS ASPHALT CONCRETE- ROT. DRYER- NAT. GAS ASPHALT CONCRETE- ROT. DRYER- DRA REFINERY COBILER- FCC REFINERY COBILER- FCC REFINERY CATALYTIC REFORMER- GENERAL F REFINERY CATALYTIC REFORMER- GENERAL GAS REFINERY CATALYTIC REFORMER- GENERAL F RETINERY CATALYTIC REFORMER- GENERAL F PERRO MARKETING- RELIEF VALVES- LPG REFINERY CATALYTIC REFORMER- GENERAL F PERRO DICHLORIDE MANUFACTURING FLARES- COMPOSITE WORKING & BREATI STODDARD CLEANING SOLVENT STODDARD CLEANING SOLVENT DEGRASING- TOLUENE SURFACE COATING EVAP- GENERAL COMPOSITE SURFACE COATING EVAP- GENERAL SOLVENT SURFACE	(Tons pe	04-wvn0vg2-gvg-gggqvgg4gg044gv0-gg4v40vgcg-gg-gg-gg
	ROFILE NAM	DN BOILER- NATURAL GAS GAS  4 GAS  5. TURBINE- NATURAL AT. GAS- RECIPROCATING AND PLACE ROAD ASPHALT COMERS FUGITIVE EMISSI OWERS FOR COVERED DRA TOWERS FOR SEALS- REFIN ATURAL GAS CLEFORMER- GENERAL F ATURAL GAS AND FACTURING AND FACTURING AND FACTURING AND FACTURING CLEANING SOLVENT E LUVENT- GENERAL AND SCAPE/PRUNING AP- GENERAL OSITE WORKING & BREATI COATING- AOPHESIVES AP- GENERAL VARNISH/SI AP- GENERAL PRIMER AP- GENERAL COMPOSITE COUCR- METAL FURNITURE COUCR- METAL SHEET AP- GENERAL COMPOSITE COMPOSITE OF ALCOHOL BASEL TO LUGHE COUCR- TOLUENE

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		ABLE 6-5 (concluded)	nged)			
CODE	PROFILE NAME	106	ROG	00	NON	SOS
518	AEROSOL SPRAYS- NON-SYNTHETIC	21.78	16.66	0.00	86.88	Ø. Br
519	AEROSOL SPRAYS- SYNTHETIC	1.53	W. WB	BB.BB	ŭ. N.S	10.00
52 <i>Ø</i>		295.76	5.46	Ø.Ø1	10.0	03.60
521	DIESEL EXHAUST(ALDEHYDES IN EMISSIONS)	17.91	15.58	31.10	94.51	15.19
522	FORMICA MANUFACTURING	1.26	88.8	W.OO	60.8	øø.øø
523	OPEN BURNING DUMP- LANDSCAPE/PRUNING (MO	22.32	9.69	119.51	83.8	00.00
524.	METHYLENE CHLORIDE	Ø.23	B. B.B	BB.BB	10. Wil	00.00
525	CATALYST EXHAUST (ALDEHYDES IN EMISSIONS	86.88	B.B.B	Ø.8.	18. St.	8.88
526	1979 EXHAUST COMPOSITE 50/50 (ALDEHYDES	3. BB	80.00	Ø.B.Ø	N. 20 S	B. B.
527	NON-CATALYST EXHAUST (ALDEHYDES IN EMISS	W. BB	B.B.B	00.00	B. B.B	B. S. B.
6 <i>BB</i>	SPECIES UNKMOWN- ALL CATEGORY COMPOSITE	42.28	29.92	56.96	32.58	6.97
6Ø6	MISCELLANEOUS	Ø.2Ø	0.14	Ø.OO	B.BB	B. B.
TOTAL		2962.42	771.99	1278.26	528.90	217.61

TABLE 6-6. Revised emission data by profile code.

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		(Tons per	day)		• )	
CODE	PROFILE NAME	7.04	ROG	00	XON	XOS .
f						
۱ <b>٠</b> -	NOTINGENOUS.	0.35	2.31	8.18	(1)	3.10
	SOILERS-KEFINERY GAS		3.818	7.23	44.60	30.00
		1.57	8.23	7.83	್ಯ ಕ್ಷ್ಮ	27.14
	INTERNAL COMBUSTION ENGINE-TURBINE-NATURAL GAS	3.85	13. S.	87.83	S S	1818
		51.47	8.73	1.85	00.00	
	INDUSTRIAL ICE-WATURAL GAS-RECIPROCATING		39. E	~		. :
	LALS	22.23	14.34	32.34	6.72	
	IRON SINTERING-PRIMARY METALS	W. 89	8.82	92.42	1.6	0.00
	ASPHALT ROOFING-BLOWING OPERATION	4.81	19.81	10.03	87.18	12. 13
		1.13	1.06	0.01	39.B	\$1.75 Va
	RATURAL	3.18	10.84	97.73	63.8	W.A.
	PSPRACH CONCRETETIN PLACE ROAD ASPRACT	85. J	66. 100. 100. 100. 100. 100. 100. 100. 1	11.93		
	/NIVan AU	7.55 2.05 2.05	50.00 50.00	11.21	3 2 2	83. 83. 83.
	7E EMISSIONS		5. 7. 7. 2. –	11 . S. S	7	2
	-REFINERY	•	5.47	10 m	18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	10. W.
	PETRO MARKETING-RELIEF VALVES-LPG	2 S	8.68	10. US		37.73
	SIMA ANITIONS IVES	9.06	3. S.	21. 21.		
	ICKAL FUGILIVE E	1. 50 7. – 10	8 9 9 1 1 1 1	27 E 2	0.7 %	: : : :
	PRINTING INK COUKING-GENERAL		1 1 1 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	33.33 3.33	े हैं इ. हे इ. हेंड	
	PESTICIDE USE-COMPOSITE DOMESTIC & COMMERCIAL	14.58	12.49	14 . XI.J		16. 35
20 /	ETHYLENE DICHLORIDE MANUFACTURING	5.47	10.07	11.00	(S) (S)	28.3
	PERCHIONOFINAL MANOFACTORING PERCHI	ا تا در	18. 18. 18. 18.	7. MS	in in	
	STODDARD CLEANING SOLVENT		2 to 1 to 2 to 2 to 2 to 2 to 2 to 2 to	1. 18 S	% (* ₹ 35	8 9 5 5 5 5
	1,1,1-TRICHLOROETHANE CLEANING SOLVENT	ښت.	: :3		जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिल्ला जिला जिला जिला जिला जिला जिला जिला जि	27.1.5
	DEGREASING-TOLUENE	3	13.5.2	11.93	1. J. 18	
	78 4 F	35. 35. 35.	27.15 21.5	18.33.7 13.7	\$6. \$.√. \$2.	
	SASOLINE EVAP-COMP WORK: & BREATH. LANK LUSS JET FOEL EVAPORATION	20. c.	86. 0 33. 0	ini Sisi Sisi	জ্ঞান জ্ঞান	351.31 351.31
	1.10		اند اند اند اند اند اند اند اند اند اند	32.32	•	2 (S) (S) (S) (S) (S) (S) (S) (S) (S) (S)
	IR DRIED	31.14.8	3.68	0.00		
	SIN	2.16	2.15	15. 15. 15. 15. 15. 15. 15. 15. 15. 15.	3	18 . A.S.
	1 N 1 N 1 N 1 N 1 N 1 N 1 N 1 N 1 N 1 N			19. 19. 19. 19.		70 (A) (A) (A)
	SURFACE COSTING EVAP-LABEL ADHESIVES		e '- '- '- '- '- '- '- '- '- '- '- '- '-	2 Z	:	8 (
	SURFACE COATING EVAP-METAL FURNITURE ADHESIVES	1. J. J. J.				1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
	URNITURE	¥.9%	13.24	19.00	75	K. N.
	D PROD. & CO	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	15. 15. 15. 15. 15. 15. 15. 15. 15. 15.	H.98	\$5,	3 : 1
	MPOSTE EMAMEL On cupwitner r		34 S	11. J.S.	in i	5
	OD FORKITORE L ENAMFI		ಶ್ರಕ್ಷ	2. 12 2. 22 2. 22 2. 23 2. 24 2. 24 24 24 24 24 24 24 24 24 24 24 24 24 2	`: `	8. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.
172	OHL BASED		1.32	99.88 10.888	(A : : : : : : :	
	악 볼 .		5,42	الايان الايان	1	25
	METHAME SOURCE	e te c t	5, 12 15, 12 5, 17	त्र १९ १९ १९	•	14 E
	ARCHITECTURAL SURFACE COATINGS-COMPOSITE SOLVENT	·	, Y.		£	
197	OSITE	27 20 20 30 30 40 40 40 40 40 40 40 40 40 40 40 40 40	1/20 04 : 0 04 : 0	13.803 13.003		6.83
⊋ .	DFILL EMISSIONS	1638.83	18, 16	Ø . Ø5	ŭ., ŭ	id . 1515

## TABLE 6-6 (continued)

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300	PROFILE NAME	706	R0G	00	XON	30%
	ROFILE NAME  EROSOL SFRAVS-SYNTHETIC  EROSOL SFRAVS-SYNTHETIC  OMPOSITE NATURAL GAS  IESEL EXHAUST (ALDEHYDES IN EMISSIONS)  ORMICA MANUFACTURING  PEN BURNING DUMP-LANDSCAPE/PRUN. (MOD. KY)  ETEYLENE CHLORIDE  ATALYST EXHAUST (ALDEHYDES IN EMISSIONS)  OYO-CATALYST EXHAUST (ALDEHYDES IN EMISSIC  ON-CATALYST EXHAUST (ALDEHYDES IN EMISSIC  OYO-CATALYST EXHAUST (ALDEHYDES IN EMISSIC  ASOLINE VAPORS-UNLEADED PREMIUM-SUMMER BIL  OYOUSTRIAL SURFACE COATING-COMPOSITE FOR INDUSTRIAL SURFACE COATING-COMPOSITE SOUNT CORE ASPHALT  OYOUSTRIAL SURFACE COATING-COMPOSITE SOUNT CORE ASPHALT  EDIUM CURE ASPHALT  OYOUSTRIAL SURFACE COATING-COMPOSITE SOUNT COMBUSTRIAL SURFACE COATING-COMPOSITE SOUNT CORE ASPHALT  OYOUSTRIAL SURFACE COATING-COMPOSITE SOUNT COURE ASPHALT  OYOUSTRIAL SURFACE COATING-COMPOSITE SOUNT COURD COUNT COMPOSITE SOUNT COURD COUNT CO	2801.2802.2833.333.333.333.333.333.333.333.333.3	105   10	2. 1. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2.		20
	LIQUID GASOLINE— LEADED REGULAR—WIRTER BLEND LIQUID GASOLINE—UNLEADED PREMIUM—WINTER SLEND LIQUID GASOLINE—UNLEADED PREMIUM—WINTER SLEND GASOLINE VAPORS—ULEADED PREMIUM—WINTER SLEND GASOLINE VAPORS—LEADED PREMIUM—WINTER SLEND LIQUID GASOLINE COMPOSITE OF PRODUCT—WINTER BLEND ASOLINE VAPORS—COMPOSITE OF PRODUCT—WINTER BLEND GASOLINE VAPORS—UNLEADED PREMIUM—WINTER SLEND GASOLINE VAPORS—UNLEAD REGULAR—SUMMER BLEND ASIT. GASOLINE VAPORS—UNLEAD REGULAR—SUMMER BLEND AGIT. GASOLINE VAPORS—UNLEAD REGULAR—SUMMER CHLOROSOLVE TRICHLOROTRIFLOUROETHANE OIL AND GAS PRODUCTION FUGITIVES—LIQUID SERVICE OIL AND GAS PRODUCTION FUGITIVES—UNSPECIFIED OIL & GAS PRODUCTION FUGITIVES—UNSPECIFIED		 	,在海道设置的"河南",河南西西西省省省, 海河南省省南部河南省省省省省省省省省省省省省省省省省省省省省省省省省省省省省省省	<b>克斯斯斯伊斯拉斯斯斯斯阿姆斯斯斯斯斯斯</b> 第二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十	a pa

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FROFILE NAME	CO NOX SOX	8	11 M 1 M 1 M 1 M 1 M 1 M 1 M 1 M 1 M 1	8. N. S.		Z Z.	18. CH	M. 7.2	18 18.		18 18.	755 	3.58 B. B. B.	B B.	100 m	18. S. S.	10	9.53	題 / 2 題	14.1.10	ज्ञा जा	26 3.6	角	32,	12.25 12.25	8 B	3. B3. B	W.233	3		1275.27 <b>513</b> .28 215.5%
EVAPORATIVE EMISSIONS-DISTILLATE FUEL EVAPORATIVE EMISSIONS-NAPITHA BTX (BENZEME/TOLUENE/XYLEME) PHTHALIC ANTYDRIDE MFG-XYLENE FLUOROCARBON-12/11 MANUFACTURING FLUOROCARBON-13/22 MANUFACTURING FLUOROCARBON-13/22 MANUFACTURING FLUOROCARBON-113/114 MANUFACTURING FLUOROCARBON-113 FLUOROCARBON-114 CHLOROCARBON-114 CHLOROCARBON-114 CHLOROCARBON-114 CHLOROCARBON-114 CHLOROCARBON-114 CHLOROCARBON-116 CHLOROCARBON-116 CHLOROCARBON-117 CACHTATE M-BUTYL ACETATE M-BUTYL ACETATE M-BUTYL ACETATE M-BUTYL ACCHTATE M-BUTYL ACCHTATE M-BUTYL ACCHTATE M-BUTYL ACCHTATE M-ROPYL ACC	RAG	4.25 8.66	Z.E.S	8.39	5.43	B.3. 16	BH. B	33.6	3.78	14.16	84.88	13. 18. 18.	£1.53	53.64	•	. •		8.86	£ . 14%	٠	•	Ø.98	5.07	77.15	٠	•	•	6.66	Ø.12		
EVAPORATIVE EMISSIONS-DISTILLATE EVAPORATIVE EMISSIONS-NAPATHA BTX (BENZENE/TOLUENE/XYLENE) PHTHALIC ANHYDRIDE MFG-XYLENE OPFLUOROCARBON-12/11 NANUFACTURING FLUOROCARBON-13/114 MANUFACTURING FLUOROCARBON-113/114 MANUFACTURING FLUOROCARBON-113/114 MANUFACTURING FLUOROCARBON-113/114 MANUFACTURING FLUOROCARBON-113/114 MANUFACTURING CARBON TETRACHLORIDE ORTHO-XYLENE ORTHO-XYLENE FLUOROCARBON MFG-VALVES, PUMPS, ISOBUTYL ALCOHOL. ISOBUTYL ACETATE NETHYL ISOBUTYL KETONE METHYL ISOBUTYL KETONE METHYL ISOBUTYL KETONE METHYL ACETATE M-PROPYL ACETATE M-PROPYL ACETATE M-PROPYL ACETATE M-PROPYL ACETATE M-PROPYL ACCOMOL HEXYLENE GLYCOL INDUSTRIAL SURFACE COATING-SYMTHETIC RUBBER MFG-STYRENE-BUTETHYL ALCOHOL CARBON BLACK MANUFACTURING MISCELLANEOUS	50.1	A. 3.7 5.82	1.1.11	5, 12	155	13.3	3.81	14,113	14.21g	4,119	19.514	11.113	11.119	86.93	11.19	5.219	19.81	H.85	<i>ii): '</i> [7	36.18	20.00	6.103	5.87	7::44	1.29	8.05	51.268	24.109	14.17	٠	2968.62
2000 2000																															

TABLE 6-7. Original classes of organic gases by OIC code. (Tons per day)

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CARB	######################################	
ARO	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
PAR	及 : 2 : 2 : 2 : 2 : 2 : 2 : 2 : 2 : 2 :	
s per day)	日	
SOURCE NAME	FUEL COMBUSTION  AGRICULTURAL OIL AND GAS PRODUCTION PETROLEUM REFINING OTHER NANUFACTURING INDUSTRIAL ELECTRIC UTILITIES OTHER SERVICES AND COMMERCE RESIDENTIAL OTHER SERVICES AND COMMERCE RESIDENTIAL OTHER SURVING AGRICULTURAL DEBRIS RANGEMENT FOREST HANAGEMENT OTHER SURFACE COATING ARCHITECTURAL COATING ARCHITECTURAL COATING OTHER SURFACE COATING ARCHITECTURAL COATING ARCHITECTURAL COATING ARCHITECTURAL SOLVENT USE OTHER SURFACE COATING ARCHITECTURAL SOLVENT USE OTHER PRINTING DOMESTIC INDUSTRIAL PAVING PETROLEUM REFINING OTHER FOOD AND AGRICULTURAL NINDUSTRIAL PROCESSES METAL PROCESSES METAL PROCESSES METAL PROCESSES METAL PROCESSES METAL PROCESSES MOD AND PAPER OTHER OTHER MISC PROCESSES METAL DROCESSES METAL NOB DEBULING DUST - UNPAVED SOLID WASTE LANDFILL ON ROAD VEHICLES LIGHT AND MEDIUM DUTY TRUCKS HEAVY DUTY GAS TRUCKS	
CODE	148	_

TABLE 6-7 (concluded)

ETH	8.68 8.68 8.67 8.74 8.28 8.16 1.47 8.62	42.64
CARB	8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	28.68
ARO	8.888 8.888 8.888 8.14 8.86 8.86 1.89 8.89	91.21
PAR	8.88 8.88 8.88 14.36 14.36 12.26 13.73 13.73 6.62	601.55
01.E	8.58 8.58 1.92 1.92 8.18 8.55 8.99 8.76	15.93
SOURCE NAME	HEAVY DUTY DIESEL TRUCKS MOTORCYCLES OTHER MOBILE OFF ROAD VEHICLES TRAINS SHIPS AIRCRAFT - GOVERNMENT AIRCRAFT - OTHER MOBILE EQUIPMENT UTILITY EQUIPMENT	-1
CODE	748 758 828 828 838 838 858 878 878	TOTAL

TABLE 6-8. Revised classes of organic gases by OIC code. (Tons ner dav)

		(Tons per	ır day)			
CODE	SOURCE NAME	01.E	PAR	ARO	CARB	H H
1 000	NOTE TO COMPLETE TO NOTE TO COMPLETE TO CO		ζ		3	:
1.15 1.15	ı	्राष्ट्र इ.स.स	19.19.18 6. ((al	हर - इ.स.स इ.स.स	9 5 5 5 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	એ જ જ એ જ એ જ
12.9	OIL AND GAS PRODUCTION		8.41	8.91	15 to	
136	PETROLEUM REFINING	•	2.62	0.83	18. E.F.	
140	OTHER MANUFACTURING/INDUSTRIAL		2.03	%.13	87.17	
ນ ປ ອີກ ອີກ	ς		6.43	¥.23	. j. j	3
9 전 기 다	SESTIDENTIAL	हें. इ. इ. इ. इ.	0.00 0.00 0.00	%. 20.2 0.2 0.2		75.8
561	OTHER		M 77	2 to 20	20 e 10 e	
20.9	WASTE BURNING		99.88		- 30 7 - 31 7 - 50 8 - 50	
21 <i>9</i>	AGRICULTURAL DEBRIS	30.0	8.01	0.83		16.161
9.77 7.72	RANGE MANAGEMENT		Ø.818	.W. O.W		Ŋ. 17.9
7 4 S	TORENT MANAGEMENT TACHNERATION	14. 15.15.	ි දින් වැනි	13. 13. 13. 888		8.55 8.55
293	OTHER		20 E	स. इ.स. इ.स.	70.0	8 8
388	SOLVENTUSE		88.5	ं कार्य (त. कार्य	_	24. E
31.0	DRY CLEANING		3.7%	3.50g		
323		30.100	24.99	66.99	10.11	1.44
33.6	ARCHITECTURAL COATING	1.22	75.67	3	2 55	
50 0 4 E 20 12 20 12	٠,	, 189 1	92.68		3.13	
000 000 000	DOTHAL PREING	9.78	23.83			
378 378	TAINING	20.0E	19.71	52.57	ر د د د	93.33 33.33
3.3.5 3.3.5	INDUSTRIAL SOLVENT USE	_	20.00 20.00 20.00		ກ ແ ກ – ຈ –	
399	UTHER		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		2	8. S.1
4.00.00	PETROLEUM PROCESS, STORAGE & TRANSFER		N. W.	13. 18.18		
<b>4.</b> 50.5	OIL AND GAS EXTRACTION	•	35.68	8.15	18. J.	6.63
# Z Z Z	PETROLEUM REFINING	•	49.7%	15.84	8.75	6.63
4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	FELKULEUM MAKKELING Otter	79. 79. 79. 79. 79.	74.13	2.28	N S	55.7
5,00	INDUSTRIAL PROCESSES	•	7./ 8 618	12.13 14.13.3	000 1000 2000 2000 2000	28.84 20.00 20.00
518	CHENICAL	8 C. T.	2. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	2.00 2.00 2.00	20.22 ∴ 12.2 ∴ 12.2	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
52 <i>B</i>	FOOD AND AGRICULTURAL	NE S	7.57	8.40	. ©: ∴ ∴ ⊙: ∷ ∴ ⊙:	12.50 12.50 12.50 13.50
2000 1	MINERAL PROCESSES		ы. 63	£ . 16 3	10.30	16.13
್ ಗ ಗ	METAL PROCESSES	•	1.29	رة. د د	18.0€ 10.0€	й. 72 
599	OTHER		35° 3	15. S	18. 3. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18	े दे व
6.83	MISC PROCESSES		1.4.1 61.0101	1 50 E	0 3 8 3	1.13.C
618	PESTICIDE APPLICATION	3. 3. 3. 3.	15.16	, (3 ; 10 ; 10 ; 10	B = B	6.88 8.83
629	ERATIONS		15.73	80.8	. 15.18	Ø.3.9
638	ON AND DEMOLITI	٠	13. 13. W	11. MU	27.4.6	6.2.9
040 010	ROAD DUST - PAVED	B.H. F.	33.3	0.00	Z = B	81.0. BI
5000 6000	KOMU DOMI - ONE	3(5° • )	99	5.93 5.93	151 to 15	;⊋;
389	WAS	2 J.	ママ・マー マコ・マー	13.50 10.00	18.00 18.00	Z (2 . 7 Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z
669	OTHER	18.30 18.30	. c.		 	) (A
7,0€ 21,0	HICLES	6,00	8319	11.31	16. x 58.	16.2.16
72 ×	DOLY PASSENGER			4.83	Ø	10
739	LIGHT AND MEDIUM DULY TRUCKS SEAVY DUTY GAS TRUCKS	7. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	ड ड हर्ड उज् <i>र</i>	74. 1833 1832	स्ट <sup>्र</sup> ्ट्स ट	ख हा हा हा हा हा
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CODE	SOURCE NAME	OLE	PAR	ARO	CARB	H L	
			-				
743	HEAVY DUTY DIESEL TRUCKS	H. S.	15.00 to 10.00 to 10.	P(B' 1)	20.00	94.8	
75.0	MOTORCYCLES	19 19 + Br	13.11	E3. E3		877.8	
888	OTHER MOBILE	19.00 B	M. 35.3	E.M. b.		8.55	
818	OFF ROAD VEHICLES	1.92	14.36	4.55	ું. •	3.67	
82.0	TRAIMS	17.18	4.62	6.14		19.74	
83,9	SHIPS	88.83	1.26	19.83	81.7. B	67.59	
850	AIRCRAFT - GOVERNMENT	15.51	$\beta$ . 1 1	10. W.	35. B.	M.30	
869	AIRCRAFT - OTHER	00.0	12.22	ان د د د		3.16	
87.0	MOBILE EQUIPMENT	6.76	13.73	2.19	11.10	2.11	
888	UTILITY EQUIPMENT	27.3	59.63	1.89	%. €.	1.67	
ું છે.	UNSPECIFIED SOURCES	N.33	M. 1515	G. 20	10.5%	10.38	
TOTAL		16.79	60%.56	182.99	17.76	41,36	

Emissions of the five reactive classes of organic gases are thus given in these tables in tons/day as methane for 1979. On-road motor vehicle emissions are excluded from these tables as well.

Emissions for two of the five classes of pollutants--OLE and ARO-increased, and emissions for three classes--PAR, CARB, and ETH--decreased in the revised inventory. One of the most important effects of the modifications to the inventory is an increase of 11.8 tons/day in the aromatic component of the inventory. This is chiefly the result of the improved speciation profiles for surface coating categories. The aromatic portion of emissions from OIC 340 (other surface coating) has increased more than 100 percent in the revised inventory. A major aromatic increase has also occurred for asphalt paving (OIC 350) emissions, but as discussed in Section 5, there is uncertainty associated with the new asphalt profiles. At the same time, substantial decreases in the aromatic portion of petroleum refining and marketing (OICs 420 and 430) emissions have resulted primarily from revised speciation profiles for qasoline. Similar conclusions regarding the aromatic component of the ROG inventory can be drawn from a comparison of Tables 6-9 and 6-10, which present classes of organic gases by profile code. Tables 6-7 through 6-10 generally indicate a more reactive mix of organic gases overall in the revised inventory.

## Organic Gas Species

Appendix E provides a listing of the 68 new speciation profiles, including individual organic gas species for each profile. One result of the use of the new profiles is the addition of many new species that were not a part of the original SOCAB inventory. Table 6-11 lists species that were added to the inventory together with their SAROAD numbers and molecular weights.

Tables 6-12 and 6-13 list TOG emissions in the MED point and area source files by reactive and unreactive species for the original and revised inventories, respectively. Individual organic gas species contained in the speciation profiles for each inventory are given in these tables on a weight percent basis. Two weight percents are listed: one column in the table lists the percent of the point and area source inventory; the other column is similar, but excludes landfill emissions, which are primarily unreactive and make up a large portion of the TOG inventory.

Finally, Table 6-14 identifies the approximate relationship between existing and new speciation profiles. This table was created by comparing the original profiles and new profiles for existing SCCs that were reassigned revised profile numbers. The relationship in the table is approximate

TABLE 6-9. Original classes of organic gases by profile code. (Tons per day)

£.

		(Tolls her day				
CODE	PROFILE NAME	OLE	PAR	ARO	CARB	ЕТН
	-	Ş	(	•		
	EXTERNAL COMBUSTION BUILER- NATURAL GAS	•	1.92	$\beta$ . 1.1	$\sim$	₹.
	BOILERS - REFINERY GAS	•	2.25	zi s	_ ∶	•
	-	•	78.8	S	j,	•
	AINE NATORAL		88.8 U	33.0	•	•
	0.00	•	ر د د د د	5 2	ર -	•
	DV METALS	,	6 . 5 - 23		٦.	٠
	77 7E	•	8 . LS	10.00 00.00	•	
	- c	20° 0	10.0			•
•	ASPHALI KOOFING BLOWING OFFKALION	10.00 10.00	ου·ω ου·	100° 100° 100° 100° 100° 100° 100° 100°		80.80 80.00
		200° 5	20° I		٠.	•
	FK- NAI.	8 8	<b>⇒</b> ∠ .	•	•	•
	KUAU A	. 85 88. 1	28.32		ત્ર. જિ.	/9·8/
		10. E	8.15	•		
	M COVEKED	M . 58	1.26			10.00 10.00
	GIAIVE EMISSI	8. S.	80.80 0.00		٠	
	VEALS - KEFIN	10.00 10.00	10.36 2.36	99.88 88.88	10.10 10.10 10.10	પ્ર <b>છ</b> ાછા જ
	1. VE 3 - LT	18.8 2.5	. й. I.З			0.00.0d ∴ ∴
	1	80.00 80.00		Ø. Ø. ĕ.		13. B
	KEFINERY - CALALYIIC KEFOKMEK GENEKAL FO	88.8 88.8	Z.13	•		0.99
	VAKNISH MANUFACIUKING	B. B. B.	18.04 2.04			13. 10. 10. 10. 10. 10. 10. 10. 10. 10. 10
	· · · · · · · · · · · · · · · · · · ·	Ø. Ø.	16.00			Ø. 00
	PESTICIDE USE- COMPOSITE DOMESTIC & COMM	Ø. Ø.	9,61			ø.on
	CTURI	00.00	Ø. WØ	Ø.00		8. Sist
	RING	Ø.9.9	Ø.03			ø.og
	PERCHLOROETHYLENE CLEANING SOLVENT	Ø.53	æ.	88.88		B. BB
		33.56	24.58	00.00		0.00 B
	1,1,1-TRICHLOROETHANE CLEANING SOLVENT	M.50	Ø.O.O	80.8		Ø.90
		Ø.93	Ø.00	0.01		Ø.00
	ENERAL	B.O. B	9	6.68		13. B
	SKIN	1.03	57.51	26.37		B.O.B
		0.60	1.57	00.00		B.BU
21	PE/PRUNING	g.ag	B. BB	ø.øø		B. B.B
22	OR- SOLID	8.88	Ø.03	B.OR		0.01
25	HOT AIR DRIE		4.19	9.77		Ø.00
27	RAL VARNISHIS	-	2.13	19.00 E		Ø.138
4 1	AAL PRIME		က	Ø.38		છે. ઉ
36	AL PRIMER	N. 3.0	29.93	3.33		Ø.00
	. AUMESIVES		2.23	4.99		10. BB
	L FUKNIJUKE	10.16g	Й.34	6 <b>8.8</b>		Ø. Ø.
φ. Σ	LIAL FURNITUR	80. 8	2 M. 9	1.38		Ø.⊍∂
4 r D A	APEKBOARD PRO	80.08	7.75	1.23		
ים ים	(AL COMPOSITE	19.00 2.000	6.83	ű. 79	1.29	ଷ୍ଟ ନ
	JSIIE WOOD FU		W. W. W	13. BB		
	METAL	0.00	Ø.00			
	ALCUHUL BASE		ا ئن	W. W.	3	-
	CAL SULVEN	80.00 10.00	c.	9	4	Ś
	LUE	; کر	~ ૅ	~ 5	35	Ø. EØ 2. 2.2
			ж. Ж.	25	<b>B</b>	Ø
195	ARCHITECTURAL SURFACE COALINGS - COMPOSIT		59.18	8.27	1.85 2.05	8.88
10%	ESTIC SOLVENIS GENERAL COMPOS	Ø 6: . Ø	g . 7	2	תכ	Z

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CARB	######################################	ж ж ж ж ж ж ж ж ж ж ж ж ж ж ж ж ж ж ж
ARO		86 86 86 86 86 86 86 86 86 86
OLE PAR	239 14.5 299 14.5 299 14.5 299 14.5 299 15.7 299 299 299 299 299 299 299 299 299 29	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
	ETON TATE S SP OL OPYL OPYL L SP AUTO S ALC SOLV SC	COH PRO NPO AUS AUS AUS AUS AUS
PROFILE NAME	ANDFILL EMISSIONS  ANDFILL EMISSIONS  NIMAL WASTE DECOMPOSITION  EER FERMENTATION - ETHANOL  OKE OVEN BLAST FURNACE - PROCESS GAS  CETONE - PAINT SOLVENT  URFACE COATING EVAP - ETHYL ACETATE  UNFACE COATING EVAP - ETHYL ALCOH  UNFACE COATING EVAP - ETHYL ALCOH  UNFACE COATING EVAP - SOLVENT - ISOPR  UNFACE COATING EVAP - SOLVENT - ISOPR  UNFACE COATING EVAP - SOLVENT - ISOPR  ETRO STORAGE - FIXED ROOF - HEXANE  RICHLOROFTHYLENE CLEANING SOLVENT - ISOPR  NATHELOROFTHYLENE CLEANING SOLVENT - ISOPR  UNFACE COATING EVAP - SOLVENT - BUTYL  UNFACE COATING EVAP - MAPHTHA  UNFACE COATING EVAP - SOLVENT - BUTYL  UNFACE COATING EVAP - SOLVENT - CELLO  ITRUS COATING EVAP - SOLVENT - CELO  ITRUS COATING EVAP - SOLVENT	IXED ROOF - BENZENE IXED ROOF - CYCLOHEXA IXED ROOF - PENTANE VALVES & FLANGES - C SEALS - COMPOSITE VALVES & FLANGES - C SEALS - COMPOSITE VALVES & POLYPROPY INTO THE SOLVENTS - MIXED IPOSITE IXPROPYLENE IXPROPYLE
CODE	0.0 L L L L L L L L L L L L L L L L L L	2222 2222 2222 2222 2222 2222 2222 2222 2222

TABLE 6-9 (concluded).

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CODE	PROFILE NAME	016	PAR	ARO	CARB	ETH
5118 5128 5128 5128 5128 5128 5128 5128	AEROSOL SPRAYS-NON-SYNTHETIC AEROSOL SPRAYS-SYNTHETIC COMPOSITE NATURAL GAS DIESEL EXHAUST(ALDEHYDES IN EMISSIONS) FORMICA MANUFACTURING OPEN BURNING DUMP-LANDSCAPE/PRUNING (MOMETHYLENE CHLORIDE CATALYST EXHAUST (ALDEHYDES IN EMISSIONS) 1979 EXHAUST COMPOSITE 50/50 (ALDEHYDES) NON-CATALYST EXHAUST (ALDEHYDES IN EMISS) SPECIES UNKNOWN-ALL CATEGORY COMPOSITE	8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	16.66 8.888 12.82 8.85 8.85 4.41 8.88 8.88 8.88 8.88 8.88	8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	, \$\rho \tilde{\rho} \r	8
TOTAL		15.98	601.56	91.21	, 28.68	42.64

TABLE 6-10. Revised classes of organic gases by profile code. (Tons per day)

CODE	PROFILE NAME	01.5	PAR	ARO	CARB	H H
ಣ⊸	EXTERNAL COMBUSTION BOILER-NATURAL GAS BOILERS-REFINERY GAS	4.28 8.58	1.92	8.11 8.82	84.88 84.88	80°8 80°8 80°8
	(	BB E	28.32	5. <b>9.3</b>	M. 555	8. E1
	INTERHAL COMBUSTION ENGINE-TURSINE-NATURAL GAS INDUSTRIAL ICE-DISTILLATE OIL	8.08 8.05	क इ.स. इ.स.	77.77 11.17	2 (	के ख इ.स. इ.स.
	INDUSTRIAL ICE-NATURAL GAS-RECIPROCATING	3.16	B. J. B	11.01	10.15	W. 15.16
	COKE OVEN STACK GAS-PRIMARY METALS IRON SINTERING-PRIMARY METALS	) (1) (2) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	5.13 3.61	हा. हा. हाड्य	18	50.00 50.00 50.00
	ASPHALT ROOFING-BLOWING OPERATION	$g_{2},g_{3}$	E S			18 . NO
		11. 12. 12. 12. 12. 12. 12. 12. 12. 12. 12. 12. 12.	1.82	1.1818 21.1818	(S. ) (S. )	27. 27.
	ASPHALT CONCRETE-ROLERY DRYER-WALDRAL GAS FIRED ASPHALT	ر نه 12.25 13.00	3.283	2. 2. 3. 3.	<u> </u>	8 8 8 1 . 1
		A . 600	. ii. 1.1	50.00	10° . 10	W. J.
	REFINERY-FUGITIVE ENISS, FROM COV, BRAIN/SEP FITS REFINERY COOLING TOWERS FUGITIVE EMISSIONS	97 - 1/1.67 54 - 676	ा । अ. । अ. ।	ब. इ.स. इ.स.	स् इ.स. इ.स.	8.1.18 8.1.18
	PETRO INDUSTRY-COMPRESSOR SEALS-REFINERY GAS	9.84	9.43	9.99	14. S. 3	13.1.18 13.1.18
	PETRO MARKETING-RELIEF VALVES-LPG	Ø.04	න. විශ •	8.83 8.53	26 2 √. 10 -	0.50g
	REFINERY-CATALYTIC REFORMER-GENERAL FUGITIVE EMISS	811. E	₹ 52 . 05 . 05 . 05	2	15. 15. 15. 15. 15. 15. 15. 15. 15. 15.	8. 7.5 3.7.5
		1. 111.	9.19	4.99	8.32	Bir. B
	PRINTING INK COUNTING-GENERAL PESTICINE HSE-COMPOSITE BUMESTIC & COMMEDCIAL	10. 11. 11.	5.5.€ •	8.00	100 to 10	क इ.स. इ.स.
	ETHYLENE DICHLORIDE MANUFACTURING	2. 10. 10. 10. 10. 10. 10. 10. 10. 10. 10	10.10 10.10 10.10	 5.8.5	18. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	55.55 52.55
	FLARES-CHEMICAL MANUFACTURING	38. B	. S. S.	16.15.4	(1) (2)	2017. 2016 1017. 1016 1017. 1016 1016 1017. 1016 1017. 1016 1016 1016 1016 1016 1016 1016 1016
	PERCHLOROE HYLENE CLEANING SOLVENI STODDARD CLEANING SOLVENT	1. 60 1. 10 10. 10 10. 10 10. 10 10. 10 10. 10 10. 10 10. 10 10. 10 10 10 10 10 10 10 10 10 10 10 10 10 1	12.18.9 24.68	9.18.1 1.18.1	24 12 13 13 13 13 14 13 15 13 16 16 16 16 16 16 16 16 16 16 16 16 16 1	
	1,1,1-TRICHLORDETHANE CLEANING SOLVENT	1.54	15. 19.16	8.67	W W.	10.11.55
	DEGKERSING-TOLDENE SURFACE COATING SOLVENT-GYNERAL	J. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	(2) (2) (2) (2) (2) (3) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	ं ए हां क हा थ	<b>ह</b> ं ें - - - -	# # # # # # # # # # # # # # # # # # #
 	GASOLINE EVAP-COMP WORK, & BREATH, TANK LOSS	7.88		8.83	( ) ( ) ( )	
) (8)	DAT FUEL EVAPORATION	9.5.E	98.11	5.35	18	10. 16. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18
71 K2 C1 K2	BAR SCREEN WASTE INCINERATOR-SOLID WASTE POLYMERIC SURFACE COATING-HOT AIR DRIFD	24.25.25 22.25.25 22.25.25	हा हा स म्ह स म	म् अक्ष	30 75 30 75 30 75 30 75	
27	SURFACE COATING EVAP-GENERAL VARNISH/SHELLAC	.j.j.j.	1.65	3.33	**************************************	1 (S) 1 (S) 1 (S)
्त्र (१)	SURFACE COATING EVAP-GENERAL PRIMER	11.64	Min. r	18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	17 17 35 35 35 35 35 35 35 35 35 35 35 35 35	[8] · 8
- T	SURFACE COATING EVAP-LABEL ADHESIVES		्र हा जिल्ला हा जिल्ला हा जिल्ला	2	ان د د از د	8
(N) (	SURFACE COATING EVAP-METAL FURNITURE ADHESIVES		11.143	ET. M	184 134	15.7° B
ام ما در در	SURFACE COMPING LACQUER-WEBAL FORNTHORE Surface coat lecourr-padesroard drop & contained		15 Mil	11 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	26 2 2 2 2 2 2	
20	MEL	• •			ু হৈ ভূ হৈ	
5.7	SURFACE COAT, EVAP-COMPOSITE WOOD FURNITURE ENAMEL	3.34	N. 1976	1	10 m	16
55 5 5 5	2	7.38	•	M. M.	- :	
328	EVAPOLATION-GRAVURE PRINTING-GENERAL SOLVETAR	20. 20. 20. 20. 20.	 	%. 	ન ં : ` : જે	ह है। ज () ज ()
ය . ස	PETRO STORAGE-FIXED ROOF-TOLUENE		1.3.14	54.43	W. 728	16.0.0
ವರಿಗಳ ರಾಧ	METHAME SOURCE	7.64	· 🛊 :	7.007		Ø . MØ
197	CIORAL SORFACE COMITNES-COMPOS C SOLVENTS-GENERAL COMPOSITE	7 - 20 20 11 - 19 26	40.00 20.62		  	6. 5. 5. 6
2.92	ILL EMISS		•	3.28	7	35.8

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1000	101	0,5	ž	Color		
	ANIMAL WASTE DECOMPOSITION	83.58	15.73	и.ви	9.55	60,716
	SEER FERMENTATION-ETHANOL	M. HU	•	11.41	:	
	COKE OVEN BLAST FURNACE-PROCESS GAS	135. 136. 136.	67. <b>91</b>	हा. इ.स.	লৈ : ড. ড	න ව ලැප
	ACELONE-FRINI SOLVEN: CHREACE CONTING EVAD-ETHVI ACETATE	20 E 20 E 20 E 20 E	•	11 (f) 12 (f)	•	2
	SURFACE COATING EVAP-METHYL ETHYL KETONE	3.5.5 3.5.5	2.26	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1		Ø.23
222	SURFACE COATING EVAP-CELLOSOLVE ACETATE	87.78		13.63	13. S.	27.18
	SURFACE COMPING EVAPEXFLENE SOLVENI	86. 8. 20. 20. 20.	—ा । इ. १ इ. १	18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	0.00 20.00	104. B
	SUKFACE CORTING EVAPHTHEMAL OFILES OFFILES	20. 17. 20. 17. 20. 20. 20. 20. 20. 20. 20. 20. 20. 20	ନ ପ ବୁଧ୍ୟ ଅଧିକ	17. 18. 18.	عر تو د در تو	? ? ?
	SONEACH CONTING SOLVENTEITHE ARCOHOL CHEACH COSTING EVAPING VENTINGOPPUL ALCOHOL	10 m			3 'E	: ``;
	SURFACE COATING EVAP-SOLVENT-ISOPROPYL ACETONE	14 . E. S.	6.14	10.00 E	:	10.1.10 10.1.10 10.1.10
	SURFACE COATING EVAP-SOLVENT-LACTOL SPIRITS	B.C. C.	1.37	11.20	16.134	18 · 18 18
	PETRO STORAGE-FIXED ROOF-HEXANE	80. 80 800. 80	0.15	•	18 · 18 1	10. 10. 10. 10. 10. 10. 10. 10. 10. 10.
	ONNEH STORY OF THATEINE CLEANING SOLVENI	18 . P.	56 57 57 56 57 57 56 57 57	28. B	اوا ان ان ان ان	
	SYNTHEIC KUBBER AUTO TIKE PRODUCTION Sube coat prince water bas anto baint spray rooth	10.10 10.00	2 2 2 2 2 3 2 3	2	9 X X	: :
		57 - 57 S	20 . 20 . 20 .	S. 63	2	
	SURFACE COATING EVAPORATION-NAPHTHA	3.88	1,69	1.93	7 B	97.79
	SURFACE COATING EVAPORATION-SOLVENT-BUTYL ALCOHOL	B.3. M	•		86.48	16.113
	SURFACE COATING EVAPORATION-SOLVENT-CELLOSOLVE	31.05	9.32	£ 55.5	15.10	80.08
	CITRUS COATING WAX-FLAVORSEAL 320-0320	19. S.S.	19.19	11.02	8.5.5	30 · 30
	COMPOSITE CRUDE OIL EVAP-FIXED ROOF-PRODUCTION	9. 15.8 11.8	æ: α:	11. 12. 13. 13.	± 2. Σειτο Σειτο	
	SKUDELOIL EVAPIVAPUK COMPIPROM PIRED KOOF LANKS Seteo etobaceleiven boog-sempene	1. S.	27.07	70 TO	त्र 'ड : : ड	5. (3) (1) (2) (3)
	FRING STONASETINED NOOF SERVENE DETTOD CTODACETETVEN DOOR CHECKINE	G 455	Z * 1 1 G 6/G	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	10 to	20 20 20 20 20 20 20 20 20 20 20 20 20
	PETRO STORAGE FIXED ROOF-UTCLOMEANNE PETRO STORAGE FIXED ROOF-UTPTIANE	5. 5. 5. 5. 5. 5. 5.	g .83	5. S.	5	
	PETRO STORAGE-FIXED ROOF-PENTANE		$g_{S,S}$	30° 13	S. 1. 18.	8.88
	EVAP-FLEXOGRAPHIC PRINTING PRESS-N-PROPYL ALCOHOL	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	2.91	11.193	18. S. A.	83.88
	OPEN HEARTH WITH OXYGEN LANCE-STEEL PRODUCTION	•	19.41.5	.U. 6%	10 miles	88 38
	FORREST FIRES	•	11.92	H.BG	. 55 55.55	22.25
	REFINERY-PIPES, VALVES & FLANGES-COMPOSITE	30.00	cn -	्राष्ट्र १८००	30 ·	51 5 1 32 5
	KEFINEKY-FUMY SEALV-COMPOSITE PATALVST LIGHT BHTV VEHICLES-EVMANST	7. M.		/ <b>* .</b> 0.	± 3.00 ∴  	10 10 10 10 10 10 10 10 10 10 10 10 10 1
	CAINETS! EXGEL BOLL VEHICLES-EXHAUS! CAROLINE EVAPORATION-CANICTER-INV HOT ROAKS		7. T.	2. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	8	1
	NON-CATALYST LIGHT DUTY VEHICLES-EXHAUST	± € €	24.27	8.17	2 · · · · · · · · · · · · · · · · · · ·	6.36
	DIESEL EXHAUST(ALDEHYDES NOT IN EMISSIONS)	W. 1611	36 6	M. 00	8.7.8	8.88
	SIDUAL	9.24	5.75	8.17	M W.	•
	PLASTICS MANUFACTURE-VC & POLYPROPYLENE MIXED	1,51	•	ED. 3	18	1.82
	HALOGENATED CLEANING SOLVENTS-MIXED			•	জ : ং. জু:	•
	OFF EXHAUST-COMPOSITE		٠		.a.; ⊗.•	
	PLASTICS NFG-VINYL CHLORIDE PLASTICS MEGLED! VPROPYLENE	: 15. 15. 15. 15. 15. 15. 15. 15. 15. 15.	10. 10. 10. 10. 10. 10.		50 E	छ . स
	COCCANTS ASSUMENT OF STREET	100 - 1	٠	300	2	•
	SURFACE COATING EVAPORATION-METHYL ISOBUTYL KETONE					
	EVAPORATION-CHEVRON WEED OIL	M. 10E		• •		
	COMPOSITE INDUSTRIAL DEGREASERS	1878.15	5.36	•	$ij \dots jj$	
	COMPOSITE DRY-CLEANING SOLVENTS		й Эй.	•	18	83. ±38
	PRINTING EVAPORATION LOSS-GENERAL	_		5.93	पा है  डिस्ट डे	20 . 20 20 . 20 20 . 20
	AEROSOL SPRAYS-WON-SYWIHELIC	B 34	۵ ۵ ۵	3.8.€ 3.8.€	57 € 72 20	£ 6

3000	PROFILE NAME	01.6	PAR	ARO	CARB	CTH
519 52 <i>8</i> 521	AEROSOL SPRAYS-SYNTHETIC COMPOSITE MATURAL GAS DIESEL EXHAUSTCALDEHYDES IN EMISSIONS)	16. 53 15. 53 15. 53 15. 53	5.34 5.34 5.34	55.25 5.25 3.35 3.3	30 10 10 10 10 10 10 10 10 10 10 10 10 10	87.78 87.18 87.18
125 120 120 120 120	DES IN EMISSIONS/	ن ن و تون و تفر	16.30 B.70		· ` ; `	
57 57 57 57	DSCAFEZFRON: (MOD: NVB 1	19.50 19.50 19.50	4 . 4 1 Ø . 6 6	11.0% 14.0%	1 (5) 1 (5) 1 (5)	4.40 10.60
525 526	CATALYST EXHAUST (ALDEHYDES IN EMISSIONS) 1979 EXHAUST COGP. 5Ø75Ø (ALDEHYDES IN EMISSIONS)	14.15.25 18.15.25	50.755 50.555	11. AB	13. 15. 13. 15. 13. 15.	1377 138 1377 138
27	(ALDEHYDES IN EMISSIONS)		े देद : ∘	•	(35) 3 (35) 3 (35) 3	
© © 1 €	ALEGURY COMPOSITE DED REGULAR~SUMMER BLEN	ड. ड. ड. ड. ड.	- 100 - 100	77 . 13 . 13 .	2	1 - 1 4 15 - 15 5
62	DED REGULAR-SUMMER BLEN	18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	\$ . \$5.6 \$ . \$5.6	12.13 13.13 13.13 13.13	25 S	ख्या इंड्र
) T :	LIQUID GASOLINE - LEADED PREMIUM-SUMMER BLEND	18. S.	8. 8. 8. 8. 8. 8. 8. 8. 8. 8. 8. 8. 8. 8			
	DED REGULAR-SUMMER BLEN DED REGHLAR-SHMMER BLEN		18.18.18 18.18.18	J. 1918 H. 186	26 25 27 25 27 25 27 25	ख. इ. १५ इ. १५
67.	DED PREMIUM-SUMMER BLEN		20 E	J.03	# 1 m	
ල ග වේ ජ	OED PREMIUM-SUMMER BLEND SITE OF PRODUCT-SUMMER RIEN	18. 18.0 18.0		क्ष.  -   63	ड <u>र</u>  र : इ	55.55 57.55
	SITE OF PRODUCT-SUMMER	3.92	67.36	2.4.3 1.4.1		
	ATING-COMPOSITE LACO	ار الله الله الله الله الله الله الله الله	٠. در ت	•	हा हर हर १९५	
1213	ATING-COMPOSITE PRIMER	3.58 3.58		3.85	6.26  1.00 1.00 1.00	
	TING-COMPOSITE ADHES	الله . الأواد الله الله الله الله الله الله الله ال			्र ्र	हा. इ.स.
	SEOW CORE ASTRALT MEDIUM CURE ASPHALT	n :0: \$ :0: 10: 10: 10: 10: 10: 10: 10: 10: 10:			8	
	COATING-VATER BASED PA	₩. State			ස දි සු	10 . M.S.
	COALING-COMPOSITE JGINE-RECIPROCATING	7 : S		<b>7.</b> . ₹ 67 . ₹ 67	18.00 1.00 1.00	्र ज्या ज्या ज्या
	DED REGULAR-WINTER BLEND	19. 19. 19. 19. 19. 19. 19.	1.2. B	2. N.	18 × 18	
	SED REGULAR-VINTER BLE	100 to 10	88.88 22.22	H. HA	) (9) (9)	15, 13 1, 13 25, 23
	ED PREMIUM-VINTER SLE	5 K . 38	5.55 5.55 5.55	1.93 19.93		2
	ED REGULAR-WINTER BLE	1878 N	Ø. W. Ø		B B.	+
	GASOLINE VAPORS- LEADED REGULAR-VIHTER BLEND GASOLINE VAPORS-HNIFADED PREMIUM-VINTER BLEND	2 5 2 7 2 7	ा हाड्ड इंड	्ट इ.स.च इ.स.च	8. 10. 10.	10 10 10 10 10 10 10 10 10 10 10 10 10 1
	ED PREMIUM-VINTER BLEN	2		7. BB	5 55 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	 
	ITE OF PRODUCT-WINTER BLEN	9.80		•		255.25
	TIE OF PRODOCIEWINIER BEEN TIMETAL BECHIABLEAUMED BEEN	2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2	5. 2. 2. 2. 2. 2. 2.		30 % 31 % 31 %	8. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3.
	UNLEAD REGULAR-SUMMER BLEND	2 (B) 2 (B)		50.00 50.00 50.00	1	ু জু ু জু জু জু
751	BUTADIENE-STYRENE (ABS) RESIN HFG.		14.16.66		F 18	10.00
752	SIN MFG.		•	٠		9577 <b>3</b> 4
ე სე უ —:	STAKERE CHLOROSGLVE		: :: :: :: :: :: ::	ರ ಕ. ಶಿವರ ಪ್ರವ	20 (20) 	
ភែរ	4			•		8.00
	VES-LIGU VES-GAS	20 To 100 To	2.50 2.50 3.70 3.70	5. 3.3 5. 3.3		18. 18. 18. 18. 18. 18. 18. 18. 18. 18.
7:5				13. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18	ر ایم در این	
	INGS-UNSPECIFIE	J. W. D.	4.10	e 18. 0.	£.∴.9	970 BI

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162.99

653.57

16.79

TOTAL

TABLE 6-11. New chemical species in the revised inventory.

Saroad ———No.	Molecular Weight	Species Name
	1194911	
43215	56.10	ISCBUTYLENE
43243	68.13	ISOPRENE
43266	112.22	C-2-OCTENE
43267	127.05	1-NONENE
43268	140.27	1-DECENE
43269	154.30	1-UNDECENE
43371	118.18	HEXYLENE GLYCOL
43561	114.19	METHYL AMYL KETONE
43825	86.47	CHLORODIFLUOROMETHANE
43826	104.46	CHLOROTR IF LUCROME THANE
43827	154.47	CHLOROPENTAFLUOROETHANE
43828	170.92	DICHLOROTETRAFLUOROETHANE
43830	118.00	CHLOROFLUOROHYDROCARBONS
45402	122.13	BENZOIC ACID
50006	40.06	PROPADIENE
50028	148.00	PHTHALIC ANHYDRIDE
50029	98.00	MALEIC ANHYDRIDE
50030	76.14	CARBON SULFIDE
50031	60.08	CARBONYL SULFIDE
50032	128.26	3,5,5-TRIMETHYLHEXANE
50033	128.26	2,2,5-TRIMETHYLHEXANE
50034.	84.16	T-2-HEXENE
50035	84.16	C-2-HEXENE
50036	72.11	ISOBUTYRALDEHYDE
50037	96.17	1-METHYLCYCLOHEXENE
50038	127.05	C9 OLEFINS
50039	140.27	C10 ALKENES
50040	84.16	2-METHYL-1-PENTENE
50041	98.19	3-HEPTENE
50042	127.05	4-NONENE
50043	120.20	ISOPROPYLBENZENE (CUMENE)
50044	118.18	INDAN
50045	134.22	M-DIETHYLBENZENE
50046	128.19	NAPHTHALENE
50047	134.22	ISOBUTYLBENZENE
50048	116.16	INDENE
50049	120.20	C9 AROMATICS
50050	134.22	Clo AROMATICS
50051	126.59	2-CHLOROTOLUENE
50052	134.22	T-BUTY LBENZ ENE

TABLE 6-11 (continued)

Saroad No.	Molecular Weight	Species Name
	_	_
50053	96.94	1,1-DICHLOROETHYLENE
50054	112.22	2,4,4-TRIMETHYL-1-PENTENE
50055	112.22	2,4,4-TRIMETHYL-2-PENTENE
50056	86.14	ISOVALERALDEHYDE
50057	98.19	ETHYLCYCLOPENTANE
50058	112.16	TRIMETHYLCYCLOPENTANE
50059	112.12	DIMETHYLCYCLOHEXANE
50060	129.27	TRIMETHYLCYCLOHEXANE
50061	112.22	ETHYLCYCLOHEXANE
50062	141.24	DIETHYLCYCLOHEXANE
50063	154.30	N-PENTYLCYCLOHEXANE
50064	112.22	C2 CYCLOHEXANE
50065	126.24	C3 CYCLOHEXANE
50066	140.27	C4 CYCLOHEXANE
50067	154.30	C5 CYCLOHEXANE
50068	126.24	C3 ALKYL CYCLOHEXANE
50069	140.27	C4 ALKYL CYCLOHEXANE
50070	142.28	C4 SUBSTITUTED CYCLOHEXANE
50071	154.30	C5 SUBSTITUTED CYCLOHEXANE
50072	170.32	C6 SUBSTITUTED CYCLOHEXANE
50073	154.26	C4 SUBSTITUTED CYCLOHEXANONE
50074	118.17	BUTYL CELLOSOLVE
50075	130.19	C5 ESTER
50076	114.19	2-METHYL-3-HEXANONE
50077	114.19	HEPTANONE
50078	72.10	ALKENE KETONE
50079	136.24	TERPINENE
50080	90.12	BUTANDIOL
50081	60.11	ISOPROPANOL
50082	127.05	ETHYLHEPTENE
50083	182.35	TRIMETHYLDECENE
50084	146.23	C2 ALKYL INDAN
50085	132.21	
50086	166.27	ALKYL SUBSTITUTED CYCLOHEXANE C2 ALKYL DECALIN
50087	156.27	
50088	150.27	CARVOMENTHOL
50089	154.26	CARVONE
50090		ISOPULEGONE
50090	112.22	METHYLHEPTENE
	128.26	DIMETHYLHEPTANE
50092	187.87	1,2-DIBROMOETHANE

TABLE 6-11 (concluded)

Saroad No.	Molecular Weight	Species Name
50093	92.57	1-CHLOROBUTANE
50094	148.68	3-(CHLOROMETHYL)-HEPTANE
50095	88.15	ETHYL ISOPROPYL ETHER
50096	130.23	DIBUTYL ETHER
50097	128.19	2-BUTYLTETRAHYDROFURAN
50098	140.23	PROPYLCYCLOHEXANONE
50099	162.18	2-(2-BUTOXYETHOXY)-ETHANOL
50100	104.15	1-ETHOXY-2-PROPANOL
50101	130.23	2-ETHYL-1-HEXANOL
50102	116.21	1-HEPTANOL
50103	188.18	2-METHYL-2,4-PENTANEDIOL
50104	102.13	METHYL ISOBUTYRATE
50105	158.24	C8 ESTER
50106	211.19	SUBSTITUTED C7 ESTER
50107	218.24	SUBSTITUTED C9 ESTER

TABLE 6-12. Original organic gas species in weight percent.

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Weight percent. HT WEIGHT NT PERCENT *	,	20	8	•	me	8 8 11.3 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	. 73	B	8	2.00	ğ. Ø	1.8		7.8	9.1	8.9	7.45	2.7	2.7	1.0	u. 1 Ω. Ω	0.0	8.8	20. C.		1	8. S	<b>8 8</b>	, В.	82 (	7 8	2 5	S	Ø	SQ ?		8 8	3.8	9.1	Ø6.8
organic gas species in WEIGH	N N N	8	Ø.2Ø	0.61	1. W	8.83.9 8.83.		ANE B.BB	NE Y	8.13J	186.8	Ø.31	8.85	1.1.8 8.863	0.72	94 (	7.27	1.23	1.28	Ø. 480	8.48, 8.88	N . N .		10.0.0 10.0.0	Ø.129	8.776	2	2000. 2000.	18. 1821	8.886 8.050	2 S	(6)3. B	0.000	Ø.090	18.000 18.0000	2000.00 2000.00	330.0	1.375	Ø.451	Ø.4Ø5
NAME NAME	OF HEXAN	OF HEPTA	OF OCTAN	OF NONAN	OF DECANE	OF TRIDECAN	OF DODECANE	OF TETRADEC	OF PENTADEC	2 9	OPARAFFI	SPIRITS	PIRITS	OF PENTEN	OF PENT	TERPENES	FIHANE	ETHVLENE			CYCL OPROPANE	1	ETYLENE	- I - P.E.N		Zυ	n N n P I H	TENE	DIENE	_ }  -  -	N-FENIANE 3-METHVI - L-BUTENE	Д	-1-B	PENTEN	NTENE	Z-MEIHYL-Z-BUIENE Z-MFTHVI PFNTANF	PENTAN		HEPTANE	OCTANE
SAROAD	-	3	7	٦,	۲ ۲		Ξ	_	_;	43115	_	_	311	$\frac{312}{312}$	312	312	2 2	320	320	280	328	320	87.5 87.5 87.5 87.5 87.5 87.5 87.5 87.5	2 1 2	321	43214	321	321	321	327	0 C 7 C 7 C	322	22	322	325	43229	323	5	323	323

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS.

TABLE 6-12 (continued)

WEIGHT PERCENT *	1	50	J. 756	200.	B. B. 4	. <b>M</b> 3	. 42	Ø.822	.82	. Ø3	Ø.691	B.BEB	Ø.858	Ø.Ø51	Ø.ØØØ	0.000	Ø. Ø.Ø	0000	Ø. Ø Ø Ø	B.BBB	Ø. ØØØ	Ø.138	0.000	0000.0	Ø . Ø Ø Ø	g.gg	0.017	0.014	8.013	0.009 0.009	8.58.58 8.88.58	0.000.00 0.000.00	2.22.2 2.22.22 2.22.22	8. BEIG	0.276	6.000	B.BUB	B. BUB	0.000	Ø.000	•	8.80B	Ø.000		8.647	89∙	٠	•	0.271	•
WEIGHT	0	2.538	1.	110.0	. 144 	8.W.S	0.635	19. Ø 1 Ø	$g$ , $g_{11}$	Ø.Ø14	Ø.3Ø8	Ø. MBB	<b>Ø.</b> 888	0.023	u.aaa	Ø. ØØØ	Ø.03Ø	Ø. Ø Ø	Ø.800	Ø. ØØØ	Ø.838	Ø.Ø62	Ø.000	ŭ.gag	D. 600	Ø.800	19. 15. 18. E	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	3 3 3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	18.18.44 18.18.44	35.55 5.55 5.55 5.55	0.000 0.000	≈ :	Ø.388 Ø.888	Ø.123		U. BBB	Ø.808	<b>Ø.</b> 808	Ø . 50 kB	<b>9.000</b>	Ø.698	Ø.000	æ.	Ø.289	0.753	Ø.	_		W.W14
NAME	NOM A ME			2 4 4 2				ı	N-TRIDECANE	N-TETRADECANE	N-PENTADECANE	METHYLCYCLOHEXANE	METHYLCYCLOPENTANE	CYCLOHEXANONE	OCTENE	3-METHYL-T-2-PENTENE	2,4-DIMETHYLPENTANE	METHYLCYCLOPENTENE	CYCLOHEXENE	2,3-DIMETHYLPENTANE	Z-METHYLHEXANE	2,2,4-TRIMETHYLPENTANE	2,4-DIMETHYLHEXANE	Z,5-DIME HYLHEXANE	2,3,4-TRIMETHYLPENTANE	Z, 3, 3KIME IHYLPENIANE	HE XAUECANE	NET - AUECANE OCTANECANE	NONADECANE	FICOSANE	HENETCOSANE	DOCOSANE	ETHYLCYCLOHEXANE	C6 OLEFINS	C8 OLEFINS	2,2-DIMETHYLBUTANE		4-METHYL-T-2-PENTENE	C7-OLEFINS		Z, Z, 3-TRIMETHYLPENTANE	4-MEIHYLHEPTANE	ليا	Z,Z,S-TRIMETHYLPENTANE	∢.	HYL ALC	Ļ	VO-FROF	N-BUITL ALCOHOL	71 00 0
SAROAD	A222E	2000	1000	1001	1004	7 7 7 7	43248	43255	43258	43259	43260	43261	43262								3275			32/8	32/3	37810		2026	2000		3286	3287	3288	3289	329Ø		3292	3293	3294		3296	7628	3298	3299	3 C	43387	າຕ	3000	4330K	3

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-12 (continued)

WEIGHT PERCENT *	0.898	•	300	22 <b>7.</b> 00	8.888		•	<b>B.B</b> BB	Ø.098	0.832	8000.80 8000.80	A. ASA		. 2 <i>B</i>	•	B.BUS	g.ggg	8.47c	0.78.8 0.78.8	0.00.00	Ø.323	0.071	0.417	Ø.199	B. B.B.B	8. 18.318 8. 85.88	0.00.00	β.179	1.854	2.333	0.837	Ø,391	10 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0.00.00 0.00.00 0.00.00	8 8 8 8	Ø.154	.18	•	0.018	•	B.BUD	Ø.000 8	20.00 20.00 20.00	0.00.00 0.00.00	
VEIGHT PERCENT	8.698	BBB.BBB	8.838 8.888	0.000 0000	ggg.g	8.000	•	Ø. ØØØ	8.044		•	gaa. g.ggg	Ø.153	Ø.893	0.252	Ø.000	19.888	8 213 8 811	2 3 3 T	8.812	8.144	0.031	ø.186	8.389	8.88.88 8.88.88	8.888 8.888	2.25.25 2.25.25 2.25.25	8.88.8	Ø.827	1.841	0.017	Ø.174	8.50.50 8.50.50 8.50.50	8.088 888	ž. žžž 8. žžď	0.046	Ø.Ø46	Ø.90g	Ø. HØ8	Ø.000	Ø. ØØØ	18.888 21.17	ন হ হ হ		
NAME		TERT-BUTYL ALCOHOL	Z-MEIHOXYEIHANOL 2-FTHOXXETUANOL	1-T-2-C-4-TM-CVCI OPENTANE	: ∸	ETHYL ETHER	GLYCOL ETHER	1	PROPYLENE GLYCOL	ETHYLENE GLYCUL TETRAHVDROFIBAN	ACETIC ACID	14	ETHYL ACETATE	ATE	N-BUTYL ACETATE	A   E	CELLUSULVE ACELATE	METHY! AMV! ACETATE	ATF	DIMETHYL FORMAMIDE	ISOBUTYL ISOBUTYRATE	2-ETHOXVETHYL ACETATE	FORMALDEHVDE	ACE IALDEHYDE PROBRIONAL DEUVOE	_	C3 ALDEHYDE	C5 ALDEHYDE	C8 ALDEHYDE		KETONE	N-BUTYL KETONE	MEIHYL ISOBULYL KETONE ETHVLENE OVIDE	, ц	, Ч	2	ETHYLAMINE	TRIMETHYL AMINE	METHYL CHLORIDE	DICHLOROMETHANE	HLOKOFOKM	CARBON TETRACHLORIDE	CARBON LETRABROMINE TRICHIORDEL DIROMETHANE	THV! CHIORIDE	ū	
SAROAD	43308	3.Q	433110	43312	43320	43351	43367	43368	43369	4337B	43404	43432	43433	43434	43435	43438	4 6 4 4 6	3445	3446	3450	3451	3452	3582		3518		3512	3513	3551	3552	4 45 5 4	4350 <i>k</i>	43602	43702	43704	3721	3740	38Ø1	2882	43383	43684	າຕ	ľ	(7)	

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-12 (continued)

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WEIGHT PERCENT *	01 5172	ONO.O	1.246	$\beta$ . 11 $\beta$	B.BBB	Ø.15Ø	8.898	8.039	Ø.174	Ø.125	0.492	2.255	BOB.	ø.638	Ø.293	. Ø. Ø32	0.451	B.BBB	0.952	3,488	0.284	Ø.000	Ø. Ø <i>B</i> Ø	Ø.000	00.00	Ø . Ø Ø Ø	Ø . Ø Ø Ø	888. B	9,9,9	M.M63	8.58.58 8.58.58	10.101010 0 0000	5.50 8.00 8.00 8.00	g . g g g	8.878	Ø. Ø. Ø	•	Ø.000	B.BBB		B.BBB	•	•		6.045	0.000	0.072	_	В
WEIGHT PERCENT	B. 224	B. BBB	₩.722	0.049	N.000	8.867	8.688	8.817	878	Ŋ. Ø56	Ø.219	1.061	Ø.WW	0.285	Ŋ.131	Ø.Ø14	$\varnothing.2\varnothing1$	Ø.900	0.424	1.611	Ø.127	Ø . Ø Ø Ø	B.BBB	Ø.BBB	Ø. BBB	M. 1338	8 . USB	80 . 00 B	8 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	878.8 888.8	8 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	м. В в в в	Z :	BEE. B	Ø.931	B.BBB	0.007	Ø.000	Ø.000	<b>B. BBB</b>	g.ngg	Ø.038	B.BBB	g. ggg	V. 1920	Ø.000	Ø.832	900.0	Ø.843
NAME	1.1.1-TRICHLOROFTHANF	ETHYLENE DICHLORIDE	PERCHLOROETHYLENE	METHYLENE BROMIDE	1,1,2-TRICHLOROETHANE	TRICHLOROTRIFLOUROETHANE	TRIMETHYLFLUOROSILANE	DICHLORODIFLUOROMETHANE	TRICHLOROETHYLENE	VINYL CHLORIDE	NAPHTHA	ISOMERS OF XYLENE	DIMETHYLETHYLBENZENE	ISOMERS OF ETHYLTOLUENE	ISOMERS OF BUTYLBENZENE	I SOMERS OF DIETHYLBENZENE	ISOMERS OF TRIMETHYLBENZENE	ISOMERS OF PROPYLBENZENE	BENZENE	TOLUENE	ETHYLBENZENE	O-XYLENE	M-XYLENE	İ	1,3,5-IRIMETHYLBENZENE	ZEN	N-FROFYL BENZENE O-FTUX TOLITAN		MIEINYLIOLOENE Yroylohyviornyrnr	CEC-BUTVIBENE	STABENE .	A-METHVI STVRENE	1.2.3-TRIMFTHVI BENZENE	TETRAMETHYLBENZENE	TRI/TETRAALKYL BENZENE	ROP	PHENOLS	XYLENE BASE ACIDS	MONOCHLOROBENZENE	1,4-DIOXANE	2,3-DIMETHYLBUTANE	2-ETHYL-1-BUTENE	C-3-HEXENE	2-METHYL-2-PENTENE	HEPTENE	PROPADIENE	METHYLNAPHTHALENE	ETHYLNAPHTHALENE	DIMETHYLNAPHTHALENE
SAROAD CODE	. 1		43817	43819	43820	43821	43822	43823	43824	43860	45101	4510/2	451.03	45104	451.05	45106	4518/	45108	18254	452ÿ2	45283	45204	45205	45206	4520/	45688	45789	4061	45212	45216	12010 12000	45221	45225	45232	45233	45234	453ØØ	454.01	45801	46201	5.0 EU	58682	5.00.003	5.0.004	50005	5 <i>Ø</i> ∅∅6	<b>5</b> ØØ1 <b>Ø</b>		50012

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-12 (concluded)

58813         PROPYLNAPHTHALENE         8.811           58814         TRIMETHYLNAPHTHALENE         8.817           58815         ANTHRACENE         8.817           58816         METHYLANTHRACENE         8.851           58817         DIMETHYL ETHER         8.892           58818         DIMETHYL ETHER         8.873           58828         CRYOFLOURANE (FREON 114)         8.808           58820         CRYOFLOURANE (FREON 114)         8.808           58821         O-CRESOL (2-METHYLBENZENOL)         8.808           58822         M-CRESOL (3-METHYLBENZENOL)         8.808           58823         B-CRESOL (4-METHYLBENZENOL)         8.808           58824         B-CRESOL (4-METHYLBENZENOL)         8.688           58825         A-PINENE         8.688           58826         B-PINENE         8.688           58826         B-LIMONENE         8.688	SAROAD	NAME	WEIGHT PERCENT	WEIGHT PERCENT *
ANIHKACENE METHYLANTHRACENE DIMETHYL-2,3,DIHYDRO-1H-INDENE DIMETHYL ETHER CRYOFLOURANE (FREON 114) B-METHYLSTYRENE O-CRESOL (2-METHYLBENZENOL) M-CRESOL (3-METHYLBENZENOL) P-CRESOL (4-METHYLBENZENOL) BENZYLCHLORIDE A-PINENE D-LIMONENE	5.0013 5.0014	PROPYLNAPHTHALENE TRIMETHYLNAPHTHALENE	9.811	8.824 8.839
DIMETHYL ETHER CRYOFLOURANE (FREON 114) B-METHYLSTYRENE O-CRESOL (2-METHYLBENZENOL) M-CRESOL (3-METHYLBENZENOL) P-CRESOL (4-METHYLBENZENOL) BENZYLCHLORIDE A-PINENE B-PINENE D-LIMONENE	58815 58816 58817	ANIHKACENE METHYLANTHRACENE DIMFTHYL -2 3 DIHVDRO-1H-INDENF	8.551 8.531 8.531	8.893 8.893 8.893
B-METHYLSTYRENE O-CRESOL (2-METHYLBENZENOL) M-CRESOL (3-METHYLBENZENOL) P-CRESOL (4-METHYLBENZENOL) BENZYLCHLORIDE A-PINENE B-PINENE D-LIMONENE	58818 58819	DIMETHY ETHER  CROCHOLOGICAL TIA	8.873 8.873	8.164 8.439
M-CRESOL (3-METHYLBENZENOL) P-CRESOL (4-METHYLBENZENOL) BENZYLCHLORIDE A-PINENE B-PINENE D-LIMONENE	58828 58828 58821	B-METHYLONDEN TO COLOR OF COLO	3.2.2. B.BBB B. BBB	2.22 6.888 8.888
BENZYLCHLORIDE A-PINENE B-PINENE D-LIMONENE	50022 50022 50023		B. BBB B. BBB B. BBB	8.888 8.888 8.888
A-P INENE B-P INENE D-L IMONENE	58824	$\vec{\Box}$	8.000	O.BEB
D-LIMONENE	5 <i>00</i> 25 5 <i>00</i> 25	A-PINENE B-PINENE	8.888 8.888	<b>B.BBB</b> 8 868
	50×27	D-LIMONENE	B.OBB	8000.0

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-13. Revised organic gas species in weight percent.

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SAROAD	NAME	VEIGHT	WEIGHT PERCENT *
		:	
	ISOMERS OF HEXANE	્ર. ઇ.13 ૧૧૧૩	1.1% 2.0%
	ISOMERS OF OCTABE	4.337	9 1
	ISOMERS OF HONANE	. 50. . 10. . 10.	~າ
	ISOMERS OF DECANE	1.119	2.5:14
	ISOMERS OF UNDECANE	4.35.4	£.122
	ISOMERS OF TRIDECANE	813.7	N. M. 1
		Fig. :	19. 19. 19. 19. 19. 19. 19. 19. 19. 19.
	ISOMERS OF TETRADECANE	5. 3.25 3. 3. 38	S: 33. 38
			8.8.8 8.88.88
	C-/ CYCLOPARAFFINS	77/-6	75.7.8
	CIC CYCLOPAKAFFINS	23 % · / ·	6 8 8 1 1 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8
	MINERAL SPIRITS	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	6 A 3
	LACTOL SPIRITS	7.946	Ø. 1.33
	I SOMERS OF BUTENE	z = 150	938. 9
	ISOMERS OF PENTENE	3.263	0.143
	ISOMERS OF PENTAME	1.346	3.0(1)
	TERPENES	9.355	W. 1955
	METHANE	78.182	34.342
	ETHANE	2.323	5.073
	ETHYLENE	1.189	2.631
	PROPANE	1.176	2.43
	PROPYLEME	8.447	360°8′
	ACETYLENE	9.477	3:10.1
	CYCLOPROPANE	3.000 E	$\mathcal{E}_{i}$
	PROPAULENE	8.803	25. € 25. € 25. €
	METHYLAGETYLENE Ommotuvi – 1 - Dentene	3.M.G	24 17 18 18 18 18 18 18 18 18 18 18 18 18 18
	OFFICATE NERVICES NOT	0.100 173	3 C C C C C C C C C C C C C C C C C C C
	BUTENE	1,855	2.1.0
	ISO-BUTANE	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	1.627
	ISOBUTYLENE	0,888	W. W. M
	TRANS-2-BUTENE	9EB. B	8.0.8
	C13-2-3UTENE	¥.028	N. 1915. A
	1,3-BUTADIENE	# 17.1. W	
	ETHYLACETYLENE N DENTAME	7. (III)	10. 10. 10. 10. 10. 10. 10. 10. 10. 10.
	N-FEM-SAME	5, 048 2, 048 2, 048	5.55 7.45 7.55 7.55 7.55 7.55 7.55 7.55
40000	SIMEINTLIFBUIENE	18.11.00 84.00.00	30 M. 20
4000V	C	5.00 to 0.00 t	10.1.0
43226	100 100 100 100 100 100 100 100 100 100	1. 12. 12. 12. 12. 12. 13. 14. 15. 15. 15. 15. 15. 15. 15. 15. 15. 15	3 (1) (1) (2) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4
43227	- 2	1	5. (1) (2) (2) (3) (4)
43228		1954	W.1.8
43229	NTANE	H. WHI	W.B.W
43230	~	0.551	11.1
43231	HEXANE	7.698 2.698	1.562
43232	HEF LANE OF TANE	0.379	≻್ಯಾ. ಶ್ವಾದ್ಧ ಶ್ವಾದ್ಯ
43234	2,3-DIMETHYL-1-BUTENE	. ⊖ . ⊝	. 🥸

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N-DECANE UNDECANE
CVCLOPENTANE ISOPRENE
1-HEXENE CYCLOHFXANF
N-DODECANE
N-TRIDECANE N-TETPANCOAME
N-PENTADECARE
METHYLCYCLOHEX
ME FHYLCYCLOPENTANE CYCLOHEXANONE
OCTENE
C-2-OCTENE
1-DECENE
1-UNDECENE
3-METHVL-T-2-PI
Z,4-DIMETHYLPENTANE MFTHYLCYCLOPENTFAF
CYCLOHEXENE
2,3-DIMETHYLPEN
Z-METHYLHEXANE 2-2-4-TRIMFTHVI
2,4-DIMETHYLHEX
2,5-DIMETHYLHEX 2 3 A-TPIMETUVI
2,3,3-TRIMETHYL
HEXADECANE
HEPTADECANE OCTADECANE
NONADECANE
EICOSANE
HENE LOSANE DOCOSANE
ETHYLCYCLOHEXAL
C6 OLEFINS
CS OLEFINS
Z,Z-DIMETHYLBUFAN
CTCLOFENIENE 4-METHVL-T-2-PENTENE
7-OLEFINS
3-METHYLHEXANE
2,2,3-TRIMETHYLP
4-METHYLHEPIANE 3-MFTHYLHEPTAME
2,2,5-TKIMETHYLPENTANE
E

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

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WEIGHT PERCENT	a feeth or describence of the feether on the feether		100.1	M - 12 C S	1.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00	26. 20. 20. 20. 20. 20. 20. 20. 20. 20. 20	458.8 25.85.8	16 18 18 18 18 18 18 18 18 18 18 18 18 18	چن	છે. છે. છે	200.0	B. B.B.B.	W. BSB	Ø. Ø⊴1	Ø.145	66.18.91	86%.8	11.93B	CCW. V.	10.31.14	B. W. D.	18.18.18. 18.18.18.	. α α α α α α α α α α α α α α α α α α α	10.2.0	1.0.4 0.00 0.00	M Water	#. E. E	W. R.S.	$\kappa$ . $\beta$ 61	0.00.00	156 157 158 158 158 158 158 158 158 158 158 158	0.836	77.7.8 70.1.8	303 B	16.50 M	M. 9.18	H . N . H	871.8	1.231	1.043	Ø.023	8.22.8	10 M 115	6.034 		35.25. M	<u> </u>		8.835 8.835
VEIGHT PERCERT		į	•	1.1.1.6.	7.82.7		113.5	2. E. S.	11.99.10	J. BAB	9.351	16.00 B	<i>5.08%</i>	H.099	9.067	9.119.6	3. 3. 3. 3. 3. 3.	A.101.10		04.5. M	13.00%	•	201.0	0 5 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	7,000 / 7,000 / 7,000 /	2000 D	J. 61.55	0.868	3.827	6.803	M.111	M . M25	3.125 g. ggu	10 M	9.555.8	3.11.05	J. W.W	B. WS.	B43.5	B.469	J.113	9.186	10. F. S. M.	2005	10.00 m	3.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5	M . MUM	3. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	3.352
NAME		101100 14 15117.1	EINTE AFCOROL	ACC BACKET CLOCKS	ISOUTENT ALCOHOL	N-BOLYL ALCOHOL	180-5017L ALCOHOL	Z-BU-VLE-IANOL	C MINION ALCOHOL	Z-METHOXYETHANOL	2-ETHOXYETHANOL	1-T-2-C-4-TM-CYCLOPENTANE	DIACETONE ALCOHOL	ETHVL ETHER	GLYCOL ETHER	LYCOL	KOPYLENE G	ELMYLENE GLYCOL	HEXYLENE GLYCOL	LEIKAHYDKOFOKAN	ACELIC ACID	METRIC ACTIVE	DOODAL ACETATE	N-RHTVI ACETATE	FTHYL ACRYLATE	CELLOSOL VE ACETATE	ISOPROPYL ACETATE	METHYL AMYL ACETATE	ISOBUTYL ACETATE	DIMETHYL FORMAMIDE	<u>~</u> :	ZHETHUXYETHYL ACETAIE	FORMALDENYDE   ACFTAL FHYDE	PROPRIONAL DEHYDE	BUTYRALDEHYDE	C3 ALDEHYDE	CS ALDEHYDE	C8 ALDEHYDE	ACETONE	METHYL ETHYL KETONE	N-BUTYL K	^: ``	7. KE 10	ELHVLENE UXIDE BRODVIENE OXIDE	FROFYLENE OAIDE	ACETONITALE	ACNTLONIINIE ETHVIAMINE	TRIMETHY! AMINE	METHYL CHLORIDE
SAROAD			0 0 0 0 0 0		4000		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	00000000000000000000000000000000000000		3318	3311	3312	3320	3351	3367	ن زير	.) (	<b>7)</b> (7	nο	7	4555	7040	0.400	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	3438	3443	3444	445	3446	345Ø	451	3.457 2.725 3.025	45582 42583	5.04	351Ø	3511	3512	513	3551	י כי		43560	43561	436.M.	78004	787.67	13721	17/54	438Ø1

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

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VEIGHT VEIGHT PERCENT	ପ୍ରସ୍ତୁ ସେ	6.933 8.956 6.936 8.497 6.222 8.497 6.342 8.986 6.343 8.193 6.193	\$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$	සබනය <b>ල ග</b> ල <b>ි</b> පැහැරිමු පුවු	
VE I	# # # # # # # # # # # # # # # # # # #	<i>அத்தைத்தை</i> பித்திக்கியாகி	2.22.23 2.22.23 2.22.23 2.22.23 2.22.23 2.22.23 2.22.23 2.22.23 2.22.23 2.22.23	<b>建筑的设计的现在分词</b> 1997年, <b>是是</b> 2016年,	2
NAME	DICHLOROMETHANE CHLOROFORM CARBON TETRACHLORIDE CARBON TETRABROMIDE TRICHLOROFLOUROMETHANE	ETHYL CHLORIDE  1,1-DICHLOROETHANE  1,1,1-TRICHLOROETHANE ETHYLENE DICHLORIDE PERCHLOROETHYLENE METHYLENE BROMIDE 1,1,2-TRICHLOROETHANE TPICHLOROETHANE	TRIMETHYLFLUOROSILANE TRIMETHYLFLUOROSILANE DICHLORODIFLUOROMETHANE TRICHLOROETHYLENE CHLORODIFLUOROMETHANE CHLOROFELUOROMETHANE CHLOROPENTAFLUOROGETHANE DICHLOROFELUOROGETHANE CHLOROFELUOROGETHANE CHLOROFELUOROGETHANE NAPHTHA ISOMERS OF XYLENE	DIMETHYLETHYLBENZENE ISOMERS OF ETHYLTOLUENE ISOMERS OF BUTYLBENZENE ISOMERS OF BUTYLBENZENE ISOMERS OF DIETHYLBENZENE ISOMERS OF TRIMETHYLBENZENE ISONERS OF PROPYLBENZENE BENZENE TOLUENE ETHYLBENZENE Ø-XYLENE M-XYLENE	1,3,5-TRIMETHYLBENZENE 1,2,4-TRIMETHYLBENZENE 1,2,4-TRIMETHYLBENZENE N-PROPYLBENZENE O-ETHYLTOLUENE M-ETHYLTOLUENE TERT-BUTYLBENZENE SEC-BUTYLBENZENE SEC-BUTYLBENZENE STYRENE 1,2,3-TRIMETHYLBENZENE 1,2,3-TRIMETHYLBENZENE TETRAMETHYLBENZENE TRI/TETRAALKYL BENZENE
SAROAD	43882 43884 43887 43887	4 4 3 8 1 2 2 4 4 4 8 8 1 2 2 4 8 8 8 1 4 4 8 8 8 1 7 4 8 8 8 2 8 2 8 2 8 2 8 2 8 2 8 2 8 2 8	43822 43822 43822 43822 43823 43823 43838 43838 45181	4 4 5 1 8 3 8 5 1 8 4 8 5 1 8 4 8 5 1 8 4 8 5 1 8 4 5 5 1 8 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	4440 4550 4500 4500 4500 4500 4500 4500

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-13 (continued)

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SAROAD	NAME	VEIGHT PERCENT	WEIGHT *
548	BASE	9.38E	M. Main
454Ø2 458Ø1	BENZOIC ACID MONOCHIORORENZENE	4.835 4.255	8.83.1 8.00.21
	1.4-DIOXANE	3.64B	M. BUB
5.20.01	2,3-DIMETHYLBUTANE	5. 666F	W. W.W
	2-ETHYL-1-BUTENE	¥.100	19.16.16
	C-3-HEXENE	9.800	B.B.B.
	Z-fielkyl-Z-fenjene 1itorene	16. 16. 16. 16. 16. 16. 16. 16. 16. 16.	31.105.02 2.2.2.17
	1 - MET - ENE PROPANTENE	E. 10 Z 10	න .නයට න නටට
	METHYLNAPHTHALENE	(4.19)	0 0 A M
	ETHYLNAPHTHALENE	3.800	
5.6012	DIMETHYLNAPHTHALENE	9.69K	B. B.B.
	PROPYLNAPHTHALENE	N.868	B. B. B
	IKIMETHYLNAPHTHALENE	<i>H.</i> ETB	8. B.B.
	ANIHKACENE METHVI ANTHRACENE	м.ююм к ска	18 - 18 - 18 8 - 64 64
	DIMETHYL-2.3.DIHYDRO-1H-INDENE	877.8	\$15.00 \$15.00 \$15.00
		6.673	Ø.164
	CRYOFLOURANE (FREON 114)	9.817	856.8
	B-METHYLSTYRENE	13.600	M. BHB
	<b>1</b> 02	9.5KZ	g. g.g.
	M-CKESOL (3-METHYLBENZENOL) D-CDESOL (4-METHYLDENSENOL)	्र इ.स.च्या इ.स.च्या	BEE SE
	RENZYICHIORIDE	20 20 20 20 20 20 20 20 20 20 20 20 20 2	13.19.28 20.00.20
	A-PINENE	5.55.50 D. 960.0	8 . Birth
	B-PINEME	M. M.	0.691
	D-LIMOMENE	B. SMB	U.BER
	PHTHALIC ANHYDRIDE	7.7.62	W. 936
	MALEIC ANHYDRIDE	5.801	13 W . W
	CARBON SULFIDE	N. 5. 18. 18. 18.	B. 65.11
	CARBONYL SULFIDE	13 . E. E. E. E.	0. A://
	5,0,0-ikime nyenekane 2-2-Elikimethuk nevane	# 200 # 20	21 21 M
	T-2-HEXENE	1000 M	3 - 13 - 13 3 - 13 - 13 3 - 13 - 13
	C-2-HEXENE	15.101 B	18.18.2
	ISOBUTYRALDEHYDE	13. USB	Ø. H.I.S
	1-METHYLCYCLOHEXENE	9.662	Ø.00.14
	C9 OLETINS	9,882	0.00€
	CIM ALKENES	8.808 8.808	35T-8
	Z-ME-MYC-T-FENIENE Z-MEDTENE	73.20.20.20.20.20.20.20.20.20.20.20.20.20.	M.3032 M. 4003
	4-NONENE	.J • 25.28 S. 6066	8 . B. S.
	ISOPROPYLBENZENE (CUMENE)		<b>1</b>
	INDAN	¥.541	Ø. 15€
	M-DIETHYLBENZENE	3.551	Ø.033
55346 52241	NAPHTHALENE	й.И13 	18.83. 20.83.
5,80,4 / Fixe 4.9	I DOBOL Y L BENZENE TROOME	N. 2012	25 10 10 10 10 10 10 10 10 10 10 10 10 10
55245 55240	INDENE CG AROMATICS	್.ಜನ್ನ ಸ. ಚನ್ನ	10 - 10 OK 10 - 10 OK
1 2 2 3		N. 11.12	

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

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WEIGHT PERCENT	 8.833	33 <b>6.</b> 8	Ø.0010	18.18.18	Ø. Ø.J1	0.900	N. NSN	Ø.941	9.932	0.270	Ø.125	50.93 50.03	% . L	8 . 10 15 15 15 15 15 15 15 15 15 15 15 15 15	6. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	9.317	W. Ø17	Ø.17%	688.6	10.0054	9.018	8.8.3	13.13.13 13.13.13	8.4.50 2.1.50	6.448	6.637	18.037	9.683	0.071	8.032	8.042	4. N. A.	20 00 00 00 00 00 00 00 00 00 00 00 00 0	8.50.50 8.61.7	2	25.00	2. ESE	30 M	8.005	10 H 10 H	9.922	97.8.0	g. #53	8.332	7. M. W.	8.8.3 8.88	
VEIGHT PERCENT	9.501	•	6.668	8.883	188.1	٠	3.688	વ્ર	3.914	Ø. 12.1	Ø . Ør. 6	5.50 to 5.50 t	2. 300 2. 52.00	1. 68.1	2 5 5 5 1 7 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	9.888	9.038	¥.876	B. 848	13.1024	g. 6508	Ø . ØÆ5	5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5	g. 172	# #146	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	2.837	19.037	5.#32	59,841	M.#19	च च इ इ इ इ इ	1.26.	19. UMB	2 803	20 P	4.550	2 2 3	2	1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	8.810	8.883	4.924	5.001	T T T T T T T T T T T T T T T T T T T	4.83.5 4.83.44	
NAME	C1Ø AROMATICS	2-CHLOROTOLUENE	T-BUTYLBENZENE	1,1-DICHLOROETHYLEWE	2,4,4-TRIMETHVL-I-PENTENE	2,4,4-TRIMETHYL-2-PENTENE	ISOVALERALDEHYDE	ETHYLCYCLOPENTANE	IKIMETHYLCYCLOPENTANE	U. MELHYLCYCLOHEXANE	TRUST COST CHENCANE	E LAYLOYOLOMEXAME	DIETHYLCYCHOREANE Nipentol Cyclosexans	CO CVCI CHEXANE	C3 CYCLOHEXANE	C4 CYCLOHEXANE	CS CYCLOHEXANE	C3 ALKYL CYCLOHEXANE	C4 ALKYL CYCLOHEXANE	C4 SUBSTITUTED CYCLOHEXANE	CS SUBSTITUTED CYCLOHEXANE	CA SUBSTITUTED CYCLOHEXANE	C4 SUBSTITUTED CYCLOREXANONE	BULTE CELLUSULVE CS FOTER	2-METHYL-3-HEXANONE	HEPTANONE	ALKENE KETONE	TERPINENE	BUTANDIOL	ISOPROPANOL	E - HYLHEY - ENE	-KIME-HYLDECENE Co al Zzi insan	ALVI CHOSTIFITED CACLOHERAND	C2 ALKVI DECALIN	CARVONFNTHOL	CARVONE	ISOPULEGONE	METHYLHEPTENE	DIMETHYLHEPTANE	1,2-DIBROMOETHANE	1-CHLOROBUTANE	3-(CHLOROMETHYL)-HEPTANE	ETHYL ISOPROPYL ETHER	DIBUTYL ETHER	Z-BOLYELETKAHYDKUFUKAN DBODYLEVELOHEVANONE	2-(2-BJTOXYETHOXY)-ETHANOL	
SAROAD																																														5885 58899	

\* WEIGHT PERCENTAGES EXCLUDING LANDFILL EMISSIONS

TABLE 6-13 (concluded)

WEIGHT *	0.015	Ø.Ø18	6.648	8. 8. 1A	0.001	Ø. Ø∷A	6.273	Ø.239	
PERCENT	LIII E	30.0.3	6. ASA	11.8166	0.091	14, 1532	0.122	ğ.129	
NAME	1-ETHOXY-2-PROPÂNOL	2-ETHYL-1-HEXANOL	1-HEPTANOL	2-METHYL-2,4-PENTANEDIOL	METHYL ISOBUTYRATE	C8 ESTER	SUBSTITUTED C7 ESTER	SUBSTITUTED C9 ESTER	
SAROAD	5.318.8	5 <u>≓</u> 1Ø1	59102	5£1£3	5.91.04	5.91.95	5 <i>i</i> 01 <i>j</i> 06	511187	

TABLE 6-14. Approximate relationship between new and existing speciation profiles.

	Existing Profile		New Profile
No.	Description	No.	Description
7	IC engineturbine, natural gas	719	IC enginereciprocating, natural gas
10	Industrial IC engine reciprocating, natural gas	719	IC enginereciprocating, natural gas
26	Asphalt concrete in-place road asphalt	715 716	Slow cure asphalt Medium cure asphalt
96	Surface coating solvent general	783	Industrial surface coating solvent-based paint
98	Gasoline evaporation	709	Liquid gasolinecomposite of
	working and breathing tank losses	710	product, summer blend Gasoline vaporscomposite of product, summer blend
125	Polymeric surface coating hot air dried	783	Industrial surface coating solvent-based paint
134	Surface coating evaporation general primer	713	Industrial surface coating composite primer
136	Surface coating evaporation metal primer	712	Industrial surface coating
	metal primer	713	composite enamel Industrial surface coating
		714	composite primer Industrial surface coating
		718	composite adhesive Architectural surface coating
		783	<pre>composite solvent Industrial surface coating solvent-based paint</pre>
141	Surface coating evaporation	714	Industrial surface coating composite adhesive

TABLE 6-14 (concluded)

	Existing Profile		New Profile
No.	Description	No.	Description
142	Surface coating evaporation metal furniture adhesive	714	Industrial surface coating composite adnesive
148	Surface coating lacquer metal furniture	711	Industrial surface coating composite lacquer
149	Surface coating lacquer paperboard products and containers	711	Industrial surface coating composite lacquer
156	Surface coating evaporation general composite enamel	712	<pre>Industrial surface coating composite enamel</pre>
		717	Architectural surface coatingwater-based paint
		783	Industrial surface coating solvent-based paint
196	Architectural surface coatingcomposite solvent	717	Architectural surface coatingwater-based paint
271.	Trichloroethylene cleaning solvent	755	Trichlorotrifluoroethane
296	Composite crude oil evaporation-production, fixed roof	758	Oil and gas production fugitivesvalves, unspecified service
514	EvaporationChevron weed oil	760	Evaporative emissions distillate fuel
520	Composite natural gas	758	Oil and gas production fugitivesvalves, unspecified service

because profiles are assigned to SCCs, and there is no direct correspondence between an original profile and a new profile assigned in this study. For example, in some cases several new profiles are used in place of one profile in the original inventory. Also, because several SCCs that were assigned profile No. 600 in the original inventory were reassigned several new profiles in the revised inventory, such profiles are not shown in the table. Furthermore, there are other new profiles used in the revised inventory that are not shown in Table 6-14. This table contains only those profiles that were updated to the new 700 series profiles. New SCCs were assigned new profiles not listed in the table because new SCCs would never have been assigned an old profile.

### INVENTORY UNCERTAINTIES

High-resolution inventories are estimates of emission rates for sources located in a defined region. The ability of such inventories to represent actual emission rates is limited by the data and procedures that are employed in the emission estimation process. Analysis of uncertainties in the emission estimates resulting from available data and procedural limitations is an important aspect of the inventory development process. Furthermore, a knowledge of inventory uncertainties is valuable in interpreting emission estimates and in the analysis of results based on these estimates.

This section discusses the findings of our investigation of uncertainties in the 1979 emission inventory, which were found to be an important aspect of the study. The primary reasons for the occurrence of inventory uncertainties were

Systematic uncertainties in emission data and speciation

Random errors

Double counting of facilities

Missing source categories

Operating deviations including equipment upsets, control technology deterioration, operational perturbations, and variances from permit requirements.

## Uncertainties for Surveyed Facilities

Uncertainty codes were developed for each modification to the emission estimates developed during the facility survey performed in Task 2. The codes represent an estimation of the uncertainty associated with several components of the emission estimate. The codes and their descriptions are provided in Table 6-15 and were noted in the documentation developed for each surveyed facility. These codes were developed for six different areas of the inventory:

Activity data (e.g., throughput, fuel use, etc.)
Emission factors
Control efficiency
Speciation
Temporal distribution
General uncertainties

## Systematic Uncertainties

The survey indicated trends in emission estimates and other information that suggest the presence of systematic uncertainties in the inventory.

### Emission Data

Emission totals in the SCAQMD EIS file were frequently derived from emission fee data. However, we found total emissions for refineries in the 1979 EIS file to be inaccurate because the results of a 1975 emission survey were commonly used to develop the 1979 EIS file for refineries. The following list shows that our revised refinery estimates using survey results and other information for emission factors and 1979 activity data resulted in a 16 percent decrease in TOG emissions from surveyed refineries compared with these emissions in the original inventory.

## Activity Data

- 1-- Usage of surface coatings or solvents are estimates and are not based on actual purchase records.
- 2-- Emissions were based on the usage of surface coatings and solvents. However, it is estimated that over 20 percent of these materials are recycled or disposed of as liquids and therefore not emitted by this facility. This approach of estimating emissions on the basis of usage data without consideration of disposal or recycling has been widely used in the past. This approach will result in an overestimation of emissions for some facilities.
- 3-- Same as code 2 except it is estimated that between 0 and 20 percent of the surface coatings and solvents are recycled or disposed of as liquids.
- 4-- A potential for double counting emissions from solvents exists because surface coatings are thinned using solvents. Therefore, emissions from these solvents may be counted both as solvents and as the organic content of the coatings.
- 5-- Emission estimates from tanks and/or loading were based on 1981 emission fee data. It is not known how representative 1981 throughput data are for 1979.
- 6-- Emission estimates from tanks and/or loading were based on 1981 emission fee data. The facility questionnaire response indicated that 1981 throughput data are representative for 1979.
- 7-- Emission estimates were based on 1979 fuel use data that should be accurate.
- 8-- Emissions were based on 1979 throughput data that should be accurate.
- 9-- Emissions were based on 1981 data. However, these data should be within 20 percent of the actual 1979 data.

continued

## Emission Factors

- 11-- Emission estimates are from 1979 emission fee data. No documentation exists on how these emission estimates were derived.
- 12-- Emission estimates are from 1979 emission fee data. Emission estimates are assumed to have been made using AP-42 storage tank equations.
- 13-- Emission factors are general to an entire class of surface coatings and solvents. The applicability of these factors in this case is somewhat in question.
- 14-- Emission factors are somewhat in question because the surface coating or solvent was classified on the basis of engineering judgment.
- 15-- Emission estimates are from the SCAQMD EDP file. No documentation exists on how these emission estimates were derived.
- 16-- Emissions were estimated by adjusting 1981 emission estimates on the basis of total refinery feed rates for 1979 and 1981.
- 17-- Emission factors for chemical processes were developed by SCAQMD prior to 1979. Documentation for these emission factors was not available.
- 18-- Emission factors for chemical processes were developed on the basis of limited source test data.

# Control Efficiency

- 21-- Emission estimates are from the SCAQMD EDP file. Emission controls exist, but it is not known whether they were accounted for in the emission estimate.
- 22-- Emission estimates for tanks and/or loading from the 1981 emission fee data were used. It is not known if any additional air pollution controls were installed between 1979 and 1981.
- 23-- Control efficiency of the incinerator is in question.

continued

24-- The amount of flashoff and overspray that might not be controlled by the incinerator is unknown.

### Speciation

- 31-- Speciation is an estimate based on engineering judgment. The weight percent of each species should be plus or minus 10 percentage points.
- 32-- Speciation is an estimate based on engineering judgment. The weight percent of each species should be plus or minus 25 percentage points.
- 33-- Speciation is uncertain and based on limited information. However, this speciation is believed to be the most accurate estimate currently available.
- 34-- Speciation is somewhat in question because the surface coating or solvent was classified on the basis of engineering judgment.
- 35-- Speciation should be altered by incineration.

### Temporal Distribution

 $\overline{\phantom{a}}$ 

- 41-- Temporal distribution was assumed on the basis of the industry type and is considered to be certain.
- 42-- Temporal distribution was assumed on the basis of the industry type and is moderately certain.
- 43-- Temporal distribution was assumed on the basis of the industry type and is somewhat uncertain.
- 44-- Temporal distribution was based on information provided by the facility and is considered to be certain.

### General Uncertainties

- 51-- Emissions from the use of vacuum trucks to transfer organic materials exist at this facility but could not be quantified.
- 52-- Emissions from storage tank cleaning exist at this facility but could not be quantified.

continued

# TABLE 6-15 (concluded)

- 53-- Emissions from operations under variance at this facility are significant and are quantified in the 1981 emission fee data.
- 54-- Emissions from component fugitives (e.g., valves and flanges) were assumed to be negligible because of the low volatility of the material being handled.

	TOG Em	issions
Facility ID Code	(kg	/day)
in Los Angeles County	Original	Revised
6	381.6	387.5
10	3711.0	2344.2
23	4540.8	3381.3
2240	549.2	446.0
4208	1445.6	747.8
4210	412.4	414.4
4215	7626.8	7335.2
4217	3063.7	3707.0
5085	1133.3	676.8
5091	610.4	264.9
Total	23,474.8	19,705.1
(tons/day)	(25.9)	(21.7)

In addition, emission estimates in the SCAQMD EDP file were generally made at the time a permit was issued. Because permits have been issued over a number of years, emissions in the EDP file do not reflect 1979 conditions. For example, the following list shows the difference between the original emissions from the EDP file and the revised emissions calculated from activity data reported in the survey for surface coating and solvent usage facilities. The results show the revised TOG emissions for these facilities to be almost four times higher than the corresponding values in the original inventory.

	TOG Emissions	
	(kg/day)	
Facility ID Code	Original	Revised
8241009.19	6.4	44.3
28249008.19	0.8	2.2
28251021.19	40.8	808.6
32227083.19	8.0	23.2
32228044.19	6.4	74.7
32233034.19	1.2	10.0
35216012.19	1.6	21.6
35225017.19	36.0	27.2
. 37230035.19	31.5	16.3
37242060.19	4.2	21.6
40227027.19	4.0	12.8
40228032.19	21.6	25.6

40230032.19	24.8	53.8
40231113.19	1.8	0.6
40234043.19	3.1	23.2
41221022.19	. 35.2	78.4
41230107.19	0.0	0.2
42224021.19	69.6	88.0
44223072.19	42.4	16.0
46231057.19	12.6	263.3
49241035.19	76.0	61.1
51234079.19	1.6	0.3
52222048.19	0.6	1.6
54232026.19	61.6	335.2
59230005.19	17.8	192.8
65235022.19	3.2	3.8
56218011.30	4.0	154.4
58208027.30	6.4	26.4
59225014.30	38.4	39.2
61219014.30	1.6	8.8
62209049.30	31.2	49.6
63221021.30	50.4	47.2
64210006.30	0.0	7.2
77223008.33	35.0	45.6
112213028.33	2.4	1.6
72235024.36	3.2	20.0
85237021.36	4.0	0.4
Total	689.4	2606.8
	(0.8)	(2.9)
(tons/day)	(0.0)	(4.3)

These results show an average 300 percent increase in TOG emissions and indicate that significant systematic trends that either over- or underpredict emissions are present in the inventory.

### Speciation

Revisions to speciation profiles assigned to several categories of sources also indicated significant systematic uncertainties in the inventory. For example, there was a large increase in aromatic emissions from surface coating facilities and a large decrease in such emissions from petroleum marketing (gasoline) operations.

Furthermore, SCCs were sometimes incorrect and nonspecific for certain facilities. For instance, the allocation of emissions in the EIS file to specific SCCs was frequently inconsistent with the emission fee data.

Such inconsistency usually leads to the assignment of inappropriate speciation profiles for such SCCs.

### Random Error

Errors of a random nature were also found in the inventory. The error of greatest significance affected the TOG estimate for the Yorba Linda oil field. As discussed in Section 4, the original TOG estimate of 27.1 tons/day for this oil field was revised to approximately one-tenth of its original value. It appeared that an error in an emission factor might have resulted from a transcription error.

### Double Counting of Facilities

Eight cases of double counting of emissions at six facilities were identified during the study. In each case, an entry from the EIS file representing the total emissions from the facility, and one or more entries from the EDP file also representing a portion of the emissions from that facility, were included in the original inventory. As a result, the entries in the EDP file were considered to represent double counting of inventory emissions.

The fact that eight cases of double counting were discovered during a survey of approximately 160 facilities is significant. We expect there are additional cases of double counting because there are several thousand facilities in the 1979 inventory. We attempted to conclusively verify that these eight entries actually represented cases of double counting, primarily through a review of permit files and emission fee data. In some cases, individuals at these facilities were also contacted.

The general approach used to identify cases of double counting was straightforward and could be used for further investigations. The first step was to compare company names and addresses for facilities in the EDP file with a complete list of all facilities in the EIS file. When a potential case of double counting was discovered, emission fee data were obtained for the facility in the EIS file and permit file information was obtained for the facility in the EDP file. By comparing these sets of information, we identified and later verified those facility emissions that were entered into the inventory more than once.

The following double-counted facilities were deleted from the revised inventory. In each instance, entries under another facility identification code, representing the total emissions for the facility, were retained in the inventory.

Facility Name	Facility ID Code	Original TOG Emissions (kg/day)
Union Carbide	33220011.19	256.8
Chevron Chemical	38214031.19	540.0
Continental Oil/Dougla GATX/Pacific Oasis	=	45.6
Good Tables	39218017.19	8.8
Continental Oil/Dougla GATX/Pacific Oasis		494.4
Continental Oil/Dougla GATX/Pacific Oasis	s/ 44221085.19	42.4
Bentley Laboratories	62209006.30	364.6
Union Oil Company Colton Terminal	89235800.36	204.0
Total (tons/day)		1956.6 (2.2)

# Missing Source Categories

Several source categories are not included in the inventory. Some of the more traditional of these categories are

Storage tank cleaning
Exempt pumps and compressors
Petroleum vacuum trucks
Industrial maintenance coatings
Certain stationary internal combustion engines
Junkyards
Marine fuel transfer
Aircraft refueling

In addition, it was suspected that fugitive losses and industrial solvent usage at some facilities were unreported.

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It was suspected that five source categories were missing from the inventory for several of the facilities surveyed in Task 2. Therefore, five questionnaires were developed to specifically address emissions from fugitive losses (valves, flanges, etc.), solvent usage, stationary internal combustion engines, vacuum trucks, and storage tank cleaning. We frequently found that component fugitive losses were not specifically identified for certain types of sources such as chemical manufacturing facilities and bulk plants. However, because of uncertainties in the composition of total TOG emissions for these sources, it was usually necessary to make a judgment regarding whether isolated changes, such as adding component fugitive emissions, were appropriate for these facilities. In the case of solvent usage, during the survey we found that facilities not normally considered as surface coating operations used solvents that were not included in the inventory. However, the addition of emissions from minor solvent usage at specific facilities might result in double counting if emissions from solvent usage were already accounted for by area source categories. We also found that a substantial number of stationary internal combustion engines were used for standby purposes, but emissions for these engines were not added to the inventory.

If identified during the survey, and where appropriate, emissions from these five source categories were added to the inventory, except in the case of vacuum trucks and storage tank cleaning. Although approximate TOG emission estimates can be made for these two source categories, such estimates were considered too uncertain to be added to the inventory. Nevertheless, the information obtained for these source categories is instructive and useful for future evaluations.

Questionnaires concerning vacuum trucks were sent to most refineries, chemical plants, and bulk terminals participating in the survey. The majority of the surveyed refineries used vacuum trucks for organic materials. On the other hand, a much smaller percentage (less than 25 percent) of the chemical plants and bulk terminals indicated that they used vacuum trucks. Moreover, there were no emission estimates given for vacuum truck usage in the survey responses. Estimates of the amount of organic materials transported annually by vacuum trucks varied widely among facilities. These estimates tended to range between 1000 and 10,000 barrels of organic materials per year. These organic materials were primarily tank bottoms and heavy oils with low vapor pressures.

Questionnaires on storage tank cleaning were also sent to the same types of facilities. The results were similar, with the majority of tank cleaning occurring at refineries. The surveyed refineries each cleaned 2 to 20 tanks in 1979 and emission estimates for some of these operations were developed. Two of these cases are presented as examples:

- (1) One refinery cleaned 7 tanks, resulting in estimated TOG emissions of 4.5 tons/yr.
- (2) Another refinery cleaned 4 tanks, resulting in estimated TOG emissions of 1.0 ton/yr.

These types of emission estimates were considered uncertain and were therefore not added to the revised inventory.

### Missing Facilities

We did not discover individual facilities to be missing from the inventory. However, considering the difficulties associated with identifying such facilities, this does not necessarily imply that all facilities operating in the Basin in 1979 are in the inventory.

## Operating Deviations

Excess emissions from equipment operating under variances are not currently included in the emission inventory, but emissions from such source operations can be significant. Starting in 1981, the SCAQMD emission fee forms requested estimates of emissions resulting from these off-design conditions. Examples of some of the more significant operating variances that we reviewed from these forms are

Flaring of excess process gas
Fluid catalytic cracking unit upset
Organic liquid spills
Equipment start-up
Repairs to seals on storage tanks

Most of the individual incidents of equipment perturbation that we reviewed resulted in less than one ton of emissions, but some of the incidents resulted in emissions in the range of three to five tons. These incidents covered operational deviations for which variances were obtained. However, we expect that there are other incidents, such as equipment malfunctions, control system failures, and so forth, for which variances are not obtained because of their temporary duration.

### Inventory Uncertainty Review by Others

Other investigators have also examined emission inventory uncertainties. In discussing the final 1979 inventory for the SOCAB, SCAQMD enumerated several specific sources of uncertainty in the data base (SCAQMD, 1982a). The following uncertainties were among those cited as leading to both imprecision and bias in the inventory:

Omitted source categories

Speciation profiles based on limited data

Bias toward the correction of erroneously high emission rates

The large number of variable components that affect motor vehicle emissions

The assumption that sources are operating in compliance with regulations

SCAQMD also made estimates in this report of the level of uncertainty in the 1979 inventory. Uncertainties in the daily emissions of ROG and  $\mathrm{NO}_{\mathrm{X}}$  for most individual source categories were estimated to be in the range of 20 to 40 percent. The overall uncertainty in the total inventory of ROG and  $\mathrm{NO}_{\mathrm{X}}$  emissions was then calculated as 10 to 11 percent. Also it was noted in the SCAQMD report (SCAQMD, 1982a) that the following activities to improve emission data for sources in the Basin were currently underway:

Continued efforts to improve emission factors and TOG speciation

SCAQMD development of the Automated Equipment Information System (AEIS) to upgrade data for permitted sources

Several specific research projects sponsored by the ARB and EPA

### Conclusion

The collective experience of the project team with TOG, ROG, and  $\mathrm{NO}_\mathrm{X}$  inventories for the state, and in particular the SOCAB, suggests that the uncertainty in the overall 1979 SOCAB inventory is in the range of 20 to 30 percent. This estimate is based on judgment rather than on mathematical calculations and is generally consistent with similar findings presented by several other inventory specialists. Because judgment is necessary to develop both estimated and calculated uncertainty estimates, various estimates of inventory uncertainty reflect alternative perspectives of the inventory process.

We conclude that progress has been made in the review and reduction of inventory uncertainties and recommend that further work focus on reducing such uncertainties.

### 7 CONCLUSIONS AND RECOMMENDATIONS

This study involved a major effort to improve the ROG and  ${\rm NO_X}$  inventory for the SOCAB. Study conclusions and recommendations for activities to further improve the inventory are discussed in this section.

### CONCLUSIONS

## Emission Data and Classes of Organic Gases

The major effects of study changes in emissions and classes of organic gases in the revised inventory are as follows:

- (1) Overall TOG and ROG emissions increased by 6.2 and 7.5 tons/day (0.2 and 1.0 percent), respectively, and  $\mathrm{NO}_{\mathrm{X}}$  emissions decreased by 15.6 tons/day (-3.0 percent). These percentage changes are within the overall estimated variation for typical inventory emission totals. Thus, on the basis of this study, the 1979 total inventory of TOG and  $\mathrm{NO}_{\mathrm{X}}$  emissions was found to be a reasonably accurate representation of the actual basin-wide inventory. In addition, significant changes to both individual source and source category emissions and to ROG speciation were made during the study.
- (2) Emissions of TOG and ROG for petroleum refining and marketing activities decreased. Although petroleum production TOG emissions increased, ROG emissions decreased as a result of a change in speciation. Emissions of TOG and ROG from certain categories of surface coating and solvent use increased.
- (3) Petroleum refining  $NO_X$  emissions increased, whereas  $NO_X$  emissions from unspecified sources of fuel combustion decreased.
- (4) Classes of organic gases: emissions of olefins and aromatics increased, and emissions of paraffins, carbonyls, and ethylene decreased. It is significant that the aromatic component of the inventory increased by 11.8 tons/day primarily as a result of the improved speciation profiles for surface coating categories;

however, substantial decreases in the aromatic portion of petroleum refining and marketing emissions were primarily the result of revised speciation profiles for gasoline. Thus, important inventory changes in organic gas reactivity have occurred as a result of this study.

### Speciation

The profiles used to speciate TOG emissions are the most uncertain component of a high-resolution inventory. Because of the need for accurate speciation data and the expectation that this need will increase in the future, the speciation profiles that currently exist fall considerably short of meeting this need. The primary source of speciation data is the EPA VOC species data manual. This document contains many species profiles for a variety of emission source categories, but is lacking in several respects.

- (1) Many important source categories are not included in the manual.
- (2) Some of the profiles are too limited to be useful. For example, profiles were developed for individual surface coating samples rather than for representative composite samples of a coating type (e.g., enamel, primer, lacquer, etc.).
- (3) Some of the profiles are outdated. For instance, the composition of solvents used in surface coatings has changed significantly as a result of regulations regarding the organic content of coatings.
- (4) The majority of the profiles are based on engineering evaluations and literature reviews employing limited data, rather than on direct sampling and analysis. As a result, many of the profiles are considered to be uncertain.

This study emphasized improvement in the state of knowledge for reactive species comprising TOG emissions from gasoline evaporative losses, coatings, and solvents. As a result, the speciation of both the overall inventory and several individual categories has been modified, particularly for aromatic organic gases. Furthermore, our new speciation profiles represent the first major update of initial efforts in profile development for non-motor-vehicle sources performed by KVB in 1978. These new profiles are also expected to be used in the development of high-resolution inventories throughout the country. Thus, the work reported here has taken on nationwide significance.

## Other Inventory Revisions

Important changes not apparent through a review of emission totals were also made to the inventory:

- (1) In particular, revisions were incorporated into the MED file for the diurnal distribution of emissions from power plants, which are the largest stationary source category of  $NO_X$  emissions in the inventory.
- (2) In addition, there were changes to the spatial resolution of emissions, particularly as a result of the redistribution of TOG emissions among sources of oil and gas production.

### Uncertainties

This research effort has improved the understanding of the primary types of uncertainties in high-resolution inventories and has reduced their extent in the 1979 SOCAB inventory. For example, we found that

- (1) Eight of the approximately 160 facilities we surveyed had been counted twice.
- (2) Eight categories of sources, such as storage tank cleaning, were not accounted for by the 1979 inventory.
- (3) A random error of about 27 tons/day of TOG emissions was found in the inventory.
- (4) Systematic trends that either over- or underpredict emissions-such as a 2 tons/day increase in TOG emissions from surface coating and solvent usage facilities in the survey--are present in the inventory.

### RECOMMENDATIONS

Several recommendations are provided here to guide future activities in the area of emission inventory development.

## Organic Gas Speciation

- (1) Additional development of TOG speciation profiles for other source categories not examined in this study will lead to further significant improvement in ROG inventories for California. Therefore, the study team recommends that future research programs continue to emphasize the improvement of organic gas speciation profiles. This recommendation is consistent with the major objective of this study—to develop improved modeling—quality inventories of ROG emissions by focusing on categories that have the greatest potential for causing uncertainty in predicting downwind ozone concentrations.
- (2) Much work needs to be done in examining and improving TOG species profiles primarily because there has been relatively little effort in this area during the last decade. It is a demanding task to set priorities regarding the source types most needing investigation because so many categories need in-depth evaluation. Furthermore, the level of effort that should be expended is considerable. Nevertheless, the study team, in conjunction with ARB staff, is in a position to assist in establishing these future priorities because of its collective experience in reactive modeling and in generating speciation profiles under various studies, and also because the first task in this project yielded extensive information for ranking source categories for this and future studies.
- (3) We recommend that improved profiles be developed from original testing programs and innovative methods such as those used in this study. Many categories need to be examined in depth. For the source types tested in this study, we can make the following statements:
  - (a) Surface coatings--The changes made to profiles for different coatings have been significant. Further work should be performed to insure that these results are representative and to include additional kinds of coatings.
  - (b) Petroleum products--Although the speciation of gasoline has been greatly improved, the speciation of other petroleum products remains questionable.
  - (c) Asphalt--Because of difficulties encountered in analyzing asphalt samples beyond the control of the study team (see Section 5), additional work is needed to examine different types of asphalt.

- Quality assurance procedures need to be carefully evaluated and applied throughout the planning and analytical phases of speciation test programs.
- (4) We further recommend that the improved speciation profiles developed in this study be used to develop ROG estimates for source categories in other portions of the state. In addition, the improved profiles should be employed for inventories to be used in future photochemical modeling studies. New modeling sensitivity studies should also be performed to focus on the trends established in this project for classes of organic gases.
- (5) Consideration should also be given to the reconstruction of profile No. 600, the all-category composite used in cases for which a suitable profile is unavailable. The revised profile should be developed on the basis of the full set of existing and new profiles used in the revised inventory. In addition, the need for improved speciation of  $\mathrm{NO}_{\mathrm{X}}$  emissions into nitric oxide (NO) and nitrogen dioxide (NO<sub>2</sub>) species should be conclusively determined.

## Inventory Data Bases

- (1) We recommend that the revisions made to the MED inventory in this study also be incorporated into various state and local emission data bases for sources in California.
- (2) Particular attention should be given to improvements in the emission estimates for chemical manufacturing facilities. In general, the existing TOG emission estimates were based on generalized emission factors in conjunction with total plant throughput. More specific estimates should be developed to improve the accuracy and level of detail of these estimates.
- (3) Since the development of the 1979 inventory, more investigation into questionable data obtained from the data base has occurred at ARB and SCAQMD. Continued emphasis on matching emission fee data to emissions in the EIS file is recommended. We also recommend consideration of a single numbering system for all point sources in the SOCAB, which is under analysis at ARB and SCAQMD, in order to reduce the occurrence of double-counted sources.

### Identification of Emissions

- (1) It is important to emphasize the need for an accurate categorization of emissions into SCCs. An accurate categorization results in the assignment of appropriate speciation profiles, which, in turn, leads to improved modeling studies. Since the development of the 1979 inventory, more attention is now being given to this activity by SCAQMD. We recommend continued emphasis on the use of the emission fee data to properly allocate emissions into SCCs because of the importance of SCCs in identifying emissions (including toxic air contaminants) by source type, in evaluating control measures, and in assigning appropriate speciation profiles. In addition, the level of detail provided by various documentation sets used to develop the inventory, such as the emission fee data, needs to be retained to the greatest extent possible in the data bases.
- (2) The creation of accurate speciation profile and SCC descriptors is also important because they form the basis for the assignment of profiles and SCCs. Furthermore, the level of detail that is selected for the data base should be consistent. It is inappropriate, for example, to assign one device per storage tank for one facility and one device for all tanks for another facility. Generally, we recommend a level of detail that will allow for accurate speciation, but will not be overly burdensome. Examples of this level of detail would be assigning one device for each type of product and storage tank (e.g., crude oil-fixed roof), and for each type of surface coating.

## Temporal Distributions

The temporal resolution of small point and area sources should be reviewed because issues relating to several diurnal distributions in the inventory have been noted in the literature (Oliver, Hogo, and Saxena, 1983). Substitute diurnal distributions should be considered to account for the compound nature of aggregated source operations throughout a daily period. Such compound distributions can be based on survey results and other information concerning source operating schedules. For instance, it should be determined if evaporative organic emissions from surface coating and solvent cleaning operations should be distributed over a diurnal period longer than the actual operating hours of the source. Evaporative losses might be

distributed equally over 24 hours, for example, to simulate the process and the rate of evaporation for a surface coating facility that operates 8 to 16 hours per day.

## Future Studies

The results of the ranking of source categories in Task 1 should be used to establish priorities during the conception and performance of other studies aimed at improving the quality of ROG and  ${\rm NO_X}$  emission inventories.

### IMPLICATIONS OF THIS WORK

This study is a part of efforts by the state of California to improve and update its emission inventories. The implications of our work for ARB regulatory programs include the following study findings.

- (1) Improvements to emission inventories for the state remain to be made; significant improvement is possible for high-resolution inventories.
- (2) Poor speciation data are currently relied upon; these data should be improved because of their importance to ozone modeling studies and to the identification and quantification of toxic air contaminants.
- (3) Revisions made in this study to speciation data may be of significance in interpreting the results of previous studies used to identify levels of source control for organic gases.
- (4) Uncertainties existing in current inventories can be reduced through detailed evaluations.

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